

Reaction of (η^3 -Acryloyl)tricarbonylferrate with Acyl Halides giving 1:1 Adducts, the Halogenodicyarbonyl(η -2-acyloxy)ferracyclopent-2-en-5-onates

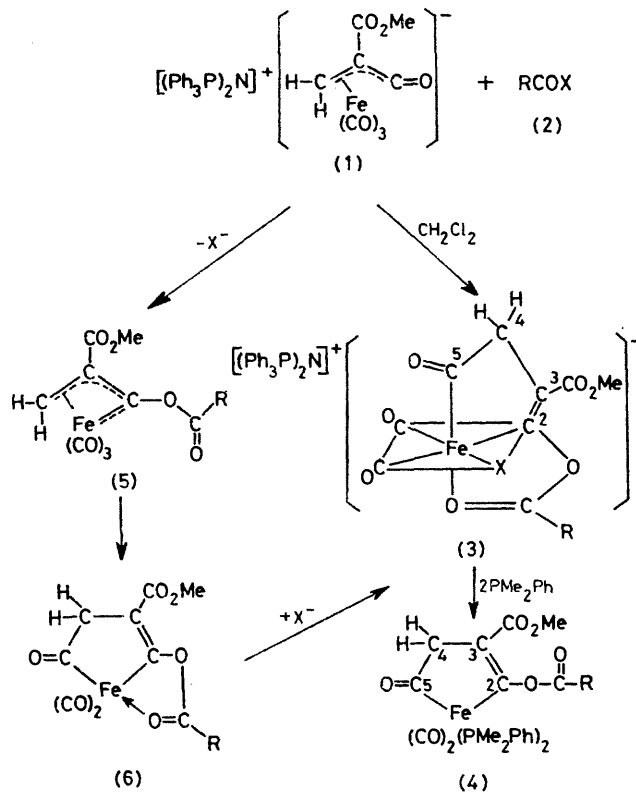
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Summary The (η^3 -acryloyl)tricarbonylferrate (1) reacted with the acyl halides (2) affording the unexpected 1:1 adducts, which were deduced to be the novel halogenodicyarbonyl(η -2-acyloxy)ferracyclopent-2-en-5-onates (3) derived by the carbonylation of an η^3 -vinylcarbene ligand.

THE reactions of acylcarbonylmetallates with acyl halides are useful in organic syntheses.¹ Recently the first examples of η^3 -acryloylferrates (1) were prepared by the reaction of $[\text{FeH}(\text{CO})_3]^-$ with acetylenic compounds; alkylation of (1) afforded the first η^3 -vinylcarbene iron complexes.² We now report an unexpected reaction of (1) with acyl halides giving the corresponding 1:1 adducts, which were deduced to be novel halogenodicyarbonyl-ferracyclopent-2-en-5-onate (1-) complexes containing an additional five-membered chelate ring.

The acryloylferrate² (1) in CH_2Cl_2 was treated with the acyl halides (2a-e) (3 mmol) (Scheme) at 0 °C for 30 min. Addition of diethyl ether gave yellow crystals of (3a-e) in yields of 36–85%. On the basis of the following observations, spectral data, and the 18-electron rule, (3a-e) were deduced to be 1:1 adducts, the bis(triphenylphosphine)iminium halogeno(dicyarbonyl)- η -(2-acyloxy)-3-methoxycarbonylferracyclopent-2-en-5-onates.† The i.r. spectrum of (3a) showed two bands in the region typical of terminal metal carbonyl groups in negatively charged complexes (1998 and 1930 cm^{-1}).³ The spectrum also showed ester $\nu(\text{C}=\text{O})$ at 1688 cm^{-1} ,² $\nu(\text{C}=\text{O})$ (C-co-ordinated to Fe) at 1660 cm^{-1} , $\nu(\text{C}=\text{O})$ (O-co-ordinated to Fe) at 1636 cm^{-1} ,⁴ and $\nu(\text{C}=\text{C})$ at 1600 cm^{-1} . The far-i.r. spectra of (3a-c) showed absorptions at 270, 219, and 181 cm^{-1} respectively showing the presence of an Fe-halogen bond.⁵ The ^1H n.m.r. spectrum of (3a) exhibited an AB quartet at δ 3.62 (1H) and 3.20 (1H) ($|J|$ 19.0 Hz) due to nonequivalent sp^3 CH_2 protons‡ in addition to signals for MeCO, MeO, and $[\text{PPN}]^+$. The ^{13}C n.m.r. spectrum of (3a) showed four signals characteristic of a ferracyclopentenone ring: sp^3 CH_2 at δ 62.4 [t, $J(\text{C}-\text{H})$ 128.9 Hz], $2 \times =\text{C}$ at δ 110.4 (s) and 219.0 (slightly broad), and acyl-iron $\text{C}=\text{O}$ at δ 265.7 p.p.m. [t, $J(\text{CCH})$ 5.9 Hz]. Spectral data for (3b-e) were entirely analogous to those for (3a). Complex (3a) reacted with 2 mol. equiv. of PMe_2Ph at 25 °C affording pale yellow crystals of the bisdimethylphenylphosphine-ferracyclopentenone† (4) (yield 27%) (Scheme) [i.r.: $\nu(\text{C}=\text{O})$ 1690, 1738 (ester), and



SCHEME

- a; R = Me, X = Cl
 b; R = Me, X = Br
 c; R = Me, X = I
 d; R = Et, X = Cl
 e; R = Prⁿ, X = Cl

1630 (C-co-ordinated); $\nu(\text{C}=\text{C})$ 1550 cm^{-1} ; n.m.r. (CDCl_3 ; Me_4Si) ^1H , 2.71 and 3.51 ($|J|$ 21.0 Hz, $-\text{CH}_2-$); ^{13}C , 60.1 (t, C-4), 124.4 (s, C-3), 207.9 (s, C-2), and 274.5 p.p.m. (m, $J_{\text{C}-\text{F}}$ 24.4 Hz, C-5)]. Co-ordination of the 2 PMe_2Ph groups caused the dissociation of the halide ion and the carbonyl group of the enol ester.

The formation of (3) by the reaction of (1) with (2) may be rationalized as follows. The (η^3 -acyloxyvinylcarbene)-iron complex (5) is first formed by acylation on the acyl oxygen atom of the acryloyl group. Co-ordination of the

† Satisfactory analytical data have been obtained.

‡ The absence of a symmetry plane including the methylene carbon proved that the stereochemistry of (3) is as shown in the Scheme.

carbonyl of the ester may then cause migratory insertion of a carbonyl group into the methylene carbon-iron bond§ affording the unstable unsaturated complex (6) whose vacant site may be occupied by a halide ion to give (3).

The complexes (3) and (4) are to our knowledge the first examples of well characterized ferracyclopent-2-en-5-one

complexes which are the keto-form of the 'ferrole' (ferracyclopenta-2,4-diene-2,4-diol) complexes⁶ derived by the reaction of acetylenes with alkali-Fe(CO)₅.

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§ The reaction of [1-3-η-(1-methoxycarbonyl)methoxycarbene]tricarbonyliron with 2 mol. equiv. of PMe₂Ph also gives a ferracyclopent-2-en-5-one complex, the structure of which was determined by X-ray analysis. Spectral data for this complex fully corresponded with those for (3) and (4) (T. Mitsudo, T. Sasaki, Y. Watanabe, Y. Takegami, K. Nakatsu, K. Kinoshita, and Y. Miyagawa, unpublished work). During this reaction the zerovalent iron complex (1) is oxidized to a divalent complex. A similar oxidation was observed in the reaction of an (η³-vinylcarbene)iron complex with CO or PR₃: T. Mitsudo, T. Sasaki, Y. Watanabe, Y. Takegami, S. Nishigaki, and K. Nakatsu, *J.C.S. Chem. Comm.*, 1978, 252.

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⁴ For example, B. L. Booth and R. G. Hargreaves, *J. Chem. Soc. (A)*, 1969, 2766.

⁵ K. Nakamoto, 'Infrared Spectra of Inorganic and Coordination Compounds,' Wiley, New York, 1970, p. 213.

⁶ A. A. Hock and O. S. Mills, *Proc. Chem. Soc.*, 1958, 233.