

Rearrangement of 4-Ylidenebutenolides to Cyclopentene-1,3-diones: Synthesis of Calythrone and Related Compounds

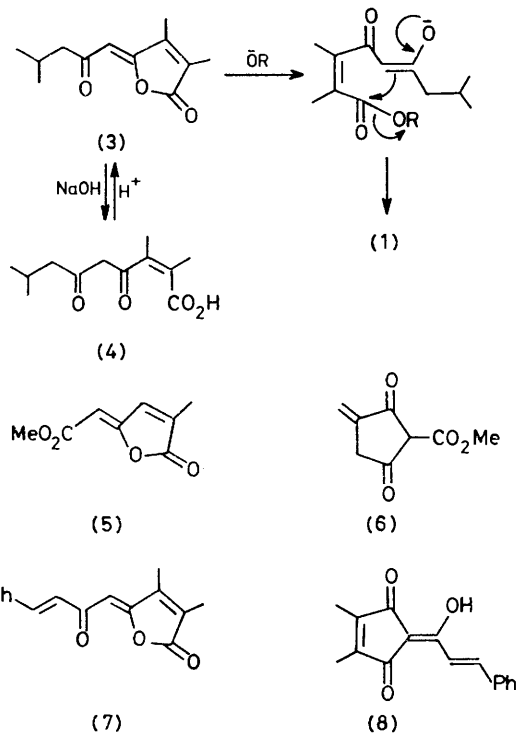
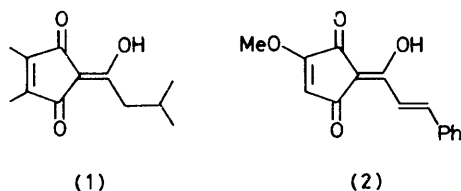
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Summary Treatment of 4-ylidenebutenolides with NaOMe in MeOH results in rearrangement to the corresponding cyclopentene-1,3-diones in high yield; the general method

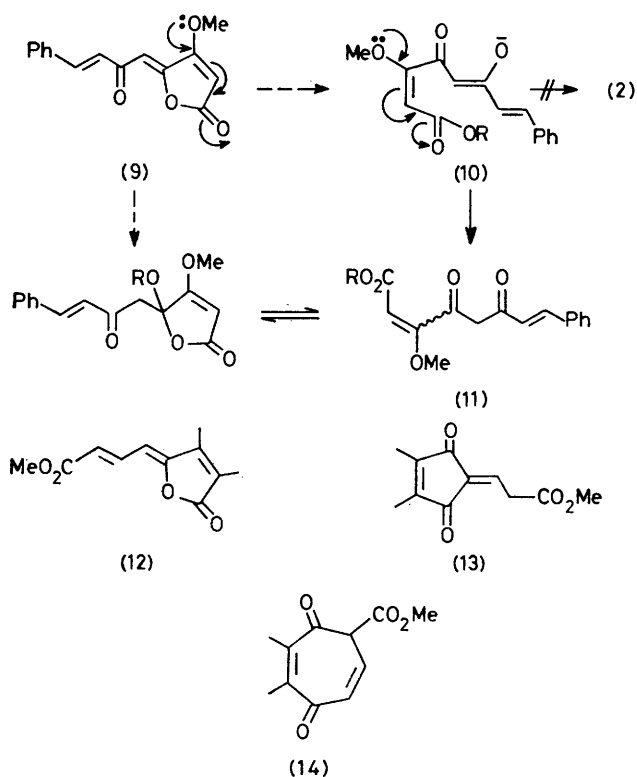
is applied in a synthesis of calythrone (**1**) from *Calythrix tetragona*, and the related cyclopentene-1,3-diones (**6**), (**8**), and (**13**).

CALYTHRONE (**1**) and lucidone (**2**) are representative members of a unique group of naturally occurring cyclopentene-1,3-diones. Their biosynthesis has been suggested to involve ring contraction of appropriate acylphloroglucinol derivatives as a key stage.^{1,2} The 4-ylidenebutenolide ring system (*viz.* **3**) is isomeric with the cyclopentene-1,3-dione system, and aromatic molecules are obligatory intermediates in the biosynthesis of several natural members of this class of compound.³ The familial relationship between the two ring systems (*viz.* **1** and **3**) can be construed into suggesting that 4-ylidenebutenolides are implicated in the biosynthesis of cyclopentene-1,3-diones from aromatic precursors. In this paper we describe a synthesis of calythrone (**1**), and the related compounds (**6**), (**8**), and (**13**) modelled on this hypothesis.



Calythrix tetragona, was originally assigned the ylidenebutenolide structure (**3**) by Penfold and Simonsen,⁶ but this was modified to (**1**) by Birch and his co-workers⁷ largely on the basis of u.v. and i.r. absorption data. The ease with which (**3**) is rearranged to (**1**) in NaOMe-MeOH⁸ might alternatively suggest that natural calythrone actually has the ylidenebutenolide structure (**3**), and that the cyclopentenedione (**1**) is an artifact produced during basic extraction and isolation. Indeed dissolution of the butenolide (**3**) in 10% aqueous sodium carbonate (25 °C, 24 h) followed by acidic work-up led to large amounts of the dione (**1**; 35%) together with the oxo-acid (**4**; 65%), ν_{\max} 3350, 1750, and 1690 cm⁻¹, τ 3.5 (OH) and 7.22 (CO.CH₂ CO). Treatment of (**3**) with 2M NaOH, and acidic work-up however gave entirely the oxo-acid (**4**) which could be converted back into (**3**) on brief treatment with hot acid.

In a similar manner the ylidenebutenolide (**5**) on treatment with NaOMe-MeOH led to the dione (**6**), resulting from simultaneous rearrangement and allylic isomerisation,[†] and the cinnamoyl-butenolide (**7**) gave the analogue (**8**) of natural lucidone (**2**). The ylidenebutenolide isomer (**9**) of



Treatment of dimethylmaleic anhydride with 2-methylbutyrylmethylenetriphenylphosphorane (refluxing toluene; 16 h) gave the *Z*-butenolide (**3**), ν_{\max} (film) 1785, 1760, and 1645 cm⁻¹, τ 4.41 (:CH), in 75% yield.⁴ The butenolide was treated with sodium methoxide in dry methanol, and acidic work-up led to calythrone (**1**) in 80% yield, with identical spectral properties to those reported;⁵ copper salt m.p. 208–210 °C (lit.⁶ m.p. 208 °C). This synthesis can be commended not only for its brevity over existing routes to calythrone,⁵ but also for its relevance to a possible biosynthetic route. Natural calythrone, from the oil of

lucidone itself was also synthesised, but attempts to rearrange the molecule to lucidone under a variety of reaction conditions have thus far been unsuccessful; the only product isolated has been the corresponding acyclic 4-oxo-unsaturated ester (**11**). This was found to be the outcome with a number of 3-methoxy-4-ylidenebutenolides investigated, and is presumably associated with the diminished electrophilicity of the acyl groups in the molecules [and/or the intermediates (*viz.* **10**) between (**9**) and (**2**)], as a result of their vinylogous disposition relative to the 3-methoxy substituents.

As a corollary to these investigations we also examined the rearrangement of the *E,E*-ylidenebutenolide (**12**) containing additional conjugation in its side chain. Treatment of (**12**) with NaOMe-MeOH led entirely to the cyclopentene-1,3-dione (**13**) (90%), ν_{\max} 1735 and 1685 cm^{-1} , τ 3.06 (t, J 7 Hz, :CH), 5.9 (d, J 7 Hz, $\text{CH}_2\text{-CH}$), 6.08

(OMe), and 7.73 ($2 \times \text{:CMe}$); interestingly none of the isomeric 7-ring dione (**14**) could be detected in the crude reaction mixture.

We thank the S.R.C. for a studentship to D. R. G.

(Received, 21st July 1978; Com. 787.)

† We are indebted to A. K. Edwards of this Department for this observation.

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⁴ Cf. M. J. Begley, D. R. Gedge, and G. Pattenden, *J.C.S. Chem. Comm.*, 1978, 60; C. F. Ingham, R. A. Massy-Westropp, G. D. Reynolds, and W. D. Thorpe, *Austral. J. Chem.*, 1975, **28**, 2499.

⁵ M. Vandewalle, L. Van Wijnsberghe, and G. Witvrouwen, *Bull. Soc. chim. belges*, 1971, **80**, 39; M. Elliott, N. F. Janes, and K. A. Jeffs, *J. Chem. Soc. (C)*, 1969, 1845.

⁶ A. R. Penfold and J. L. Simonsen, *J. Chem. Soc.*, 1940, 412.

⁷ A. J. Birch, *J. Chem. Soc.*, 1951, 3026; A. J. Birch and R. J. English, *ibid.*, 1957, 3805.

⁸ Cf. R. G. Alexander, D. R. Buckle, and J. M. Tedder, *J.C.S. Perkin I*, 1977, 1191; P. Hrnčiar, and L. Kuruc, *Chem. Zvesti*, 1967, **21**, 267 (*Chem. Abs.*, 1967, **67**, 73304v) and refs. therein.