## Synthesis of Tris(imino)thietans; X-Ray Crystal Structure of 2,3-Bis-(t-butylimino)-4-p-tolylsulphonyliminothietan

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Summary The first examples of tris(imino)thietans (2) have been prepared by treatment of arylsulphonyl isothiocyanates with t-butyl isonitrile; the structure of the NN'-di-t-butyl-N''-p-tolylsulphonyl derivative has been determined by X-ray crystallography.

EXTERNALLY stabilized 1,3-dipoles of type (1) are probably intermediates in the reversible transfer of sulphur from aryl isothiocyanates to aryl isonitriles [equation (1),  $R^1$  and  $R^2$  = aryl]. By varying the substituents  $R^1$  and  $R^2$ , we were

$$R^{1}NC + R^{2}NCS \Longrightarrow R^{1}N = C \longrightarrow S$$

$$R^{1}N = C \longrightarrow S$$

$$R^{2} \longrightarrow R^{2}$$

$$R^{2} \longrightarrow R^{2}$$

$$R^{1}NCS + R^{2}NC$$

$$R^{1}NCS + R^{2}NC$$

able to trap the dipolar species with a second molecule of isonitrile to yield the title compounds (2a-c):  $^2$ (2a), 60% yield, m.p.  $131\cdot 5$  °C (decomp.) (from Et<sub>2</sub>O); (2b), 65%, m.p. 140-142 °C (decomp.) (from Et<sub>2</sub>O); (2c), 61%, m.p. 86-87 °C (from hexane).† Thus, when arylsulphonyl isothiocyanates were treated with 2 equiv. of t-butyl isonitrile in ether at room temperature (1 day), the yellow adducts (2a and b) crystallized out upon cooling of the mixture.

The adduct (2c) was isolated by column chromatography on silica gel with ether—n-hexane (50:50) as eluant. This also furnished a side product (6%, m.p. 178 °C, decomp.) which was identified as (3) by X-ray analysis.

$$P-XC_{6}H_{4}SO_{2}N=C=S+2Bu^{t}NC \longrightarrow Bu^{t}N$$

$$Bu^{t}NH \longrightarrow SO_{2}C_{6}H_{4}X-P$$

$$(2)$$

$$a_{1}X=Me$$

$$b_{1}X=C1$$

$$c_{2}X=H$$

$$(3)$$

The i.r. spectra (KBr) of (2a—c) showed characteristic imine stretching absorptions at 1720—1730w, 1660m, and 1600—1610s, br cm<sup>-1</sup>, while the <sup>1</sup>H n.m.r. spectra indicated the presence of two t-butyl groups ( $\delta$  1·40 and 1·50). The

† All compounds gave satisfactory elemental analyses. Yields were not optimized; those for (2a) and (2b) refer to the recrystallized compounds, while that for (2c) is for the compound following column chromatography.

<sup>13</sup>C n.m.r. spectra (CDCl<sub>3</sub>), showed ring carbon absorptions at  $\delta$  145—146, 162—163, and 176—177 p.p.m. The structure of (2a) was determined by X-ray analysis, and a drawing of the structure is shown in the Figure, which

indicates the stereochemistry about the three imine double bonds.‡ Crystal data:  $C_{18}H_{25}N_3O_2S_2$ , monoclinic, space group  $P2_1/n$ ,  $a = 22 \cdot 104(9)$ ,  $b = 8 \cdot 616(3)$ ,  $c = 11 \cdot 038(3)$  Å,  $\beta = 100.83(3)^{\circ}$ ; Z = 4. Intensities were determined with a Syntex diffractometer with Mo- $K_{\alpha}$  radiation;  $2\theta_{
m max}$ = 42°. The structure was solved by MULTAN 773 and refined with the X-RAY 72 system4 to an R-value of 0.049 for 1504 significant reflections. As expected, the nitrogen atoms are nearly coplanar with the thietan ring. The average deviation from the least-squares plane passing through the seven atoms is 0.037 Å.

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‡ The atomic co-ordinates for this work are available on request from the Director of the Cambridge Crystallographic Data Centre, University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW. Any request should be accompanied by the full literature citation for this communication.

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