# Di- $\mu$-(Cyclopentadienyl)-bis(tri-isopropylphosphine)dipalladium(I) 

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Summary Di- $\mu$-(cyclopentadienyl)-bis(tri-isopropylphosphine)dipalladium( I ) was prepared by the reaction of $\left[\left(\mathrm{C}_{5} \mathrm{H}_{5}\right) \mathrm{Pd}\left(\mathrm{MeCO}_{2}\right)\left(\mathrm{PPr}_{3}{ }_{3}\right)\right]$ with $\mathrm{NaK}_{2 \cdot 8}$; its sandwichtype structure is proved by reactions with $\mathrm{Me}_{3} \mathrm{SiCl}$, PhSH, or $\mathrm{PhCO}_{2} \mathrm{H}$ which give $\left[\left(\mu-\mathrm{C}_{5} \mathrm{H}_{5}\right)(\mu-\mathrm{X}) \mathrm{Pd}_{2}\left(\mathrm{PPr}_{3}\right)_{2}\right]$ ( $\mathrm{X}=\mathrm{Cl}, \mathrm{PhS}$, or $\mathrm{PhCO}_{2}$ ).

Binuclear sandwich-type complexes possessing an $\mathrm{M}_{2}$ unit co-ordinated by two $\mathrm{C}_{n} \mathrm{H}_{n}$ ring systems are very rare. Recently, we discovered a simple and straightforward route to obtain ( $\mathrm{Pd}-\mathrm{Pd}$ )-complexes of general composition $(\mu-\mathrm{X})(\mu-\mathrm{Y}) \mathrm{Pd}_{2} \mathrm{~L}_{2}$ in which the oxidation state of palladium is $+I$ [equation (l)]. ${ }^{1}$ The $1+1$-addition can be applied to

prepare the corresponding complexes with $\mathrm{X}=\mathrm{C}_{5} \mathrm{H}_{5}$, $\mathrm{Y}=2-\mathrm{RC}_{3} \mathrm{H}_{4} ; \mathrm{X}=\mathrm{Y}=2-\mathrm{RC}_{3} \mathrm{H}_{4} ; \mathrm{X}=2-\mathrm{RC}_{3} \mathrm{H}_{4}, \mathrm{Y}=\mathrm{Cl}$, $\mathrm{I} ; 1,2 \mathrm{X}=2-\mathrm{RC}_{3} \mathrm{H}_{4}, \mathrm{Y}=\mathrm{MeCO}_{2}, \mathrm{CF}_{3} \mathrm{CO}_{2}$, or $\mathrm{PhCO}_{2},{ }^{3}$ and $\mathrm{L}=\mathrm{PPr}^{\mathrm{i}}{ }_{3}, \mathrm{PBu}_{3}^{\mathrm{t}}$, or $\mathrm{P}\left(\mathrm{C}_{6} \mathrm{H}_{11}\right)_{3}$. There is no possibility, however, of obtaining the binuclear sandwich $\left[\left(\mu-\mathrm{C}_{5} \mathrm{H}_{5}\right)_{2} \mathrm{Pd}_{2} \mathrm{~L}_{2}\right]$ according to equation (l) since the necessary starting compound $\operatorname{Pd}\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)_{2}$ has as yet not been prepared. ${ }^{4}$


(1)
(2)
(3)

Scheme 1. Reagents: i, $\mathrm{TlC}_{5} \mathrm{H}_{5}, \mathrm{THF}-\mathrm{C}_{6} \mathrm{H}_{8}, 25^{\circ} \mathrm{C}, 3 \mathrm{~h}$; ii, $\mathrm{NaK}_{2 \cdot 8}, \mathrm{C}_{6} \mathrm{H}_{6}, 25^{\circ} \mathrm{C}, 5 \mathrm{~h}\left(\mathrm{~L}=\mathrm{PPr}_{3}{ }_{3}\right)$.

Therefore, we have used a different route, which is outlined in Scheme 1. The acetate-bridged complex (1) ${ }^{5}$ reacts with $\mathrm{TIC}_{5} \mathrm{H}_{5}$ to give the new mononuclear compound (2) $\left[{ }^{1} \mathrm{H}\right.$ n.m.r. $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right): \delta 5 \cdot 67\left(\mathrm{~d}, 5 \mathrm{H}, J_{\mathrm{PH}} 2.0 \mathrm{~Hz}\right)$ and 2.00 ( $\mathrm{s}, 3 \mathrm{H}$ )] which, on treatment with sodium-potassium alloy $\mathrm{NaK}_{2} \cdot 8,{ }^{6}$ forms (3) in $55 \%$ yield. Compound (3) is also obtained by reaction of $\left[\left(\mu-\mathrm{C}_{5} \mathrm{H}_{5}\right)\left(\mu-\mathrm{MeCO}_{2}\right) \mathrm{Pd}_{2}\left(\mathrm{PPr}_{3}\right)_{2}\right]^{3}$
with excess of $\mathrm{TlC}_{5} \mathrm{H}_{5}$ in benzene. The structure of (3) $[m / e$ (field desorption) 663.5 ; calc. 663.5] is confirmed by its ${ }^{1} \mathrm{H}$ n.m.r. spectrum which shows a $1: 2: 1$ triplet $\left[\delta\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)\right.$ $5 \cdot 70, J_{\mathrm{PH}} 2 \cdot 2 \mathrm{~Hz}$ ] for the cyclopentadienyl protons and excludes the formation of the isomer $\left[\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)(\mathrm{L}) \mathrm{Pd}-\mathrm{Pd}(\mathrm{L})-\right.$ $\left.\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)\right] . \dagger$

(5)

Scheme 2. Reagents: i, $\mathrm{Me}_{3} \mathrm{SiCl}\left(\mathrm{l}: 1\right.$ ), $\mathrm{C}_{6} \mathrm{H}_{6}, 25^{\circ} \mathrm{C}$; ii, PhSH (1:1), $\mathrm{C}_{6} \mathrm{H}_{8}, 25^{\circ} \mathrm{C}, 30 \mathrm{~min}$; iii, $\mathrm{PhCO}_{2} \mathrm{H}^{(1: 1)}, \mathrm{C}_{8} \mathrm{H}_{6}, 25{ }^{\circ} \mathrm{C}$ ( $\mathrm{L}=\operatorname{PPr}_{3}{ }^{1}$ ).

There is also strong support for the sandwich-type structure of (3) by its reactions with $\mathrm{Me}_{3} \mathrm{SiCl}, \mathrm{PhSH}$, or $\mathrm{PhCO}_{2} \mathrm{H}$ (Scheme 2). Compound (4) had originally been prepared by reduction of $\left(\mathrm{C}_{5} \mathrm{H}_{5}\right) \mathrm{Pd}(\mathrm{L}) \mathrm{Cl}$ with Mg or LiAlH$\left(\mathrm{Bu}^{\mathrm{t}} \mathrm{O}\right)_{3}$ in THF whereas (5) had been obtained by a metathetical reaction of $\left[\left(\mu-\mathrm{C}_{5} \mathrm{H}_{5}\right)(\mu-\mathrm{Br}) \mathrm{Pd}_{2} \mathrm{~L}_{2}\right]$ and TlSPh in benzene. ${ }^{7}$ The hitherto unknown complex (6) [ ${ }^{1} \mathrm{H}$ n.m.r. $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right): \delta 5 \cdot 15\left(\mathrm{t}, J_{\mathrm{PH}} 2.5 \mathrm{~Hz}, 5 \mathrm{H}\right), \mathrm{l} \cdot 22(\mathrm{~d} \times \mathrm{vt} 36 \mathrm{H})$, ${ }_{\ddagger}^{+}$ $8 \cdot 26\left(\mathrm{~m}, 2 \mathrm{H}\right.$ of $\left.\mathrm{PhCO}_{2}\right)$, and $7 \cdot 14\left(\mathrm{~m}, 3 \mathrm{H}\right.$ of $\left.\left.\mathrm{PhCO}_{2}\right)\right]$ is an analogue of the above mentioned compound $\left[\left(\mu-\mathrm{C}_{5} \mathrm{H}_{5}\right)(\mu-\right.$ $\left.\left.\mathrm{MeCO}_{2}\right) \mathrm{Pd}_{2} \mathrm{~L}_{2}\right]$ which is formed by the reaction of $\left[\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)\right.$ $\left.\mathrm{Pd}\left(\mathrm{MeCO}_{2}\right)\left(\mathrm{C}_{6} \mathrm{H}_{8}\right)\right]$ and $\mathrm{PdL}_{2} .{ }^{3}$

In contrast to $\left[\left(\mu-\mathrm{C}_{5} \mathrm{H}_{5}\right)\left(\mu-2-\mathrm{Bu}^{\mathrm{t}} \mathrm{C}_{3} \mathrm{H}_{4}\right) \mathrm{Pd}_{2} \mathrm{~L}_{2}\right],{ }^{2}$ complex (3) does not dissociate in solution to give $\operatorname{Pd}\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)_{2}$ and $\mathrm{PdL}_{2}$ indicating that co-ordination of two bridging cyclopentadienyl ligands to the $\mathrm{Pd}_{2} \mathrm{~L}_{2}$ unit stabilises the binuclear sandwich-type structure.

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$\dagger$ See, e.g., the complex $\left[\left(\mathrm{C}_{5} \mathrm{H}_{5}\right)(\mathrm{L}) \operatorname{Pd}-\mathrm{Pd}(\mathrm{L})\left(2-\mathrm{ClC}_{3} \mathrm{H}_{4}\right)\right]$ which shows a doublet for the cyclopentadienyl protons (ref. 2).
$\ddagger$ Doublet of virtual triplets, $N={ }^{3} J_{\mathrm{PH}}+{ }^{6} J_{\mathrm{PH}}=c a .13 \cdot 4,{ }^{3} J^{\prime}{ }_{\mathbf{H H}}=6.7 \mathrm{~Hz}$.
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