Reaction of $[M_3(CO)_{12}]$ (M = Ru or Os) with the Chalconide Elements; X-Ray Crystal and Molecular Structure of $[Os_4(CO)_{12}H_2Se_2]$

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Summary Reaction of $[Os_3(CO)_{12}]$ with X_n (X = S, Se, or Te) in n-octane under reflux yields $[Os_3(CO)_9H_2X]$, $[Os_3(CO)_9X_2]$, and $[Os_4(CO)_{12}H_2X_2]$, but a CO/H₂ pressure of 35 atm is required to prepare the ruthenium analogues $[Ru_3(CO)_9H_2X]$ and $[Ru_3(CO)_9X_2]$; an X-ray analysis of $[Os_4(CO)_{12}H_2Se_2]$ shows that it has a novel trigonal prismatic Os_4Se_2 core.

ALTHOUGH an extensive chemistry of sulphur, selenium, and tellurium derivatives of the iron carbonyls has been de-

veloped,¹ relatively little is known of the chemistry of the ruthenium and osmium analogues. In this communication we report the synthesis of a series of compounds of general formulae $[M_3(CO)_9H_2X]$ (M = Ru or Os; X = S, Se, or Te), $[M_3(CO)_9X_2]$ (M = Ru or Os; X = S, Se, or Te), and $[Os_4(CO)_{12}H_2X_2]$ (X = S or Se).

Dodecacarbonyltriosmium was treated with X_n (X = S, Se, or Te) in n-octane under reflux for 15 h. After separation on silica (t.l.c.) three products were identified on the basis of their analytical and spectroscopic data (Table) as

TABLE.	Spectroscopic dat	a for [M ₃ (CO) ₉	X_2], $[M_8(CO)_9H_2X]$,	and $[Os_4(CO)_{12}H_2X_2]$.
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IABLE.	Spectroscopic data for $[M_3(CO)_9A_2]$, $[M_3(CO)_9H_2A]$, and [$OS_4(CO)_{12}H_2X_2$].
Compound	$v_{\rm CO}/{\rm cm}^{-1}$	$\tau^{a}/p.p.m.$	m/e Þ
$[\mathrm{Ru}_{3}(\mathrm{CO})_{9}\mathrm{S}_{2}]$	2078, 2057, 2024, 2016		628
$[\mathrm{Ru}_{3}(\mathrm{CO})_{9}\mathrm{Se}_{2}]$	2076, 2053, 2019, 2010		728
$[Os_8(CO)_9S_2]$	2078s, 2058s, 2016s, 2010m		892
$[Os_8(CO)_9Se_2]$	2075s, 2054s, 2012s, 2008sh		992
$[Os_3(CO)_9Te_2]$	2067s, 2046s, 2006s, 2002sh		1088
$[H_2Ru_3(CO)_9S]$	2118m, 2084s, 2060vs, 2046s, 2016s, 2008s, 1996m, 1991w	28.93	598
$[H_2Ru_3(CO)_9Se]$	2115m, 2082s, 2058s, 2045s, 2015s, 2007s, 1995m, 1989w	30.00	648
$[H_2Ru_3(CO)_9Te]$	2111m, 2078s, 2054vs, 2043s, 2012s, 2006s, 1993m, 1987w		696
$[\mathrm{H_2Os}_4(\mathrm{CO})_{12}\mathrm{S}_2]$	2115w, 2095s, 2078s, 2051s, 2034m, 2022s, 2018m, 2006m, 1993w, 1983w, 1978w	24.25	1170
$[H_2Os_4(CO)_{12}Se_2]$	2111w, 2091s, 2076s, 2049s, 2031m, 2019s, 2017sh, 2004m, 1990w, 1981w, 1975m	24.92	1270

⁸ Relative to Me₄Si τ 10.00. ^b For ¹⁹²Os, ¹⁰⁴Ru, ³²S, ⁸²Se, and ¹³⁰Te.

 $[Os_3(CO)_9H_2X]$, $[Os_3(CO)_9X_2]$, and $[Os_4(CO)_{12}H_2X_2]$. Under similar conditions no reaction of X₂ with [Ru₃(CO)₁₂] was observed. Improved yields of the osmium derivatives were obtained from the reaction of [Os₃(CO)₁₂] with X₂ under a pressure of CO/H_2 (35 atm; 1:1). Under similar conditions the two ruthenium analogues $[Ru_3(CO)_9H_2X]$ and $[Ru_3(CO)_9X_2]$ were also prepared. Under CO pressure, but in the absence of H₂, the yields of the hydrido-species $[M_3(CO)_9H_2X]$ and $[Os_4(CO)_{12}H_2X_2]$ decreased significantly. The molecular structure of [Os₄(CO)₁₂H₂Se₂] was established by X-ray diffraction. Yellow platelets were obtained by recrystallisation from hexane.

Crystal data: $C_{12}H_2O_{12}Os_4Se_2$, M = 1400.86, triclinic, a = 10.004(3), b = 12.764(5), c = 9.711(3) Å, $\alpha = 98.14(2), c = 9.711(3)$ $\beta = 113.36(2), \ \gamma = 100.73(2)^{\circ}, \ U = 1086.1 \text{ Å}^3, \ Z = 2, \ D_c$ = 4.28 g cm^{-3} ; $\mu(\text{Mo-}K_{\alpha}) = 267.0 \text{ cm}^{-1}$, space group $P\overline{1}$. 5557 intensities $(2\theta_{\text{max}} = 60.0^{\circ})$ were recorded on a Philips PW1100 four-circle diffractometer using graphite-monochromated Mo- K_{α} radiation. The data were corrected for absorption and Lorentz polarisation factors, and averaged to give 3286 unique observed intensities $[F > 4\sigma(F)]$. The Os and Se atoms were located by multisolution Σ_2 sign expansion, and the C and O atoms from a subsequent electron density difference synthesis. The structure was refined by blocked cascade least-squares (Os and Se anisotropic, C and O isotropic). A weighting scheme of the form $w = [\sigma^2(F) + 0.001 F^2]^{-1}$ was employed. The final residuals were R = 0.035 and $R = \left[\sum w^{\frac{1}{2}} \Delta / \sum w^{\frac{1}{2}} \right] F_0$ 0.034.†

The molecular geometry is shown in the Figure which includes some important bond parameters. The four Os and two Se atoms define a distorted trigonal prism. Each Se atom caps a triangular arrangement of Os atoms; two Os-Os distances in each 'Os₃Se' unit are ca. 4 Å and hence non-bonding. Three terminal carbonyl ligands are bonded to each Os atom. The two hydride ligands probably edge bridge the two long Os–Os bonds [Os(1)–Os(4)] and Os(2)–Os(4)Os(3)] since the carbonyl ligands bend away from these edges. The cis Os-Os-C angles for the two bonds have an average value of 117.4° compared to 86.3° for the Os(1)-Os(2) bond.

If the two Se atoms act as 4-electron donors the molecule is a 66-electron system with 3 more electron pairs than

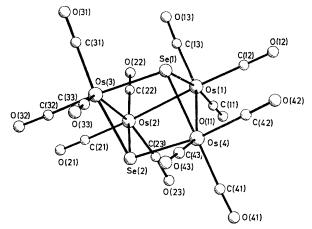


FIGURE. The molecular geometry of $[Os_4(CO)_{12}H_2Se_2]$, including the atom numbering scheme. Bond lengths: Os(1)-Os(2), 2·884(1); Os(1)-Os(4), 2·965(1); Os(2)-Os(3), 2·966(1); Os(1)-Os(4), 2·965(1); Os(2)-Os(3), 2·966(1); Os(1)-Se(1), 2·543(1); Os(3)-Se(1), 2·571(1); Os(4)-Se(1), 2·537(1); Os(3)-Se(2), 2·546(1); and Os(4)-Se(2), 2·546(1); Os(3)-Se(2), 2·546(1); Os(3)-Se(2), 2·546(1); Os(3)-Se(2), 2·546(1); Os(3)-Se(2), 2·546(1); Se(3)-Se(2), 2·546(1); Se(3)-Se(2)-Se(2)-Se(2), 2·546(1); Se(3)-Se(2)-2.576(1) Å; bond angles: Os(2) - Os(1) - Os(4), 86.1(1) and Os(1) - Os(4). $Os(2) - Os(3), 85 \cdot 7(1)^{\circ}$.

required for the 'electron precise' tetrahedral configuration (cf. $[Os_4(CO)_{12}H_4]$). The addition of 3 electron pairs causes the breaking of three tetrahedral edges to give the observed cluster geometry.

Trinuclear clusters of iron, ruthenium, and osmium with capping sulphur atoms are known,² and selenium has been found to act as a μ_3 -ligand in the clusters [FeCo₂(CO)₉Se] and $[Co_3(CO)_9Se]$.³ This is the first structural example of a tetranuclear osmium cluster with two capping chalconide ligands. The structure most closely resembles that of $[Fe_3(CO)_9(Bu^tNS)S]^4$ in which one $M(CO)_3$ unit in the Os cluster has been replaced by the 'Bu^tN' fragment of the η^{4} -(Bu^tNS) ligand, and the hydrides removed. The structures of the $[M(CO)_9X_2]$ compounds are analogous to that of $[Fe_3(CO)_9S_2]$, and those in the $[M_3(CO)_9H_2X]$ series to $[Os_3(CO)_9H_2S]^2$.

We thank the S.R.C. for financial support and Johnson Matthey and Co. Ltd. for the loan of OsO_4 .

(Received, 23rd May 1979; Com. 548.)

† The atomic co-ordinates for this work are available on request from Prof. Dr. G. Bergerhoff, Institut für Anorganische Chemie, Universität, Gerhard-Domagk-str. 1, D-5300 Bonn 1, West Germany. Any request should be accompanied by the full literature citation for this communication.

¹ W. Hieber and J. Gruber, Z. anorg. Chem., 1958, 296, 91.
² C. H. Wei and L. F. Dahl, Inorg. Chem., 1965, 4, 493; B. F. G. Johnson, J. Lewis, D. Pippard, P. R. Raithby, G. M. Sheldrick, and K. D. Rouse, J.C.S. Dalton, 1979, 616.
³ C. E. Strouse and L. F. Dahl, J. Amer. Chem. Soc., 1971, 93, 6032.
⁴ R. Meij, J. van der Helm, D. J. Stufkens, and K. Vrieze, J.C.S. Chem. Comm., 1978, 506.

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