

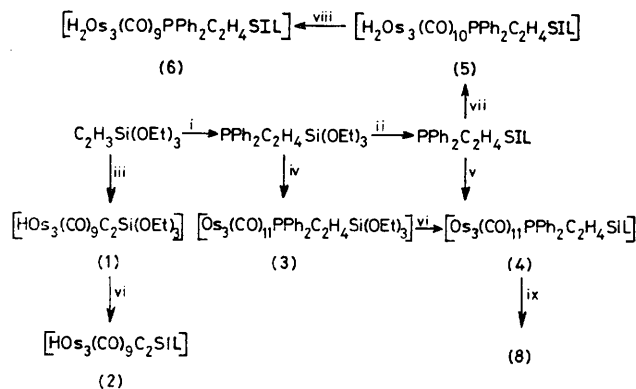
Anchoring of Osmium Clusters to Silica†

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Summary Clusters of the types $[\text{Os}_3(\text{CO})_{11}\text{PPh}_2\text{R}]$, $[\text{H}_2\text{Os}_3(\text{CO})_{10}\text{PPh}_2\text{R}]$, and $[\text{HOs}_3(\text{CO})_9\text{CCR}']$ have been anchored to silica and characterised by i.r. spectroscopy (R = $\text{C}_2\text{H}_4\text{SIL}$, R' = SIL, SIL = silica gel.)

INTERACTION of metal carbonyl clusters with refractory supports has been studied recently as a potential way of preparing a highly uniform dispersion of metal particles.¹ Two problems arise from this approach. Firstly sintering can occur during the attempted thermal labilisation of the carbonyl ligands² and secondly these materials are difficult to characterise. We report specific syntheses of a variety of osmium carbonyl cluster complexes anchored to silica by appended ligands. This approach was adopted to counter the two problems of direct interaction between the cluster and the support; sintering should be less acute and characterisable analogues can be readily prepared.



SCHEME. i, PPh_2H , u.v. ii, Silica gel, toluene, reflux. iii, $[\text{Os}_3(\text{CO})_{12}]$, n-octane, reflux. iv, $[\text{Os}_3(\text{CO})_{12}]$, toluene, reflux. v, $[\text{Os}_3(\text{CO})_{11}\text{MeCN}]$, CH_2Cl_2 , 20 °C. vi, Silica gel, benzene, reflux. vii, $[\text{H}_2\text{Os}_3(\text{CO})_{10}]$, hexane, 20 °C. viii, Hexane, reflux. ix, n-Nonane, reflux.

The reactions carried out are presented in the Scheme. $\text{C}_2\text{H}_3\text{Si}(\text{OEt})_3$, $\text{PPh}_2\text{C}_2\text{H}_4\text{Si}(\text{OEt})_3$, and $\text{PPh}_2\text{C}_2\text{H}_4\text{SIL}$ (SIL = silica gel) were used as complexing agents.³ Under conditions similar to those in which ethylene reacts with $[\text{Os}_3(\text{CO})_{12}]$ to give $[\text{H}_2\text{Os}_3(\text{CO})_9\text{CCH}_2]$,⁴ vinyltriethoxysilane yields $[\text{HOs}_3(\text{CO})_9\text{CCSi}(\text{OEt})_3]$ (1) in high yield. This reaction is unusual in that the vinyl group is completely dehydrogenated; complexes of this type are normally prepared from terminal acetylenes.⁵ Refluxing a benzene solution of (1) with silica gel leads to (2), the anchored version of (1), as identified by its carbonyl vibrations.

$[\text{Os}_3(\text{CO})_{12}]$ and $\text{PPh}_2\text{C}_2\text{H}_4\text{Si}(\text{OEt})_3$ in refluxing toluene give rise to $[\text{Os}_3(\text{CO})_{11}\text{PPh}_2\text{C}_2\text{H}_4\text{Si}(\text{OEt})_3]$ (3) and the di- and tri-substituted clusters. The anchored complex (4) was prepared by interaction of (3) with silica gel and also by the

reaction of $[\text{Os}_3(\text{CO})_{11}\text{MeCN}]$ ⁶ with $\text{PPh}_2\text{C}_2\text{H}_4\text{SIL}$ at room temperature. The phosphinated silica complexes were characterised by comparison of their carbonyl stretching vibrations with those of the PPh_2Et analogue† (see Table). The i.r. spectra of (4) and $[\text{Os}_3(\text{CO})_{11}\text{PPh}_2\text{Et}]$ are displayed in the Figure. The close agreement between them indicates

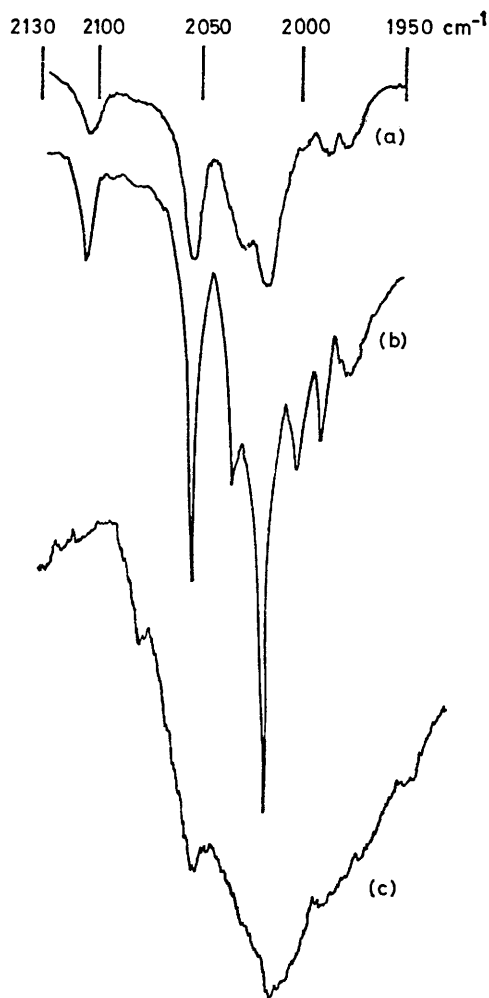


FIGURE. I.r. spectra of (a) (4) in Nujol mull, (b) $[\text{Os}_3(\text{CO})_{11}\text{PPh}_2\text{Et}]$ in cyclohexane, and (c) the product of refluxing $[\text{Os}_3(\text{CO})_{12}]$ with $\text{PPh}_2\text{C}_2\text{H}_4\text{SIL}$ in toluene in a Nujol mull.

that (4) was prepared specifically by these two routes. In contrast, direct interaction of $[\text{Os}_3(\text{CO})_{12}]$ and $\text{PPh}_2\text{C}_2\text{H}_4\text{SIL}$ in refluxing toluene gives a material which exhibits a very broad spectrum and may contain a mixture of mono- and di-substituted products with the former predominating.

† No reprints available.

‡ A simple procedure was recently adopted to characterise a polymer bound $\text{Ir}_4(\text{CO})_{11}\text{PPh}_2\text{R}$ complex (J. J. Rafalko, J. Lieto, B. C. Gates, and G. L. Schrader, Jr., *J.C.S. Chem Comm.*, 1978, 540).

TABLE. Carbonyl stretching frequencies of anchored and free clusters^a

Compound	$\nu_{\text{CO}}/\text{cm}^{-1}$
(1)	2102.5 w, 2077 vs, 2054 vs, 2020 vs, 2016.5 sh, 1983 sh, 1981 m
(2)	2078 s, 2054 s, 2022 s, 2014 s, 1983 m
[Os ₃ (CO) ₁₁ PPh ₂ Et]	2107 m, 2054s, 2035 s, 2019 vs, 2002 m, 1990 m, 1976 m, 1950 w
(4)	2107 m, 2054 s, 2035 s, 2019 vs, 2003 m, 1990 m, 1980 m
[H ₂ Os ₃ (CO) ₁₀ PPh ₂ Et]	2105.5 m, 2066 s, 2051 s, 2024.5 vs, 2016 sh, 2007 ms, 1999 m, 1982.5 m, 1972 m
(5)	2103 m, 2065 s, 2049 s, 2023 vs, br, 1984 m
[H ₂ Os ₃ (CO) ₉ PPh ₂ Et]	2097.5 m, 2054 s, 2019 s, 2010 vs, 1994 w, 1987.5 w, 1977.5 m, 1970 m, 1949 m
(6)	2095 m, 2063 sh, 2053 s, 2034 sh, 2015 vs, 1979 sh
(7)	2086.5 m, 2065 vs, 2044 s, 2010 sh, 2007 s, 1991 s, 1983 w, 1972 w
(8)	2094 w, 2075 m, 2061 sh, 2043 s, 2010 vs, br, 1960 sh

^a Spectra of discrete complexes were recorded in cyclohexane solution and those of anchored materials in Nujol mulls.

Anchoring of [H₂Os₃(CO)₁₀] can also be effected at room temperature since it forms an adduct (5) with appended as well as free phosphine.⁷

Preliminary work on comparative reactivity studies of these free and anchored clusters indicates that identical reactivity between the two states cannot be assumed. Like its PPh₂Et counterpart, ^{7a} (5) readily loses one carbonyl group to form [H₂Os₃(CO)₉PPh₂C₂H₄SIL] (6) as the major product. However there is a divergence in behaviour between the anchored complex (4) and [Os₃(CO)₁₁PPh₂Et].

Refluxing the latter complex in n-nonane for 5 h causes conversion (*ca.* 80%) into [Os₃(CO)₉PEtC₆H₄] (7). Compound (4), however, yields an off-white material (8) with a substantially different i.r. spectrum (see Table).

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