Palladium(II)-catalysed Asymmetric Cyclisation of 2-Allylphenols by the Use of Optically Active β-Pinene

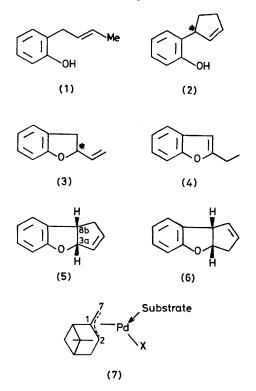
By TAKAHIRO HOSOKAWA,* SHYOGO MIYAGI, SHUN-ICHI MURAHASHI, and AKIO SONODA (Department of Chemistry, Faculty of Engineering Science, Osaka University, Machikaneyama, Toyonaka, Osaka, Japan 560)

Summary The palladium(II)-catalysed asymmetric cyclisation of 2-(but-2-enyl)phenol (1) to optically active 2,3dihydro-2-vinylbenzofuran (3) can be achieved in 12%optical yield by using a catalytic amount of $(-)-\beta$ pinene, while in the presence of excess of β -pinene, cyclisation of (1) does not occur. The oxidative ring closure of 2-allylphenols by palladium(II) salts generally gives 2-substituted benzo- and/or dihydrobenzo-furans,¹ and cyclisations of this type have been shown to be useful for synthesizing a variety of heterocyclic compounds.² We report here that the Pd^{II}-catalysed asymmetric synthesis of 2,3-dihydrobenzofurans from the 2-

An excess of (S)-2,3-dihydro-2-vinylbenzofuran (3) was obtained by cyclisation of 2-(but-2-envl)phenol (1) with palladium(II) acetate in the presence of (-)- β -pinene in 12% optical yield as follows. Into a heterogeneous stirred mixture of Pd(OAc)₂ (1 mmol), Cu(OAc)₂·H₂O (10 mmol), and (-)- β -pinene {1 mmol; $[\alpha]_{D}^{25} - 20.0$ (neat) } in MeOH-H₂O (19:1 v/v) was added the allylphenol (1; 10 mmol; trans/cis 93/7) at 35 °C under O₂ (ca. 1 atm.). The allylphenol (1) was completely consumed within 3 h. After the usual work-up, distillation gave a 62% combined yield of 2,3-dihydro-2-vinylbenzofuran (3) {[α]_D²⁶ + 2.79 (CCl₄, c 4.19) after g.l.c.} and 2-ethylbenzofuran (4) in the ratio 89:11. The optical yield and absolute configuration of (3) were determined by conversion into ethyl 2,3-dihydrobenzofuran-2-carboxylate of known optical rotation and absolute configuration,³ by oxidation with potassium permanganate followed by esterification.

When excess of β -pinene was used, cyclisation of (1) did not occur; in the presence of 2 equiv. of β -pinene with respect to $Pd(OAc)_2$, the extent of cyclisation to (3) was < 1%after 24 h. In contrast, in the presence of excess of (-)- or (+)- α -pinene, the cyclisation proceeded smoothly, \dagger but the product (3) was optically inactive. These results indicate that co-ordination of the substrate molecule to Pd^{II} is in competition with co-ordination of the olefinic additives, and that the exo-double bond of β -pinene more easily coordinates to Pd^{II} than does the internal double bond of (1). Co-ordination of β -pinene probably leads to a catalytic species having a π -allylic ligand such as (7), since bis[acetoxy- $(7,1,2-\eta$ -pinene)palladium(II)]⁺ reacts catalytically with (1) [(1)/Pd=10/1] under the foregoing conditions, affording optically active (3) in similar optical yield: $[\alpha]_{p}^{25} + 3.01$ $(CCl_4, c \ 7.13) [(3)/(4) = 87/13 \text{ in } 71\% \text{ combined yield}; 5 \text{ h}].$ In contrast to the cyclisation of (1), (\pm) -2-(cyclopent-2-

envl)phenol (2) reacts with $Pd(OAc)_2$ even in the presence of



excess of β -pinene. The use of 10 equiv. of (-)- β -pinene with respect to Pd(OAc)₂ under similar conditions led to a 93:7 mixture of (5) and (6) having a cis-configuration at C-3a and C-8b⁴ (16% yield after 3 h); $[\alpha]_D^{25}$ of mixture + 7.7 (CCl₄, $c \ 0.57$). In this case the recovered (2) was also optically active, $[\alpha]_D^{25} + 0.31$ (CCl₄, c 3.76). Thus, in this system, co-ordination of the carbon-carbon double bond of (2) to Pd^{II} is not prevented to a significant extent by the presence of excess of β -pinene.

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[†] When 10 equiv. of α-pinene was used, the cyclisation of (1) was completed within 2 h. Commercially available (-)- or (+)-α-pinene contains ca. 5% of β-pinene, which was removed by treatment with $Pd(OAc)_2$ followed by distillation.

 \ddagger Prepared by treatment of the π -allyl palladium(II) chloride dimer derived from (-)- β -pinene and PdCl₂ (B. M. Trost and P. E. Strege, Tetrahedron Letters, 1974, 2603) with AgOAc.

¹T. Hosokawa, H. Ohkata, and I. Moritani, Bull. Chem. Soc. Japan, 1975, 48, 1533; T. Hosokawa, S. Yamashita, S-I. Murahashi, and A. Sonoda, *ibid.*, 1976, 49, 3663; T. Hosokawa, S. Miyagi, S-I. Murashi, and A. Sonoda, J. Org. Chem., in the press.

^aD. E. Korte, L. S. Hegedus, and R. K. Wirth, J. Org. Chem., 1977, **42**, 2674; A. Kasahara, T. Izumi, K. Sato, M. Maemura, and T. Hayasaka, Bull. Chem. Soc. Japan, 1977, **50**, 1899, and references cited therein. ^aD. M. Bowen, J. I. DeGraw, Jr., V. R. Shah, and W. A. Bonner, J. Medicin. Chem., 1963, **6**, 315; W. A. Bonner, N. I. Burke, W. E. Fleck, R. K. Hill, J. A. Joule, B. Sjöberge, and J. H. Zalkow, Tetrahedron, 1964, **20**, 1419. ⁴ The cis-configuration was assigned on the basis of n.m.r. spectroscopy; see, T. Hosokawa, S. Miyagi, S-I. Murahashi, A. Sonoda, V. Mathemar, S. Chem. 1972, **42**, 210.

Y. Matsuura, S. Tanimoto, and M. Kakudo, J. Org. Chem., 1978, 43, 719.