

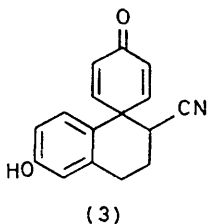
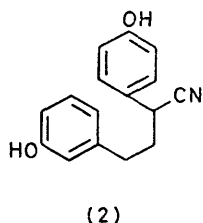
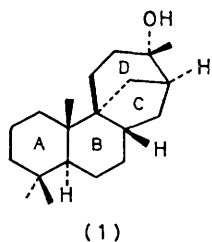
## Regio- and Stereo-controlled Total Synthesis of the Stemodane Nucleus; an Unusual Diterpene Skeleton

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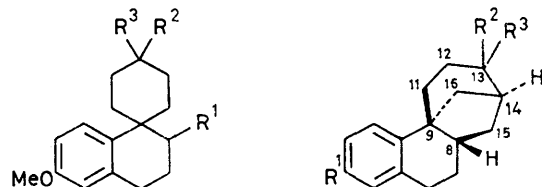
**Summary** Photocatalysed intramolecular oxidative phenol coupling and subsequent regiospecific intramolecular C-H insertion have been utilized as the key steps for a totally synthetic route to the unusual tetracyclic diterpene, deoxystemodin (**1**).

COMPOUND (**1**), which has an unusual carbon framework, and which is a recent addition to the diterpene skeletal variants, is encountered in stemodin and stemodinone.<sup>1</sup> An earlier synthetic approach<sup>2</sup> towards this skeleton envisaged the ABC + D → ABCD sequence for forming the ring system. The present communication describes an improved and stereospecific route (ABD + C → ABCD) towards (**1**) utilizing photocatalysed intramolecular oxidative phenol coupling<sup>3</sup> and regiospecific intramolecular C-H insertion (Bamford-Stevens reaction) as the key steps. This projected synthesis employs fewer steps as compared to the one reported<sup>2</sup> earlier and furnishes an overall yield of 34% from start to finish.



The known conversion<sup>3</sup> (**2** → **3**) via phenol coupling has been modified with an improved yield† by the use of trifluoroacetic acid (TFA) as the co-solvent. Treatment of (**2**) in TFA-ether (3:1) solution with VOCl<sub>3</sub> (2.5 × 10<sup>-3</sup> M solution) at -78 °C and under illumination (250 W tungsten lamp) for 3 h gave (**3**) which on catalytic hydrogenation over Pd-C (10%) and subsequent *O*-methylation by dimethyl sulphate and aqueous alkali afforded the known ketone (**4**). The carbonyl group in this ketone was protected via acetalization with ethylene glycol to obtain (**5**)‡

which on reduction<sup>4</sup> with di-isobutylaluminium hydride furnished the aldehyde (**6**), m.p. 47 °C. Deacetalization in the presence of aqueous acid gave (**7**), m.p. 59 °C. Attempts to effect the C-ring cyclization through internal aldol cyclization under various experimental conditions were not rewarding. The corresponding tosylhydrazone (**8**), m.p. 191 °C, was selectively formed according to the method of Farnum.<sup>5</sup> Thermal decomposition of (**8**) in acetamide solution in the presence of sodium at 140 °C for 3 h gave almost exclusively (87%) the known insertion product (**9**),<sup>2a</sup> found to be homogeneous by g.l.c. Molecular model examinations revealed that intramolecular C-H insertion at C-14 in the reactive conformation of the molecule would be easy from the β-face and this would lead to the desired relative stereochemistry at C-8, C-9, and C-14.



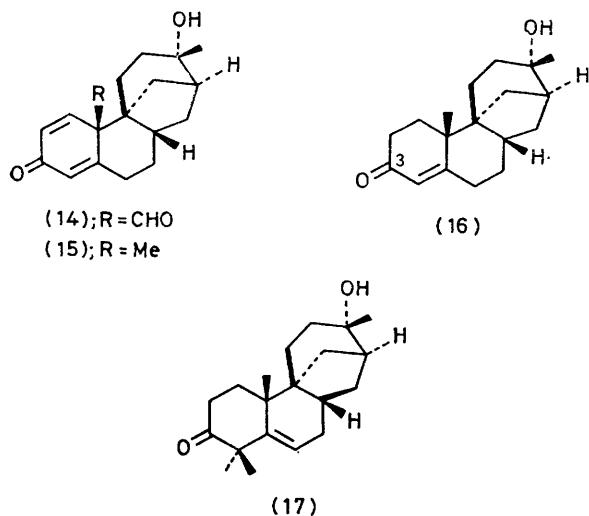
	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>		R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>
(4);	CN	=O		(9);	OMe	=O	
(5);	CN	-O-[CH <sub>2</sub> ]-O-		(10);	OMe	α-Me	β-OH
(6);	CHO	-O-[CH <sub>2</sub> ]-O-		(11);	OMe	β-Me	α-OH
(7);	CHO	=O		(12);	OMe	β-Me	α-OAc
(8);	CH:NNHTs	=O		(13);	OH	β-Me	α-OH

Ts = MeC<sub>6</sub>H<sub>4</sub>SO<sub>2</sub>-*p*

Addition of methylmagnesium iodide to (**9**) gave, as expected,<sup>2b</sup> a 40:60 mixture of the epimeric alcohols (**10**) and (**11**). Separation of the alcohol mixture was effectively done by converting them into their corresponding acetates and by subsequent chromatographic separation over neutral alumina; the desired acetate (**12**) had m.p. 111 °C. Demethylation<sup>6</sup> of (**12**) by boron tribromide and alkaline work-up afforded (**13**), m.p. 132.5 °C. When treated<sup>7</sup> with TFA and hexamethylphosphoric triamide, (**13**) gave the aldehyde (**14**), m.p. 91 °C, λ<sub>max</sub> 246 nm. Steric influence due to the C-16 methylene bridge is expected to direct the introduction of the angular formyl group from the opposite β-face. Based on the significant difference in the electrophilicity of the cross-conjugated dienone and the aldehyde, preferential Wolff-Kishner reduction of (**14**) furnished (**15**), m.p. 75.5 °C. Controlled catalytic hydrogenation (1 mol of H<sub>2</sub>) over Pd-C (10%) resulted in the

† The previously reported yield of this conversion was 49% which has been improved to 85% in this study.

‡ Satisfactory C and H combustion analyses and spectral data (i.r., u.v., and n.m.r.) were obtained for all new compounds. Experimental details will be published elsewhere.



selective reduction of the sterically more exposed<sup>8</sup> double bond to give (16), m.p. 79 °C. The C-13 hydroxy group was converted into the corresponding acetate. Methylation of this acetate employing methyl iodide and potassium t-butoxide afforded, after brief aqueous alkali treatment, the dimethyl derivative (17), 123 °C. Catalytic hydrogenation of (17) over Pd-C (5%) yielded the desired stemodane nucleus (1),<sup>1</sup> m.p. 144 °C,  $\delta(\text{CDCl}_3)$  0.90, 0.93, 0.97, and 1.13 (four 3H singlets due to the methyl protons),  $\delta[(\text{CD}_3)_2\text{SO}]$  3.23 (1H, s, -OH),  $\nu_{\text{max}}$  3590 and 3430  $\text{cm}^{-1}$ . Hydrogenation is expected to have taken place from the less hindered  $\alpha$ -face.

The derivative (1) as obtained from this synthesis was found to be completely identical with the deoxy natural derivatives of stemodine and stemodinone, thus corroborating the structures deduced for these unusual diterpene alcohols.

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<sup>1</sup> P. S. Manchard, J. D. White, H. Wright, and J. Clardy, *J. Amer. Chem. Soc.*, 1973, **95**, 2705.

<sup>2</sup> (a) P. K. Ghosal, D. Mukherjee, and P. C. Dutta, *Tetrahedron Letters*, 1976, 2997; (b) S. Chatterjee, P. K. Ghosal, D. Mukherjee, and P. C. Dutta, *J.C.S. Perkin I*, in the press.

<sup>3</sup> S. Chatterjee and P. K. Ghosal, *Tetrahedron Letters*, 1977, 1451.

<sup>4</sup> J. A. Marshall, N. H. Anderson, and J. W. Schlicher, *J. Org. Chem.*, 1970, **35**, 858.

<sup>5</sup> D. G. Farnum, *J. Org. Chem.*, 1963, **28**, 870.

<sup>6</sup> J. F. W. Meomie, M. L. Watts, and D. E. West, *Tetrahedron*, 1968, **24**, 2289.

<sup>7</sup> W. E. Smith, *J. Org. Chem.*, 1972, **37**, 3972.

<sup>8</sup> H. J. Brodie, M. Hayano, and M. Gut, *J. Amer. Chem. Soc.*, 1962, **84**, 3766.