

Synthesis of the Antibiotic Anticapsin¹

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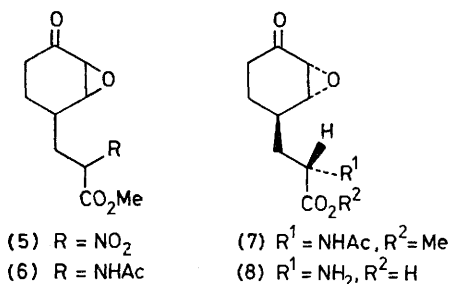
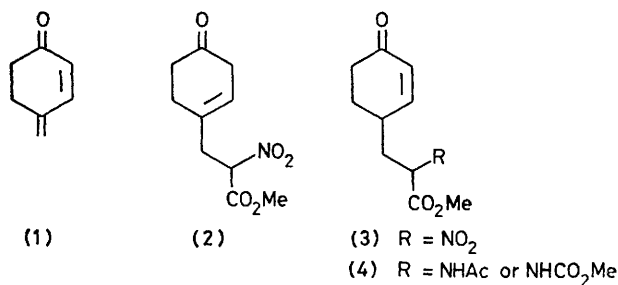
Summary The antibiotic anticapsin (**8**) has been synthesised by a route involving Michael addition of methyl nitro-acetate to 4-methylenecyclohex-2-enone (**1**) and enzymic deacylation of the derived (\pm)-*N*-acetylanticapsin methyl ester (**7**).

ANTICAPSIN, an amino-acid antibiotic from *Streptomyces griseoplanus*² and *Bacillus subtilis*³ which inhibits hyaluronic acid capsule formation in *Streptococcus pyogenes*,² has been assigned^{3,4} the structure and absolute stereochemistry (**8**). It occurs also as the C-terminal amino-acid in bacilysin,³ a substance formed by certain bacteria which causes lysis in growing staphylococci and which has been shown to be identical with tetaine and bacillin.⁵ We report here a synthesis of anticapsin.

Synthetic approaches utilising reduced tyrosine derivatives of type (**4**) were precluded by the ease of intramolecular addition of the amino-function to the conjugated ketone to give 1,2-disubstituted-6-oxo-octahydroindoles.⁶ Accordingly the carbon skeleton of anticapsin was generated by Michael addition of methyl nitro-acetate to the exocyclic double bond of the known⁷ 4-methylenecyclohex-2-enone (**1**), catalysis by *N*-benzyltrimethylammonium hydroxide affording the adduct (**2**) in 95% yield. Equilibration of

the $\beta\gamma$ -enone (**2**), ν_{\max} (film) 1753 (CO₂Me), 1717 (CO), and 1560 cm⁻¹ (NO₂), δ (CDCl₃) 5.64 (1H, m, =CH), 5.32 (1H, dd, *J* 6 and 10 Hz, CHNO₂), and 3.84 (3H, s, CO₂Me), with 1% hydrochloric acid in dimethyl sulphoxide at room temperature gave a 1:1 mixture, separable by silica gel chromatography, of the $\beta\gamma$ - and $\alpha\beta$ -enones. The $\alpha\beta$ -enone (**3**) (a 1:1 mixture of diastereoisomers), ν_{\max} (film) 1750 (CO₂Me), 1675 (conjugated CO), and 1555 cm⁻¹ (NO₂), δ (CDCl₃) 6.78 [1H, m, C(3)-H], 6.00 [1H, dd, *J* 10 and 2 Hz, C(2)-H], 5.32 (1H, m, CHNO₂), and 3.84 (3H, s, CO₂Me), was oxidised with alkaline hydrogen peroxide in methanol at -20 °C to form the $\alpha\beta$ -epoxyketone (**5**) (79%), ν_{\max} (film) 1750 (CO₂Me), 1715 (CO), and 1560 cm⁻¹ (NO₂), δ (CDCl₃) 5.30 (1H, m, CHNO₂), 3.84 (3H, s, CO₂Me), 3.45 [1H, m, C(3)-H], and 3.25 [1H, d, *J* 4 Hz, C(2)-H]. The product (**5**) was a mixture of stereoisomers, but was too unstable to allow separation at this stage.

Catalytic reduction of the nitro-group in (**5**) with Raney nickel in acetic anhydride gave the *N*-acetylamino-epoxyketone (**6**) (35%), readily separated by silica gel chromatography into *cis* and *trans* components (as regards the epoxy and alanyl substituents) in 4:1 ratio. The two *cis* diastereoisomers (1:1 ratio), δ (CDCl₃) 3.55 [dd, *J* 4 and 2 Hz, C(3)-H] and 3.23 [d, *J* 4 Hz, C(2)-H], and δ (CDCl₃)



3.41 [dd, J 4 and 2 Hz, C(3)-H] and 3.21 [d, J 4 Hz, C(2)-H], were readily distinguished from the two more polar *trans* diastereoisomers (1:1 ratio), δ (CDCl₃) 3.74 [d, partly obscured by CO₂Me, C(3)-H] and 3.25 [d, J 4 Hz, C(2)-H], and δ (CDCl₃) 3.41 [d, J 4 Hz, C(3)-H] and 3.22 [d, J 4 Hz, C(2)-H], by the vicinal coupling constant of 2 Hz between C(3)-H and C(4)-H in the *cis* compounds, in contrast to the virtual absence of any such coupling in the *trans* compounds.^{4,8}

Careful silica gel chromatography separated the two *trans* diastereoisomers of the *N*-acetylamino-epoxy-ketone (6).

† Calbiochem salt-free lyophilised B grade.

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³ J. E. Walker and E. P. Abraham, *Biochem. J.*, 1970, **118**, 563.

⁴ N. Neuss, B. B. Molloy, R. Shah, and N. DeLaHiguera, *Biochem. J.*, 1970, **118**, 571.

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⁶ C. N. Chong and R. W. Rickards, unpublished data.

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⁸ L. M. Jackman and S. Sternhell, 'Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry,' Pergamon Press, Oxford, 1969, p. 280.

⁹ J. B. Jones and J. F. Beck, in 'Techniques of Chemistry, Vol. X. Applications of Biochemical Systems in Organic Chemistry,' eds. J. B. Jones, C. J. Sih, and D. Perlman, Wiley, New York, 1976, p. 210; J. P. Greenstein and M. Winitz, 'Chemistry of the Amino Acids,' Wiley, New York, 1961, Vol. 1, p. 740.

The more polar isomer showed identical ¹H n.m.r. characteristics, δ (CDCl₃) 6.04 (1H, brs, exchangeable NH), 4.77 (1H, m, CHNHAc), 3.77 (3H, s, CO₂Me), 3.41 [1H, d, J 4 Hz, C(3)-H], 3.22 [1H, d, J 4 Hz, C(2)-H], and 2.06 (3H, s, COMe), to *N*-acetylanticapsin methyl ester prepared⁴ from authentic anticapsin, and accordingly has the relative stereochemistry represented in (7), corresponding to that of (-)-anticapsin (8).

Conversion of (\pm)-*N*-acetylanticapsin methyl ester (7) into anticapsin was accomplished by alkaline hydrolysis of the methyl ester (1.25 M aqueous sodium hydroxide, 1 mol. equiv., 5 °C, 15 min), followed by enzymic *N*-deacetylation with hog kidney acylase† (38 °C, pH 7, 16.5 h). A relatively high ratio (2:1) of enzyme:substrate was required. The product, isolated in 9% yield (based on utilisation of one enantiomer) after gel filtration and cellulose chromatography, was identical to natural (-)-anticapsin (8) by amino-acid analysis, t.l.c. in two solvent systems, and ¹H n.m.r. spectroscopy, δ (D₂O, relative to external Me₄Si) 4.36 (t, J 7 Hz, CHNH₂), 4.20 [d, J 4 Hz, C(3)-H], and 3.87 [d, J 4 Hz, C(2)-H]. Comparison of c.d. spectra of this product and authentic anticapsin indicated a content of 87% of the natural (-)-enantiomer (8). In view of the limited quantity of material available for chiroptical measurements, and the virtually exclusive preference of the renal acylases for L- α -amino-acid substrates,⁹ the enzymic deacylation is probably enantiospecific.

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