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New Synthesis of Sultams via readily generated Iminium Ions

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Summary N-Arylsulphonylpropyl chlorides (4) react instantaneously with silver trifluoromethanesulphonate (5) at room temperature to give the iminium salts (8), which have provided convenient routes to the sultams 9,10,11,lla-tetrahydronaphtho[1,8-de]pyrrolo[1,2-b]thiazine

7,7-dioxide (9) and 2,3-dihydro-1H-pyrrolo[1,2-b][1,2,4]-benzothiadiazine 5,5-dioxide (11).

The most successful previous methods for preparing iminium salts were recently reported to include (a) condensation of a carbonyl and a secondary amine, (b) addition to amides, and (c) oxidation of a tertiary amine,¹ and a new method was reported¹ involving heating acid chlorides of α -tertiary amino acids. We now report an alternative method for preparing iminium salts.

Under the conditions under which N-acetyl- or N-trifluoroacetyl-prolyl chloride acylated 1,4-dimethoxynaphthalene in the presence of silver trifluoromethanesulphonate to give the ketones (1a) and (1b) in high yields, N-phenyl-



sulphonylprolyl chloride gave a sulphone (2) and a sulphonamide (3) rather than the expected ketone (1c); this anomalous reaction was shown to proceed *via* an iminium ion.² We have now found that the prolyl chlorides (4a-g)

† Satisfactory elemental analyses were obtained.

reacted instantaneously at room temperature with silver trifluoromethanesulphonate (5) to give the iminium salts (8). Trifluoromethanesulphonic acid but not silver acetate or nitrate led to similar results, suggesting that the reaction proceeds through a mixed anhydride intermediate (Scheme) parallel to that proposed by Effenberger and Epple.³



We have used the iminium ions (8g) and (8d) to prepare the SS-dioxides (9) and (11). The chloride (4g) was readily converted into the iminium ion (8g) which when refluxed in carbon tetrachloride for 18 h gave a brown solid (60% yield), m.p. 130-132 °C, shown to be the expected sultam (9);† M^+ m/e 259, δ (CDCl₃) 2·1 (4H, m, 10- and 11-H₂), 3·2 (2H, m, 9-H₂), 4·4 (1H, t, 11a-H), and 7·3-8·8



(6H, ArH). The sultam (11) was synthesised as follows: the nitro-compound (4d) was converted into the iminium salt (8d) and concentrated aqueous ammonia added. After 30 min at room temperature, chromatography on silica gel gave the required amine (10) as a thick oil in 40% yield: $M^+ m/e \ 271; \delta \ (CDCl_3) \ 2\cdot 1 \ (4H, m, 1-, and \ 2-H_2), \ 3\cdot 6 \ (2H, m, m, 1-, and \ 2-H_2), \ 3\cdot 6 \ (2H, m, m, m, m) \ (2H, m, m, m) \ (2H, m, m) \ (2H, m$

3-H₂), 4·4 (1H, t, 10-H), 6·4 (2H, disappears with D₂O, NH₂), and 7.6-8.2 (4H, Ar-H); vmax (film) 3480, 3380 (NH2), 1540, 850 (NO₂), 1350, and 1150 cm⁻¹ (SO₂N). Reductive cyclisation of (10) by heating with iron in acetic acid⁴ for 8 h gave (11),[†] m.p. 190-192 °C, M⁺ m/e 224, the i.r. spectrum of which is similar to those of the drugs hydrothiazide and hydroflumethiazide,⁵ both of which have the same dihydro-benzothiadiazine nucleus. Further support for structure (11) comes from the n.m.r. spectrum: δ (CD₃-SOCD₃) 2·1 (4H, m, 1- and 2-H₂), 3·4 (2H, m, 3-H₂), 4·5 (1H, t, 10a-H), 7·2-8·0 (4H, ArH), and 10·3 (1H, br s, NH).

Hydrothiadiazides are used as diuretic and hypotensive agents. This versatile and simple synthesis, which is expected to be applicable to most α -secondary aminoacids, therefore provides a convenient route to many of these drugs.

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