

Intramolecular Cyclisation of Phenolic α -Diazoketones. Novel Synthesis of the Spiro[4.5]decane Carbon Framework

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Summary Copper(I) halide-catalysed decomposition of phenolic α -diazoketones gave spiro[4.5]deca-6,9-diene-2,8-dione in high yield.

INTRAMOLECULAR reactions between diazoalkanes and benzene rings in the presence of metal salt give cycloheptatriene derivatives *via* intramolecular carbene addition.¹ Recently, Ledon and his co-workers² reported that acidic hydrolysis of the cycloheptatriene derivative obtained by the copper-catalysed decomposition of *p*-methoxybenzyl methyl diazomalonate, gave the 2-oxaspiro[4.5]decane derivative. We examined the intramolecular cyclisation of phenolic α -diazoketones in the presence of copper(I) halides and report a novel synthesis of spiro[4.5]decane carbon framework.

A solution of the phenolic α -diazoketone (Ia) in benzene [or tetrahydrofuran (THF)] in the presence of copper(I) chloride [or copper(I) iodide] was heated to reflux for 4.5–5.5 h. After removing the insoluble material, preparative

t.l.c. of the crude product gave spiro[4.5]deca-6,9-diene-2,8-dione (IIa)³ in high yield (Table). Similarly, catalytic decomposition of (Ib), † m.p. 109–110 °C (decomp.), prepared from the corresponding acid chloride and diazoethane, gave 1-methylspiro[4.5]deca-6,9-diene-2,8-dione (IIb), m.p. 93–94 °C, in high yield (Table). ‡

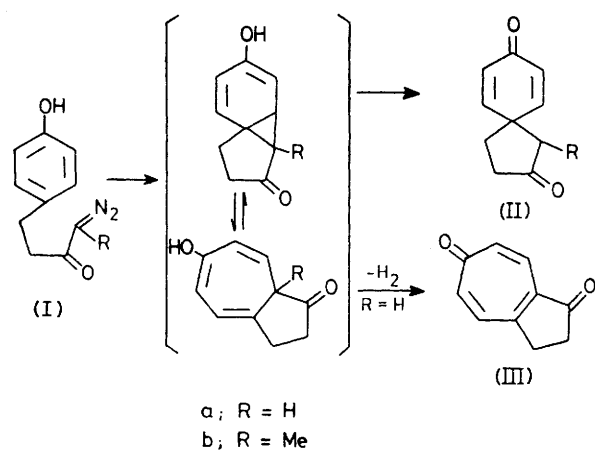
TABLE

Diazo-ketone	Copper halide	Solvent (concentration/M)	Reaction time/h	Spiro[4.5]-decane (yield/%)
(Ia)	CuCl	THF (0.30)	4.5	(IIa) (61)
"	"	THF (0.16)	4.5	" (79)
"	"	C ₆ H ₆ (0.16)	5.5	" (80)
"	CuI	C ₆ H ₆ (0.16)	5.5	" (55)
(Ib)	CuCl	THF (0.16)	2	(IIb) (60)
		C ₆ H ₆ (0.16)	2	" (90)

When (Ia) in benzene in the presence of copper(I) chloride was heated to reflux for only 20 min (the starting material completely disappeared), preparative t.l.c. of the crude pro-

† Satisfactory analytical and spectral data were obtained for new compounds.

‡ In our hands, application of Mander's method gave (IIb) in 5–10% yield.



duct gave (IIa) (34%), 2,3-dihydroazulene-1,6-dione (III) (8%), and several unidentified products. Compound (III) is crystalline, pale brown in colour, m.p. 125–126 °C, and is very unstable. The structure of (III) was determined by the following spectral data: $\nu(\text{CHCl}_3)$ 1718, 1637, 1617, and 1583 (typical of tropone⁴) cm^{-1} ; $\delta(\text{CDCl}_3)$ 7.43 [1H, d, J 12 Hz,⁵ C(8)-H], 7.1–7.2 [2H, m, C(5)-H and C(4)-H], 6.99 [1H, br d, J 12 Hz, C(7)-H], and 3.2–2.8 and 2.8–2.5 [each 2H, m, C(2)-H and C(3)-H]; $\lambda(\text{EtOH})$ 233 (log ϵ 4.35), 240 sh, 309 sh, and 316 (log ϵ 3.89) nm;^{5,6} m/e 160 (M^+), 132 ($M-\text{CO}$), and 104 ($M-2\text{CO}$).

The above results indicate that the catalytic decomposition of phenolic α -diazoketones proceeds *via* norcaradiene derivatives to give the spiro-dienone.

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⁴ Y. Ikegami, *Bull. Chem. Soc. Japan*, 1962, **35**, 967.

⁵ O. L. Chapman and T. H. Koch, *J. Org. Chem.*, 1966, **31**, 1042.

⁶ Y. Hirose, B. Tomita, and T. Nakatsuka, *Tetrahedron Letters*, 1966, 5875; I. Murata, *Bull. Chem. Soc. Japan*, 1961, **34**, 577.