## New Synthesis of $\beta$ -Lactams

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Summary The insertion of carbon monoxide into various 2-bromo-3-aminopropene derivatives, (1) and (7–11) in the presence of a catalytic amount of  $Pd(OAc)_2$  and  $PPh_3$  gave the corresponding  $\alpha$ -methylene- $\beta$ -lactams (2) and (12–16) in fairly good yields.

The development of  $\beta$ -lactam antibiotics prompted various methods for synthesising  $\beta$ -lactams, including [2+2] cycloadditions, intramolecular cyclization, ring contraction of five-membered rings, and ring expansion of threemembered rings.<sup>1</sup> However, a synthesis which can be represented as [N-C-C + C=O] has not been described.<sup>2</sup> We now report a new synthesis of  $\beta$ -lactams according to this scheme as an extension of lactam synthesis by the use of organometallic complexes.<sup>3</sup>

A typical procedure is as follows. 2-Bromo-3-(N-benzyl)aminopropene (1, 5·16 mmol) (which was prepared from 2,3-dibromopropene, benzylamine, and potassium carbonate) and Bu<sup>n</sup><sub>3</sub>N (6·45 mmol) in hexamethylphosphoramide (HMPA) were stirred at 100 °C under CO (1 atm) with a catalytic amount of Pd(OAc)<sub>2</sub> (0·1 mmol) and PPh<sub>3</sub> (0·4 mmol) for 5 h to afford the N-benzyl- $\alpha$ -methylene- $\beta$ lactam (2) [m.p. 32 °C (from ether), 66·9% yield] (Scheme). The spectral data [ $\nu_{C=0}$ (CHCl<sub>3</sub>) 1740 cm<sup>-1</sup>;  $\delta$  (CDCl<sub>3</sub>) 3·65 (2H, t, J 1 Hz, -CH<sub>2</sub>-N-CH<sub>2</sub>Ph), 4·54 (2H, s, -N-CH<sub>2</sub>-Ph), and 5·17 and 5·75 (both 1H, dd, J 1 and 3 Hz, vinyl H); m/e 173 ( $M^+$ ) and 133] and elemental analysis



SCHEME. i, PhCH<sub>2</sub>NH<sub>2</sub>, K<sub>2</sub>CO<sub>3</sub>; ii, CO, Pd catalyst, PPh<sub>3</sub>, Bu<sup>n</sup><sub>3</sub>N-HMPA.

supported this assignment. The use of 0.5 mol% of Pd(OAc)<sub>2</sub> as catalyst gives (2) in 53.3% yield. Pd(acac)<sub>2</sub> may be used instead of  $Pd(OAc)_2$  with similar results.

In confirmation of these results, (2) was reduced with sodium borohydride in EtOH-tetrahydrofuran (THF) to afford the N-benzyl- $\alpha$ -methyl- $\beta$ -lactam (3), [61.3% yield,  $v_{c=0}$  (neat) 1740 cm<sup>-1</sup>], which was also obtained by hydrogenation of (2) with  $PtO_2$  in THF-EtOH in 86.3% yield (equation 1). The Michael addition of piperidine to the



 $\beta$ -lactam (2) in Bu<sup>t</sup>OH proceeded slowly at room temperature to afford compound (4) [46.6% yield,  $v_{c=0}$  (neat)  $1740 \text{ cm}^{-1}$ ] (equation 2). On addition of benzylamine and after standing at room temperature for one week, followed by acetylation, (2) gave only a small amount of the desired product (5) (6.8% yield) and the cleaved compound (6) (12.6% yield) (equation 3).

This method has been further extended to syntheses of N-substituted  $\alpha$ -methylene- $\beta$ -lactams (12)-(14) from the corresponding 2-bromo-3-aminopropenes (7)-(9) in 61.9, 62.9, and 37.6% yields, respectively to confirm its broad applicability. Substrates (10) and (11) were synthesized from styrene in five steps.<sup>†</sup> The insertion of carbon monoxide into the Z-isomer (10) was successfully carried out as before to afford Z- $\alpha$ -benzylidene-N-benzyl- $\beta$ -lactam (15) [75.9%; m.p. 67-67.5 °C (from hexane-ether);  $v_{c=0}$  (Nujol) 1730 cm<sup>-1</sup>;  $\delta$  (CDCl<sub>3</sub>) 6.19 (1H, s, vinyl H); m/e 249( $M^+$ )]. The E-isomer (11) afforded the E- $\alpha$ benzylidene- $\beta$ -lactam (16) [89.5%; m.p. 140.5—141 °C (from ether);  $v_{C=0}$  (Nujol) 1730 cm<sup>-1</sup>;  $\delta$  (CDCl<sub>3</sub>) 6.99 (1H, s, vinyl H);  $m/e 249(M^+)$ ]. These results demonstrate that



an olefinic compound like styrene can be easily converted into a vinyl halide bearing an aminomethyl group at the halogeno position, which should be a useful precursor for the synthesis of substituted  $\alpha$ -methylene- $\beta$ -lactams by the present method.

(11)

The key intermediate of this reaction is assumed to be an acylpalladium complex (18), which may be generated from the vinylpalladium complex (17) co-ordinated to carbon





monoxide.  $\alpha$ -Methylene- $\beta$ -lactams should be important in the search for biologically active substances in this series,4 and an extension of this method is being investigated.

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† The details of the synthesis of the compounds (10) and (11) from styrene will be described in full papers.

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<sup>2</sup> A new synthetic method for  $\beta$ -lactams using Fe(CO)<sub>5</sub> has been reported in which the carbon monoxide of Fe(CO)<sub>5</sub> reacted with the internal amino group of the ammonium salt: P. K. Wong, M. Madhavarao, D. F. Marten, and M. Rosenblum, *J. Amer. Chem. Soc.*, 1977, 99, 2823. <sup>3</sup> M. Mori, K. Chiba, and Y. Ban, J. Org. Chem., 1978, 43, 1684.

<sup>4</sup> Syntheses of  $\alpha$ -methylene- $\beta$ -lactams by other methods have been reported recently: S. R. Fletcher and I. T. Kay, J.C.S. Chem. Comm., 1978, 903; T. Minami, M. Ishida, and T. Agawa, ibid., p. 12; E. J. Moriconi and J. F. Kelly, J. Amer. Chem. Soc., 1966, 88, 3657.