Addition of Some Unsymmetrical Electrophilic Reagents to Cyclic $\alpha\beta$ -Unsaturated Silanes

By ELIZABETH J. THOMAS and GORDON H. WHITHAM*

(Dyson Perrins Laboratory, South Parks Road, Oxford OX1 3QY)

Summary Additions of some iodine based electrophilic reagents to 1-trimethylsilylcyclohexene (1), and its 4-tbutyl derivative, occur with high regio- and stereoselectivity to give adducts in which iodine is attached to C-1 (2 and 5): these adducts have been transformed into a number of functionalised silanes.

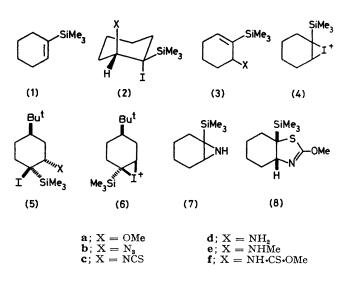
The chemistry of $\alpha\beta$ -unsaturated silanes has recently been fairly extensively investigated^{1,2} particularly from the point of view of electrophilic attack at the carbon atom bonded to silicon with subsequent or concomitant desilvlation,² and several synthetically useful processes have been developed. We report here the addition of some iodine based electrophilic reagents³ (formally IX) to 1-trimethylsilylcyclohexene (1) and a conformationally biassed analogue.

Thus treatment of (1) with (a) I₂-MeOH-sulpholan gave (2a; 77%), (b) I₂-LiN₃-CHCl₃-sulpholan gave (2b; 80%), and (c) I₂-LiSCN-CHCl₂-sulpholan gave (2c; 65%). The constitution of the adducts (2a-c) follows from i.r. and n.m.r. data (particularly band width of the signal ascribable to C-2-H) and from the chemical evidence that treatment with 1,5-diazabicyclo[5.4.0]undec-5-ene in benzene under reflux in each case gave a high yield of the allylically functionalised vinyl silanes (3a), (3b), and (3c), respectively.

The high regioselectivity seems to be most conveniently interpreted in terms of an intermediate iodonium ion (4) which undergoes nucleophilic attack (diaxial opening) at the carbon atom not bonded to silicon. This orientation contrasts with the mode of opening postulated for the corresponding protonated epoxide,⁴ where attack is at the carbon bonded to silicon. To our knowledge this is the first example of such a clear dichotomy, and indeed a parallel between the two processes is often drawn.⁵ Possibly the transition state for the iodonium ion opening has relatively high carbonium ion character and can benefit from the well known ability of silicon to stabilise a β carbonium ion.⁶ Other evidence in favour of the latter suggestion is provided by formation of the isothiocyanate (2c). Addition to cyclohexene under the same conditions gives predominantly trans-1-iodocyclohexyl thiocyanate.7

Addition to 4-t-butyl-1-trimethylsilylcyclohexene gave in high yield the adduct (5a) (from I₂-MeOH) and (5b)

(from I_2 -LiN₃). The high selectivity with this conformationally biassed substrate is explicable in terms of reversible formation of diastereoisomeric iodonium ions⁸ via attack from either side of the olefin, followed by product determining nucleophilic attack (diaxial opening) trans- to t-butyl involving (6) only.



The adducts (2a-c) and (5a, b), as indicated above. provide convenient synthetic precursors to a variety of novel allylically substituted vinyl silanes via elimination. Further transformations, which show some of the potentialities of the adducts, have been effected leading to: (3d) [LiAlH₄ reduction of (3b)]; (3e) [LiAlH₄ reduction of (3c)]; and (3f) [treatment of (3c) with KOH-MeOH]. We have also shown that reduction of (2b) with LiAlH₄ gave the aziridine (7) and treatment of (2c) with KOH-MeOH. gave the thiazole (8).

(Received, 15th December 1978; Com. 1331.)

- ¹ M. van der Leij and B. Zwanenburg, Tetrahedron Letters, 1978, 3383; H. Westmijze, J. Meijer, and P. Vermeer, *ibid.*, 1977, 1823;
 ⁸ K. Uchida, K. Utimoto, and H. Nozaki, Tetrahedron, 1977, 33, 2987 and references therein.
 ⁸ T. H. Chan, P. W. K. Lau, and W. Mychajlowskij, Tetrahedron Letters, 1977, 3317; G. Büchi and H. Wüest, *ibid.*, p. 4305;
 K. Yamamoto, O. Nunokawa, and J. Tsuji, Synthesis, 1977, 721 and references therein.
 ⁸ R. C. Cambie, W. I. Noall, G. J. Potter, P. S. Rutledge, and P. D. Woodgate, J.C.S. Perkin I, 1977, 226.
 ⁶ C. M. Robbins and G. H. Whitham, J.C.S. Chem. Comm., 1976, 697.
 ⁶ P. L. Barili, G. Bellucci, F. Marioni, and V. Scartoni, J. Org. Chem., 1975, 40, 3331.

- C. Eaborn, J.C.S. Chem. Comm., 1972, 1255.
 P. D. Woodgate, H. H. Lee, P. S. Rutledge, and R. C. Cambie, Tetrahedron Letters, 1976, 18, 1531.
- ⁸ Cf. G. Bellucci, G. Berti, M. Ferretti, G. Ingrosso, and E. Mastrorilli, J. Org. Chem., 1978, 43, 422.