# Reaction of $\left[\mathrm{M}_{3}(\mathrm{CO})_{12}\right]$ ( $\mathrm{M}=\mathrm{Ru}$ or Os ) with the Chalconide Elements; $\boldsymbol{X}$-Ray Crystal and Molecular Structure of $\left[\mathrm{Os}_{4}(\mathrm{CO})_{12} \mathrm{H}_{2} \mathrm{Se}_{2}\right.$ ] 

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#### Abstract

Summary Reaction of $\left[\mathrm{Os}_{3}(\mathrm{CO})_{12}\right]$ with $\mathrm{X}_{n}(\mathrm{X}=\mathrm{S}$, Se , or Te ) in n-octane under reflux yields $\left[\mathrm{Os}_{3}(\mathrm{CO})_{9} \mathrm{H}_{2} \mathrm{X}\right]$, $\left[\mathrm{Os}_{3}(\mathrm{CO})_{9} \mathrm{X}_{2}\right]$, and $\left[\mathrm{Os}_{4}(\mathrm{CO})_{12} \mathrm{H}_{2} \mathrm{X}_{2}\right]$, but a $\mathrm{CO} / \mathrm{H}_{2}$ pressure of 35 atm is required to prepare the ruthenium analogues $\left[\mathrm{Ru}_{3}(\mathrm{CO})_{9} \mathrm{H}_{2} \mathrm{X}\right]$ and $\left[\mathrm{Ru}_{3}(\mathrm{CO})_{9} \mathrm{X}_{2}\right.$ ]; an $X$-ray analysis of $\left[\mathrm{Os}_{4}(\mathrm{CO})_{12} \mathrm{H}_{2} \mathrm{Se}_{2}\right.$ ] shows that it has a novel trigonal prismatic $\mathrm{Os}_{4} \mathrm{Se}_{2}$ core.


Although an extensive chemistry of sulphur, selenium, and tellurium derivatives of the iron carbonyls has been de-
veloped, ${ }^{1}$ relatively little is known of the chemistry of the ruthenium and osmium analogues. In this communication we report the synthesis of a series of compounds of general formulae $\left[\mathrm{M}_{3}(\mathrm{CO})_{9} \mathrm{H}_{2} \mathrm{X}\right]$ ( $\mathrm{M}=\mathrm{Ru}$ or Os; $\mathrm{X}=\mathrm{S}$, Se , or Te ), $\left[\mathrm{M}_{3}(\mathrm{CO})_{9} \mathrm{X}_{2}\right]$ ( $\mathrm{M}=\mathrm{Ru}$ or Os; $\mathrm{X}=\mathrm{S}, \mathrm{Se}$, or Te ), and $\left[\mathrm{Os}_{4}(\mathrm{CO})_{12} \mathrm{H}_{2} \mathrm{X}_{2}\right](\mathrm{X}=\mathrm{S}$ or Se$)$.

Dodecacarbonyltriosmium was treated with $\mathrm{X}_{n}(\mathrm{X}=\mathrm{S}$, Se , or Te ) in n-octane under reflux for 15 h . After separation on silica (t.l.c.) three products were identified on the basis of their analytical and spectroscopic data (Table) as

Table. Spectroscopic data for $\left[\mathrm{M}_{3}(\mathrm{CO})_{9} \mathrm{X}_{2}\right],\left[\mathrm{M}_{8}(\mathrm{CO})_{9} \mathrm{H}_{2} \mathrm{X}\right]$, and $\left[\mathrm{Os}_{4}(\mathrm{CO})_{12} \mathrm{H}_{2} \mathrm{X}_{2}\right]$.

| Compound | $\mathrm{v}_{\mathrm{co}} / \mathrm{cm}^{-1}$ | $\tau^{\text {a }}$ /p.p.m. | $m / e^{\text {b }}$ |
| :---: | :---: | :---: | :---: |
| $\left[\mathrm{Ru}_{3}(\mathrm{CO})_{9} \mathrm{~S}_{2}\right]$ | 2078, 2057, 2024, 2016 |  | 8 |
| $\left[\mathrm{Ru}_{3}(\mathrm{CO})_{8} \mathrm{Se}_{2}\right]$ | 2076, 2053, 2019, 2010 |  | 728 |
| $\left[\mathrm{Os}_{8}(\mathrm{CO})_{9} \mathrm{~S}_{2}{ }^{2}\right]$ | 2078s, 2058s, 2016s, 2010m |  | 892 |
| $\left[\mathrm{Os}_{3}(\mathrm{CO})_{9} \mathrm{Se}_{2}\right]$ | 2075s, 2054s, 2012s, 2008sh |  | 992 |
| $\left[\mathrm{Os}_{3}(\mathrm{CO}){ }_{9} \mathrm{Te}_{2}\right]$ | 2067s, 2046s, 2006s, 2002sh |  | 1088 |
| $\left[\mathrm{H}_{2} \mathrm{Ru}_{3}(\mathrm{CO})_{8} \mathrm{~S}\right]$ | 2118m, 2084s, 2060vs, 2046s, 2016s, 2008s, 1996m, 1991w | 28.93 | 598 |
| $\left[\mathrm{H}_{2} \mathrm{Ru}_{3}(\mathrm{CO})_{9} \mathrm{Se}\right]$ | $2115 \mathrm{~m}, 2082 \mathrm{~s}, 2058 \mathrm{~s}, 2045 \mathrm{~s}, 2015 \mathrm{~s}, 2007 \mathrm{~s}, 1995 \mathrm{~m}, 1989 \mathrm{w}$ | 30.00 | 648 |
| $\left[\mathrm{H}_{2} \mathrm{Ru}_{3}(\mathrm{CO})_{9} \mathrm{Te}\right]$ | $2111 \mathrm{~m}, 2078 \mathrm{~s}$, 2054vs, $2043 \mathrm{~s}, 2012 \mathrm{~s}, 2006 \mathrm{~s}, 1993 \mathrm{~m}, 1987 \mathrm{w}$ | - | 696 |
| $\left[\mathrm{H}_{2} \mathrm{OS}_{4}(\mathrm{CO})_{12} \mathrm{~S}_{2}\right]$ | $2115 \mathrm{w}, 2095 \mathrm{~s}, 2078 \mathrm{~s}, 2051 \mathrm{~s}, 2034 \mathrm{~m}, 2022 \mathrm{~s}, 2018 \mathrm{~m}, 2006 \mathrm{~m}$, 1993w, 1983w, 1978w | 24.25 | 1170 |
| $\left[\mathrm{H}_{2} \mathrm{Os}_{4}(\mathrm{CO})_{12} \mathrm{Se}_{2}\right]$ | $2111 \mathrm{w}, 2091 \mathrm{~s}, 2076 \mathrm{~s}, 2049 \mathrm{~s}, 2031 \mathrm{~m}, 2019 \mathrm{~s}, 2017 \mathrm{sh}, 2004 \mathrm{~m}$, 1990w, 1981w, 1975m | 24.92 | 1270 |

$\left[\mathrm{Os}_{3}(\mathrm{CO})_{9} \mathrm{H}_{2} \mathrm{X}\right],\left[\mathrm{Os}_{3}(\mathrm{CO})_{9} \mathrm{X}_{2}\right]$, and $\left[\mathrm{Os}_{4}(\mathrm{CO})_{12} \mathrm{H}_{2} \mathrm{X}_{2}\right]$. Under similar conditions no reaction of $\mathrm{X}_{2}$ with $\left[\mathrm{Ru}_{3}(\mathrm{CO})_{12}\right]$ was observed. Improved yields of the osmium derivatives were obtained from the reaction of $\left[\mathrm{Os}_{3}(\mathrm{CO})_{12}\right]$ with $\mathrm{X}_{2}$ under a pressure of $\mathrm{CO} / \mathrm{H}_{2}(35 \mathrm{~atm} ; 1: 1)$. Under similar conditions the two ruthenium analogues $\left[\mathrm{Ru}_{3}(\mathrm{CO})_{9} \mathrm{H}_{2} \mathrm{X}\right]$ and $\left[\mathrm{Ru}_{3}(\mathrm{CO})_{9} \mathrm{X}_{2}\right.$ ] were also prepared. Under CO pressure, but in the absence of $\mathrm{H}_{2}$, the yields of the hydrido-species $\left[\mathrm{M}_{3}(\mathrm{CO})_{8} \mathrm{H}_{2} \mathrm{X}\right]$ and $\left[\mathrm{Os}_{4}(\mathrm{CO})_{12} \mathrm{H}_{2} \mathrm{X}_{2}\right]$ decreased significantly. The molecular structure of $\left[\mathrm{Os}_{4}(\mathrm{CO})_{12} \mathrm{H}_{2} \mathrm{Se}_{2}\right]$ was established by $X$-ray diffraction. Yellow platelets were obtained by recrystallisation from hexane.

Crystal data: $\mathrm{C}_{12} \mathrm{H}_{2} \mathrm{O}_{12} \mathrm{Os}_{4} \mathrm{Se}_{2}, \quad M=1400 \cdot 86$, triclinic, $a=10 \cdot 004(3), b=12.764(5), c=9.711(3) \AA, \alpha=98 \cdot 14(2)$, $\beta=113.36(2), \gamma=100.73(2)^{\circ}, U=1086.1 \AA^{3}, Z=2, D_{\mathrm{c}}$ $=4.28 \mathrm{~g} \mathrm{~cm}^{-3} ; \mu\left(\mathrm{Mo}-K_{\alpha}\right)=267.0 \mathrm{~cm}^{-1}$, space group $P \overline{1}$. 5557 intensities $\left(2 \theta_{\max }=60 \cdot 0^{\circ}\right)$ were recorded on a Philips PW1100 four-circle diffractometer using graphite-monochromated Mo- $K_{\alpha}$ radiation. The data were corrected for absorption and Lorentz polarisation factors, and averaged to give 3286 unique observed intensities $[F>4 \sigma(F)]$. The Os and Se atoms were located by multisolution $\Sigma_{2}$ sign expansion, and the C and O atoms from a subsequent electron density difference synthesis. The structure was refined by blocked cascade least-squares (Os and Se anisotropic, C and O isotropic). A weighting scheme of the form $w=\left[\sigma^{2}(F)+0.001 F^{2}\right]^{-1}$ was employed. The final residuals were $R=0.035$ and $R=\left[\Sigma w^{\frac{1}{2}} \Delta / \Sigma w^{\frac{1}{2}}\left|F_{0}\right|\right]=$ $0 \cdot 034 . \dagger$

The molecular geometry is shown in the Figure which includes some important bond parameters. The four Os and two Se atoms define a distorted trigonal prism. Each Se atom caps a triangular arrangement of Os atoms; two Os-Os distances in each ' $\mathrm{Os}_{3} \mathrm{Se}^{\prime}$ ' unit are ca. $4 \AA$ and hence non-bonding. Three terminal carbonyl ligands are bonded to each Os atom. The two hydride ligands probably edge bridge the two long Os-Os bonds [Os(1)-Os(4) and Os(2)Os(3)] since the carbonyl ligands bend away from these edges. The cis Os-Os-C angles for the two bonds have an average value of $117 \cdot 4^{\circ}$ compared to $86 \cdot 3^{\circ}$ for the $\mathrm{Os}(1)-$ Os(2) bond.

If the two Se atoms act as 4 -electron donors the molecule is a 66 -electron system with 3 more electron pairs than


Figure. The molecular geometry of $\left[\mathrm{Os}_{4}(\mathrm{CO})_{12} \mathrm{H}_{2} \mathrm{Se}_{2}\right]$, including the atom numbering scheme. Bond lengths: $\mathrm{Os}(1)-\mathrm{Os}(2)$, $2 \cdot 884(1) ; \mathrm{Os}(1)-\mathrm{Os}(4), 2 \cdot 965(1) ; \mathrm{Os}(2)-\mathrm{Os}(3), 2 \cdot 966(1) ; \mathrm{Os}(1)-$ $\mathrm{Se}(1), 2.543(1) ; \mathrm{Os}(3)-\mathrm{Se}(1), 2.571(1)$; $\mathrm{Os}(4)-\mathrm{Se}(1), 2.537(1)$; $\mathrm{Os}(2)-\mathrm{Se}(2), 2 \cdot 538(1) ; \mathrm{Os}(3)-\mathrm{Se}(2), 2 \cdot 546(1)$; and $\mathrm{Os}(4)-\mathrm{Se}(2)$, $2 \cdot 576(1) \AA$; bond angles: $\mathrm{Os}(2)-\mathrm{Os}(1)-\mathrm{Os}(4), 86 \cdot 1(1)$ and $\mathrm{Os}(1)-$ $\mathrm{Os}(2)-\mathrm{Os}(3), 85 \cdot 7(1)^{\circ}$.
required for the 'electron precise' tetrahedral configuration (cf. $\left[\mathrm{Os}_{4}(\mathrm{CO})_{12} \mathrm{H}_{4}\right]$ ). The addition of 3 electron pairs causes the breaking of three tetrahedral edges to give the observed cluster geometry.

Trinuclear clusters of iron, ruthenium, and osmium with capping sulphur atoms are known, ${ }^{2}$ and selenium has been found to act as a $\mu_{3}$-ligand in the clusters $\left[\mathrm{FeCo}_{2}(\mathrm{CO})_{9} \mathrm{Se}\right]$ and $\left[\mathrm{Co}_{3}(\mathrm{CO})_{9} \mathrm{Se}\right]^{3}$ This is the first structural example of a tetranuclear osmium cluster with two capping chalconide ligands. The structure most closely resembles that of $\left[\mathrm{Fe}_{3}(\mathrm{CO})_{9}\left(\mathrm{Bu}^{\mathrm{t}} \mathrm{NS}\right) \mathrm{S}^{4}\right.$ in which one $\mathrm{M}(\mathrm{CO})_{3}$ unit in the Os cluster has been replaced by the ' $B u^{t} N$ ' fragment of the $\eta^{4}$-( $\left.\mathrm{Bu}^{\mathrm{t}} \mathrm{NS}\right)$ ligand, and the hydrides removed. The structures of the $\left[\mathrm{M}(\mathrm{CO})_{9} \mathrm{X}_{2}\right]$ compounds are analogous to that of $\left[\mathrm{Fe}_{3}(\mathrm{CO})_{9} \mathrm{~S}_{2}\right]$, and those in the $\left[\mathrm{M}_{3}(\mathrm{CO})_{9} \mathrm{H}_{2} \mathrm{X}\right]$ series to $\left[\mathrm{Os}_{3}(\mathrm{CO})_{9} \mathrm{H}_{2} \mathrm{~S}\right] .{ }^{2}$

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