

Structures of Cathedulin Alkaloids from *Catha edulis* (Khat) of Kenyan and Ethiopian Origin

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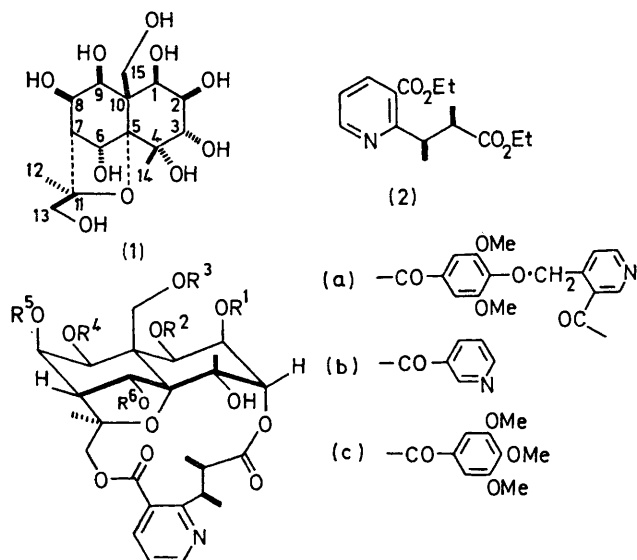
Summary The structures of cathedulins K-1, K-2, K-6, K-12, and K-15, isolated from Kenyan khat, are formulated as (4), (3), (5), (8), and (6) respectively; cathedulin E-3 (Ethiopian khat) is formulated as (7) and E-4, E-5, and E-6 can be treated similarly.

IN earlier reports^{1,2} the isolation and structures of a number of alkaloids from *Catha edulis* (khat) of Ethiopian origin was discussed.† Chromatographic examination of khat of

Kenyan origin has yielded five weakly basic new alkaloids,^{3,4} together with cathedulin K-11 which is identical with cathedulin E-3.¹ The large alkaloid K-12 is connected structurally with E-3, -4, -5, and -6 whilst K-1, -2, -6, and -15 are closely related.

Mass spectral data (electron impact and chemical ionisation) indicate the following formulae: K-1, $C_{42}H_{53}NO_{20}$, m.p. 165—168 °C; K-2, $C_{40}H_{51}NO_{19}$, m.p. 181—184 °C; K-6, $C_{38}H_{49}NO_{18}$, m.p. 176—180 °C; K-15, $C_{36}H_{47}NO_{17}$, m.p. 191—194 °C; K-12, $C_{54}H_{62}N_2O_{23}$, m.p. 268—272 °C.

† These alkaloids were designated (refs. 1 and 2) by the generic term cathedulin followed by a number: this system is expanded by prefixing the number with the letter E to indicate geographic origin of the khat. The new Kenyan series is prefixed by K.



Cathedulin

- K-2 (3) $R^1=H, R^3=Ac; R^2, R^4, R^5, R^6=3 \times Ac$ and $AcO \cdot CMe_2CO-$
- K-1 (4) $R^1=R^3=Ac; R^2, R^4, R^5, R^6=3 \times Ac$ and $AcO \cdot CMe_2CO-$
- K-6 (5) $R^1=R^3=H; R^2, R^4, R^5, R^6=3 \times Ac$ and $AcO \cdot CMe_2CO-$
- K-15 (6) $R^1=R^3=H; R^2, R^4, R^5, R^6=3 \times Ac$ and $HO \cdot CMe_2CO-$
- E-3 (7) $R^1=R^6=Ac; R^2, R^4=Ac$ and $AcO \cdot CMe_2CO-$; $R^5, R^3=(a)$
- K-12 (8) $R^1=R^6=Ac; R^2, R^4=Ac$ and $AcO \cdot CMe_2CO-$; $R^5, R^3=(b); R^5=(c)$
- E-4 (9) $R^1=Ac, R^6=H; R^2, R^4=Ac$ and $AcO \cdot CMe_2CO-$; $R^5, R^3=(a)$
- E-5 (10) $R^1=R^6=Ac; R^2, R^4=PhCO$ and $AcO \cdot CMe_2CO-$; $R^5, R^3=(b), R^5=(c)$
- E-6 (11) $R^1=Ac, R^6=H; R^2, R^4=PhCO$ and $AcO \cdot CMe_2CO-$; $R^5, R^3=(b), R^5=(c)$

Cathedulin K-2 is the most abundant alkaloid in this sample of Kenyan khat and on ethanolysis it yields euonyminol (1) [identified by comparison of (1) and its octaacetate, m.p. 192–195 °C, with authentic specimens] and diethyl evoninate (2). Quantitative ethanolysis indicated five acetate residues (g.l.c.). 1H and ^{13}C n.m.r. data showed signals consistent with an acylated euonyminol core having a free 2-hydroxy-group (2-H, δ 4.10; OH, 2.94br) and a free tertiary hydroxy-group (OH, 4.53 sharp), together with an evoninate diester bridge spanning C-3 to C-13 (shown by comparison of resonances with alkaloids having this feature). Left unassigned are four ^{13}C signals: two OR quartets in the range δ 18–25 p.p.m. (2Me), one OR singlet in the range 69–74 p.p.m. ($\equiv C-O$), one OR singlet 168–175 p.p.m. (C=O), and two 1H resonances (each 3H) in the range δ 1.5–1.7. Only an α -oxyisobutyrate unit fits these data and in K-2 it must take the form of an α -acetoxyisobutyrate.

Careful ethanolysis permitted identification of α -hydroxyisobutyric acid as the free acid (g.l.c., t.l.c.) and as its ethyl ester (g.l.c.) and trimethylsilyl derivative (g.l.c.) using a

† Added in proof: Partial hydrolysis of K-2 (3) has yielded, *inter alia*, two compounds having, as defined by 1H n.m.r. and mass spectra, $R^1, R^2, R^3, R^5=H, R^4=AcO \cdot CMe_2CO-, R^6=Ac$, and $R^1, R^2, R^3, R^4, R^5=H, R^6=Ac$. The α -acetoxyisobutyrate can thus be placed at C-9.

¹ R. L. Baxter, L. Crombie, D. J. Simmonds, and D. A. Whiting, *J.C.S. Chem. Comm.*, 1976, 463.

² R. L. Baxter, L. Crombie, D. J. Simmonds, and D. A. Whiting, *J.C.S. Chem. Comm.*, 1976, 465.

³ United Nations Document MNAR/5/1976.

⁴ United Nations Document MNAR/2/1977.

synthetic specimen for comparison. Analysis is complicated by ready reaction of ethyl α -hydroxyisobutyrate with ethoxide ion ($B_{AL}2$ perhaps) to give the free acid whose water solubility and steam volatility makes micro-detection difficult.

Cathedulin K-2 is thus (3) in which only the relative siting of acetate and α -acetoxyisobutyrate remains unclarified.† Cathedulin K-1 has one more acetate than K-2 and this occupies C-2 (2-H, δ 5.21): it is thus (4). Cathedulin K-6 has one less acetate than K-2 and has a free primary hydroxy-group at C-15 (15-H_a, δ 4.24; 15-H_b, 4.62; J_{AB} 13 Hz), leading to structure (5). Cathedulin K-15 has one acetate less than K-6. The sesquiterpene core protons are undisturbed in the 1H n.m.r. spectrum whilst the $O \cdot CMe_2CO_2R$ signals move slightly (δ 1.65 and 1.59 in K-6 to δ 1.51 in K-15): a parallel movement is found in models (δ 1.55 for 3β -cholestanyl α -acetoxyisobutyrate and δ 1.40 for 3β -cholestanyl α -hydroxyisobutyrate). Further evidence that the α -hydroxyisobutyrate residue is present in unacetylated form in K-15 is the absence of a fragment at m/e 129 in the mass spectrum. This, assigned to the acetoxyisobutyrate fragment $[MeCOO \cdot CMe_2CO]^+$, is evident in K-1, K-2, and K-6.

The new information allows development of the structures for cathedulins E-3 to E-6, for which incomplete formulations have been assigned.¹ E-3 has now been crystallised, m.p. 245–248 °C, and improved spectral data (^{13}C and 1H n.m.r.) confirm earlier deductions and allow recognition of an α -acetoxyisobutyrate residue. Previously, only three acetates had been recognised (by n.m.r.) in E-3: a fourth, at C-2, and shielded, has now been located at δ 1.37. Quantitative ethanolysis is in agreement with the presence of four acetates, and α -hydroxyisobutyric acid has been chemically identified from E-3. Re-evaluation of the mass-spectral data now allows assignment of the formula $C_{54}H_{80}N_2O_{23}$ (Found: $M^+ 1104.360$, requires 1104.360) to E-3. All 54 carbons and 60 hydrogens can be observed in the new spectra, leaving no unassigned resonances. Both crystalline and non-crystalline specimens of E-3, however, show a small peak at m/e 1166, originally assumed to be probably the molecular ion. It apparently belongs to a contaminating companion alkaloid in which one acetate is replaced by a benzoate. Cathedulin E-3 is thus allocated structure (7), the only ambiguity being the placing of the α -acetoxyisobutyrate residue: work on this aspect is in hand.

Cathedulins E-4, E-5, and E-6 present a similar situation and are given molecular formulae $C_{52}H_{58}N_2O_{22}$ ($M^+ 1062$), $C_{59}H_{64}N_2O_{23}$ ($M^+ 1168$), and $C_{57}H_{62}N_2O_{22}$ ($M^+ 1126$) respectively. From our earlier information¹ the formulation of E-4, E-5, and E-6 can be developed analogously to E-3 to give (9), (10), and (11) respectively. Cathedulin K-12 has 1H n.m.r. data very similar to those of E-3, but with the cathate diester bridge signals replaced by those of separate gallate and nicotinate esters. Structure (8) accounts for the known information: K-12 may well be the immediate biosynthetic precursor of E-3.

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