## Novel Ring Transformation of a Thiophen into a Pyrrole<sup>1</sup>

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Summary Whereas anils (4) derived from 3-nitrothiophen-2-carbaldehyde undergo reductive cyclisation with triethyl phosphite to give the expected 2-arylthieno-[3,2-c]pyrazoles (5), the isomeric anils (6) and (7), derived from 2-nitrothiophen-3-carbaldehyde, give the 1-arylpyrrole-3-carbonitriles (8) and (9), respectively, as the major products.

REDUCTIVE cyclisation of anils with the general structure (1) with tervalent phosphorus reagents yields the corresponding 2-aryl[1]benzothieno[3,2-c]pyrazole (3).<sup>2</sup> By contrast, however, the isomeric anils (2) give mixtures of the corresponding benzothieno[2,3-c]pyrazole and benzo[b]-thiophen-3-carbonitrile.<sup>2</sup> We have proposed a mechanism which accounts for the formation of the latter compound through an interesting ring-opening-ring-closure sequence of the thiophen ring. In the light of these results it was of interest to study the behaviour of the corresponding thiophen anils.

$$(1), R^{1} = CH * NAr, R^{2} = NO_{2}$$

$$(2), R^{2} = NO_{2}, R^{2} = CH * NAr$$

$$(3)$$

$$(3)$$

$$(3)$$

$$(4)$$

$$(5)$$

As expected, treatment of anils with the general structure (4) with triethyl phosphite in t-butylbenzene (1:3) gave the corresponding thieno[3,2-c]pyrazole (5).3 Compound (6), however, after 14 h under these conditions gave 1-phenylpyrrole-3-carbonitrile (8) as the major product (55% yield), isolated by successive distillation (10-3 mmHg) and chromatography on alumina, b.p.  $105\,^{\circ}\text{C}$  at  $5\times10^{-3}\,\text{mmHg}$ (Kugelrohr apparatus);  $\nu_{max}$  2255 cm<sup>-1</sup> (CN);  $\tau$  (CDCl<sub>3</sub>) 2·40—3·50 (6H, m, Ph and 2-H), 3·03 (1H, m, 5-H), and 3.48 (1H, m, 4-H); m/e 168.0692 (M+). This compound was an unstable, yellow oil and it was not possible to obtain a satisfactory elemental analysis, although the values obtained were close to those expected. However, compound (8) was synthesised unambiguously from the known4 ethyl 3-cyano-1-phenylpyrrole-2-carboxylate by hydrolysis to the corresponding acid, m.p. 198-200 °C (from hexanechloroform) (lit.4 194—195 °C);  $\nu_{max}$  (Nujol) 3300—2400br (OH), 2230 (CN), and 1680 cm<sup>-1</sup> (CO), and decarboxylation

of the acid (in 66% yield) with copper in quinoline. Reductive cyclisation of the anil (7) with triethyl phosphite in t-butylbenzene likewise gave 1-(4-dimethylaminophenyl)-pyrrole-3-carbonitrile (70% crude), m.p. 150—151 °C (sublimed at 65—69 °C and 7 × 10<sup>-3</sup> mmHg);  $\nu_{\rm max}$  2255 cm<sup>-1</sup> (CN);  $\tau$  (CDCl<sub>3</sub>) 2-60 (1H, m, 2-H), 2-79 and 3-25 (4H, 2 × d, J 9-0 Hz, ArH), 3-09 (1H, m, 5-H), 3-45 (1H, m, 4-H), and 6-99 (6H, s, NMe<sub>2</sub>); m/e 211·1104 ( $M^+$ ). This compound sublimed as white crystals which turned pale blue immediately on exposure to air.

CH=NAr

$$S = NO_2$$

(6), Ar = Ph

(7), Ar =  $C_6H_2$  NMe<sub>2</sub>-p

CH=NAr

 $CH = NAr$ 
 $CH =$ 

Scheme

A mechanism which accounts for our observations is shown in the Scheme; the formation of the bicyclic intermediate and the ring-opening steps are analogous to those given in our earlier, related communication.<sup>2</sup> Possible reasons for the difference in behaviour of anils (2) and (6) will be examined in the full paper later.

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See H. C. Van Der Plas, 'Ring Transformations of Heterocycles,' Vol. 1, Academic Press, London and New York, 1973, ch. 3, p. 198.

<sup>&</sup>lt;sup>2</sup> K. E. Chippendale, B. Iddon, and H. Suschitzky, *J.C.S. Perkin I*, 1973, 129. <sup>3</sup> V. M. Colburn, B. Iddon, and H. Suschitzky, unpublished results.

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