

Resolution of Presqualene and Prephytoene Alcohols

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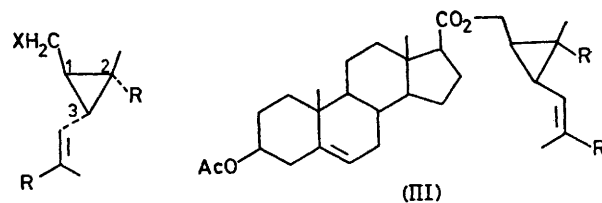
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Summary Synthetic presqualene and prephytoene alcohols have been resolved through their etienic acid derivatives; the biologically active prephytoene alcohol (as its pyrophosphate) has the (1*R*,2*R*,3*R*) absolute configuration.

THE stereospecific head-to-head condensation of two molecules of farnesyl pyrophosphate during the biosynthesis of squalene¹ and geranylgeranyl pyrophosphate in phytoene^{2,3} biosynthesis proceeds through cyclopropyl carbanyl intermediates presqualene pyrophosphate (Ib) and prephytoene pyrophosphate (IIb), respectively. The absolute configuration of natural (+)-presqualene alcohol was shown to be (1*R*,2*R*,3*R*).⁴ Although natural prephytoene alcohol has been suggested¹⁶ to have the same absolute configuration as presqualene alcohol, this has not received experimental support. In 1972, Porter and his co-workers reported the isolation of 'prelycopersene pyrophosphate' from a tomato enzyme system.³ This intermediate exhibited a positive optical rotatory dispersion curve analogous to presqualene pyrophosphate. However, the results of Rilling^{2b} strongly suggest that Porter's intermediate was a product of squalene synthetase and not of the carotogenic enzyme responsible for the head-to-head dimerization of geranylgeranyl pyrophosphate.



(Ia); R = C₁₁H₁₉, X = OH

(Ib); R = C₁₁H₁₉, X = OPP

(IIa); R = C₁₆H₂₇, X = OH

(IIb); R = C₁₆H₂₇, X = OPP

(III)

a; R = C₁₁H₁₉

b; R = C₁₆H₂₇

We now report the resolution⁵ of synthetic presqualene and prephytoene alcohols through their etienic acid derivatives (IIIa and IIIb) and the demonstration that the biologically active prephytoene alcohol (as its pyrophosphate) exhibits a positive optical rotatory dispersion curve and that its absolute configuration is (1*R*,2*R*,3*R*).

(±)-Presqualene and (±)-prephytoene alcohols were synthesized as previously described.^{1a,2a} Treatment of the alcohols with readily available 3β-acetoxy-17β-chloroformyl androst-5-ene⁶ in dry pyridine gave presqualene

TABLE. Optical rotatory dispersion data^a

Wavelength λ/nm	(+)-Prephytoene c 0.0037 [α]	(-)-Prephytoene c 0.0035 [α]	(+)-Presqualene c 0.0010 [α]	(-)-Presqualene c 0.0011 [α]
365 Hg	149	-143	193	-206
436 Hg	90	-83	103	-110
546 Hg	45	-46	48	-52
578 Hg	39	-40	48	-48
589 Na	38	-37	47	-48

^a Rotations were measured at 20.0 °C in CHCl₃ using a Perkin Elmer Model 241 Polarimeter; deviations < 10% of the reported values were observed.

etienate (IIIa) and prephytoene etienate (IIIb) as a mixture of diastereomers. Presqualene etienate, $[\alpha]_D^{20} - 14.5^\circ$ in CHCl_3 , and prephytoene etienate, $[\alpha]_D^{20} - 6.4^\circ$ in CHCl_3 , gave satisfactory elemental analyses. The diastereomers were separated by high pressure liquid chromatography on μ -Porasil (Waters Assoc. 3.9 mm I.D. \times 30 cm; eluent CH_2Cl_2 ; 1 ml min^{-1}). The pure diastereomers of presqualene and prephytoene yielded identical n.m.r. data except for a characteristic octet of an ABX system for the diastereotopic cyclopropyl carbinyl protons that each showed.

The individual diastereomers of (IIIa) and (IIIb) were reduced with LiAlH_4 to give the enantiomeric alcohols in 85–90% yield. The o.r.d. data for the enantiomeric alcohols are given in the Table.

The (+)-enantiomers of presqualene alcohol and prephytoene alcohol showed a normal positive o.r.d. curve, identical to natural presqualene alcohol⁷ thus establishing the absolute configuration of (+)-prephytoene as (1*R*,2*R*,3*R*).

For biological studies each of the enantiomeric alcohols was tritiated at the cyclopropyl carbinyl position by reduction of the corresponding aldehyde with tritium labelled NaBH_4 . The labelled alcohols were then converted into the corresponding pyrophosphates. Only the pyrophosphates derived from (+)-presqualene alcohol and (+)-prephytoene alcohol proved to be biologically active.[†]

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[†] (+)-Presqualene pyrophosphate was utilised by yeast subcellular particles and (+)-prephytoene was utilised by phycomyces *Blakesleeanus* strains C_9 and C_6 .

¹ (a) L. J. Altman, R. C. Kowerski, and H. C. Rilling, *J. Amer. Chem. Soc.*, 1971, **93**, 1782; (b) H. C. Rilling, C. D. Poulter, W. W. Epstein, and B. Larsen, *ibid.*, 1783; R. M. Coates and W. H. Robinson, *ibid.*, 1785; R. V. M. Campbell, L. Crombie, and G. Pattenden, *Chem. Comm.*, 1971, 216; (c) H. Wasner and F. Lynen, *FEBS Letters*, 1970, **12**, 54; (d) J. Edmond, G. Popjak, S. Wong, and V. P. Williams, *J. Biol. Chem.*, 1971, **246**, 6254; F. Muscio, J. P. Carlson, L. Kuehl, and H. C. Rilling, *ibid.*, 1974, **249**, 3746; R. V. M. Campbell, L. Crombie, D. A. R. Findley, R. W. King, G. Pattenden, and D. A. Whiting, *J.C.S. Perkin I*, 1975, 897.

² (a) L. J. Altman, L. Ash, R. C. Kowerski, W. W. Epstein, B. R. Larson, H. C. Rilling, F. Muscio, and D. E. Gregonis, *J. Amer. Chem. Soc.*, 1972, **94**, 3257; (b) D. E. Gregonis and H. C. Rilling, *Biochemistry*, 1974, **13**, 1538.

³ A. A. Qureshi, F. J. Barnes, and J. W. Porter, *J. Biol. Chem.*, 1972, **247**, 6730; A. A. Qureshi, F. J. Barnes, E. J. Semmler, and J. W. Porter, *ibid.*, 1973, **248**, 2755; F. J. Barnes, A. A. Qureshi, E. J. Semmler, and J. W. Porter, *ibid.*, 1973, **248**, 2768.

⁴ G. Popjak, J. Edmond, and S. Wong, *J. Amer. Chem. Soc.*, 1973, **95**, 2713.

⁵ R. B. Boar and K. Damps, *Tetrahedron Letters*, 1974, **42**, 3731; *J.C.S. Perkin I*, 1977, 709.

⁶ J. Staunton and E. J. Eisenbraun, *Org. Synth.*, 1962, **42**, 4.

⁷ G. Popjak, J. Edmond, K. Clifford, and V. Williams, *J. Biol. Chem.*, 1969, **244**, 1897.