Di-μ-(Cyclopentadienyl)-bis(tri-isopropylphosphine)dipalladium(1)

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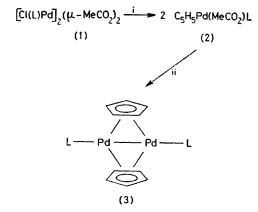
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Summary Di- μ -(cyclopentadienyl)-bis(tri-isopropylphosphine)dipalladium(I) was prepared by the reaction of $[(C_5H_5)Pd(MeCO_2)(PPr^l_3)]$ with NaK_{2·8}; its sandwichtype structure is proved by reactions with Me₃SiCl, PhSH, or PhCO₂H which give $[(\mu$ -C₅H₅)(μ -X)Pd₂(PPr l_3)₂] (X = Cl, PhS, or PhCO₂).

BINUCLEAR sandwich-type complexes possessing an M_2 unit co-ordinated by two C_nH_n ring systems are very rare. Recently, we discovered a simple and straightforward route to obtain (Pd-Pd)-complexes of general composition $(\mu$ -X)(μ -Y)Pd₂L₂ in which the oxidation state of palladium is +1 [equation (1)].¹ The 1+1-addition can be applied to

$$(X)Pd(Y) + PdL_2 \longrightarrow L \longrightarrow Pd \longrightarrow Pd \longrightarrow L$$
 (1)

prepare the corresponding complexes with $X=C_5H_5$, $Y=2\text{-RC}_3H_4$; $X=Y=2\text{-RC}_3H_4$; X=Cl, $I;^{1,2}$ $X=2\text{-RC}_3H_4$, $Y=MeCO_2$, CF_3CO_2 , or $PhCO_2,^3$ and $L=PPr^l_3$, PBu^l_3 , or $P(C_6H_{11})_3$. There is no possibility, however, of obtaining the binuclear sandwich $[(\mu\text{-}C_5H_5)_2Pd_2L_2]$ according to equation (1) since the necessary starting compound $Pd(C_5H_5)_2$ has as yet not been prepared.⁴



Scheme 1. Reagents: i, TIC_5H_5 , $THF-C_6H_6$, 25 °C, 3 h; ii, $NaK_{2\cdot 8}$, C_6H_6 , 25 °C, 5 h (L = PPr^i_3).

Therefore, we have used a different route, which is outlined in Scheme 1. The acetate-bridged complex (1)⁵ reacts with TlC_5H_5 to give the new mononuclear compound (2) [1H n.m.r. (C_6D_6): δ 5·67 (d, 5H, J_{PH} 2·0 Hz) and 2·00 (s, 3H)] which, on treatment with sodium-potassium alloy NaK_{2·8},⁶ forms (3) in 55% yield. Compound (3) is also obtained by reaction of $[(\mu-C_5H_5)(\mu-MeCO_2)Pd_2(PPr^1_3)_2]^3$

with excess of TlC_5H_5 in benzene. The structure of (3) $\lceil m/e \rceil$ (field desorption) 663.5; calc. 663.5] is confirmed by its ¹H n.m.r. spectrum which shows a 1:2:1 triplet δ (C₆D₆) 5.70, $J_{\rm PH}$ 2.2 Hz] for the cyclopentadienyl protons and excludes the formation of the isomer $[(C_5H_5)(L)Pd-Pd(L) (C_5H_5)].\dagger$

Scheme 2. Reagents: i, Me₂SiCl (1:1), C₆H₆, 25 °C; ii, PhSH (1:1), C₆H₆, 25 °C, 30 min; iii, PhCO₂H (1:1), C₆H₆, 25 °C (L = PPrl₃).

There is also strong support for the sandwich-type structure of (3) by its reactions with Me₃SiCl, PhSH, or PhCO₂H (Scheme 2). Compound (4) had originally been prepared by reduction of (C₅H₅)Pd(L)Cl with Mg or LiAlH-(ButO)₃ in THF whereas (5) had been obtained by a metathetical reaction of $[(\mu-C_5H_5)(\mu-Br)Pd_2L_2]$ and TISPh in benzene.⁷ The hitherto unknown complex (6) [¹H n.m.r. (C_6D_6) : δ 5·15 (t, J_{PH} 2·5 Hz, 5H), 1·22 (d \times vt 36H), 8.26 (m, 2H of PhCO2), and 7.14 (m, 3H of PhCO2)] is an analogue of the above mentioned compound $[(\mu-C_5H_5)(\mu-C_5H_5)]$ MeCO₂)Pd₂L₂] which is formed by the reaction of [(C₅H₅)- $Pd(MeCO_2)(C_6H_8)$] and PdL_2 .³

In contrast to $[(\mu-C_5H_5)(\mu-2-Bu^tC_3H_4)Pd_2L_2]$, complex (3) does not dissociate in solution to give $Pd(C_5H_5)_2$ and PdL₂ indicating that co-ordination of two bridging cyclopentadienyl ligands to the Pd₂L₂ unit stabilises the binuclear sandwich-type structure.

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 $\uparrow \text{ See, } \textit{e.g.}, \text{ the complex } [(C_5H_5)(L)\mathrm{Pd-Pd}(L)(2\text{-}ClC_3H_4)] \text{ which shows a doublet for the cyclopentadienyl protons (ref. 2)}.$

‡ Doublet of virtual triplets, $N = {}^3J_{\rm PH} + {}^6J_{\rm PH} = ca. 13.4$, ${}^3J'_{\rm HH} = 6.7$ Hz.

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