Asymmetric Total Synthesis of the Antileukaemic Lignans (+)-trans-Burseran and (-)-Isostegane

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Summary (-)-Isostegane and (+)-trans-burseran were synthesized by a highly specific asymmetric pathway from a chiral butenolide.

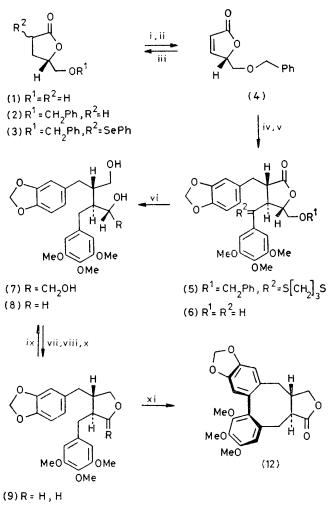
As part of a continuing study directed towards the asymmetric total synthesis of antileukaemic lignan lactones,¹ we describe here the total synthesis of the antileukaemic lignans (+)-trans-burseran $(9)^2$ and (-)-isostegane (12) having the steganacin skeleton³ by a highly specific asymmetric pathway from the easily available chiral butyrolactone (2) as shown in the Scheme.

The key intermediate, the chiral butenolide (4) $\{[\alpha]_{D}^{20}\}$ -107° (ethanol) } was prepared in two steps from (2) in 50% yield.† Compound (4) was confirmed to be optically pure by its catalytic hydrogenation into (1),⁴ whose optical rotation value showed that no racemization had occurred. Conjugate addition⁵ of the trimethoxybenzaldehyde dithioacetal anion to (4) followed by quenching with piperonyl bromide gave (5) { $[\alpha]_{\rm D}^{20} + 1.78^{\circ}$ (chloroform), ca. 95% pure} in 96% yield. Reduction of (5) with Raney nickel afforded, after careful chromatography (SiO₂, ether), the optically pure lactone (6) $\{[\alpha]_{D}^{20} + 63.5^{\circ} \text{ (ethanol), } 57\%\}$ and an isomer $\{[\alpha]_{D}^{20}\}$ $-12\cdot 2^{\circ}$ (ethanol), 1% }. Although the structure of this isomer has not been determined as yet, comparison of ¹³C n.m.r. spectra shows that this compound is a stereoisomer of (6). The structure and optical purity of (6) were confirmed by its conversion into (+)-deoxypodorhizon (11) the optical antipode of the naturally occurring (-)-compound.⁶ This means that conjugate addition of the anion to (4) proceeds in a highly specific asymmetric manner from the β -side of (4) without any base induced racemization of (4).

Reduction of (6) with LiAlH₄ followed by oxidation with sodium metaperiodate to remove the original chiral centre and to transfer the position of the carbonyl group gave the hemiacetal (10) { $[\alpha]_{D}^{26} + 48 \cdot 1^{\circ}$ (ethanol)} in good yield. Collins oxidation of (10) gave optically pure (+)-deoxypodorhizon (11) { $[\alpha]_{D}^{26} + 21 \cdot 3^{\circ}$ (chloroform)} in 95% yield. I.r., ¹H n.m.r., ¹³C n.m.r. spectra and the absolute optical rotation value of (11) were identical with those of natural (-)-deoxypodorhizon.⁶

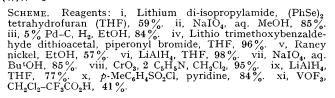
Non-phenolic oxidative coupling of (11) afforded (--)-isostegane (12) {m.p. 169–170 °C, $[\alpha]_2^{23} - 161^\circ$ (chloroform) } which was optically pure. ¹H and ¹³C n.m.r. spectra of optically active (12) were identical with those of the racemic material prepared by the reported method.⁶

Reduction of (11) with LiAlH₄ gave the *trans* diol (8) {m.p. 94.5-95.5 °C, $[\alpha]_{D}^{20} + 31.6^{\circ}$ (chloroform)} in 77% yield. Treatment of (8) with toluene-*p*-sulphonyl chloride in



(10)R = H, OH

(11)R = 0



pyridine afforded (+)-trans-burseran (9) $\{[\alpha]_D^{20} + 37 \cdot 5^\circ (\text{chloroform})\}$ in 84% yield. I.r. and mass spectra of synthetic (9) were not distinguishable from those reported for natural burseran.²

† Satisfactory spectral and analytical data were obtained for all new compounds.

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The first successful, highly specific asymmetric total synthesis of (-)-isostegane and (+)-trans-burseran indicates that the syntheses of other pharmacologically active lignans in optically pure states can be achieved.

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