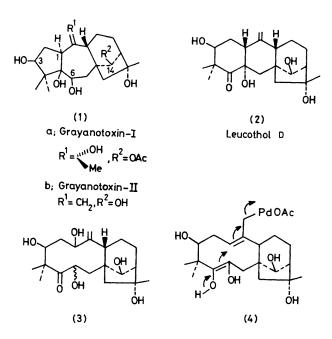
One-step Synthesis of the Diterpenoid Leucothol D from Grayanotoxin-II

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Summary Treatment of grayanotoxin-II (1b) with palladium acetate in methanol at room temperature gave leucothol D (2).

A NUMBER of toxic diterpenoids have been isolated from Leucothoe grayana Max. and classified into three structural groups: grayanotoxins (1),¹ leucothols (2),² and grayanols (3),³ but they have not yet been chemically inter-related. Kakisawa and his co-workers⁴ reported that the five-seven membered A-B ring system of grayanotoxin-I (1a), changes into a six-six membered ring system on treatment with toluene-*p*-sulphonyl chloride in pyridine. In this case,



however, the C-1 proton has an α orientation. We recently became interested in the reaction of grayanotoxins with organometallic compounds. Treatment of grayanotoxin-II (1b) with palladium acetate in methanol at room temperature gave a product, $C_{20}H_{30}O_5$, m.p. 272-274 °C. The ¹³C n.m.r. spectrum indicated the presence of the following groups; three methyls, five methylenes, three methines, two quaternary carbons, two secondary carbinyl carbons, two tertiary carbinyl carbons, a vinylidene, and a ketone. Other spectral data (¹H n.m.r. and c.d.) were very similar to those reported for leucothol D.^{2b} Comparison showed that the product was identical with an authentic specimen isolated from the plant. Hence we have achieved a one-step synthesis of leucothol D from grayanotoxin-II.

We regard the palladium compound (4) as a likely intermediate in the observed conversion of grayanotoxin-II into leucothol D.

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¹ J. Sakakibara, T. Shirai, K. Kaiya, and H. Nakata, *Phytochemistry*, 1978, 17, 1672 and references therein.

- ² (a) A. Furusaki, N. Hamanaka, H. Miyakoshi, T. Okuno, and T. Matsumoto, *Chem. Letters*, 1972, 783; N. Hamanaka, H. Miyakoshi, A. Furusaki, and T. Matsumoto, *ibid.*, p. 787. (b) H. Hikino, S. Koriyama, and T. Takemoto, *Tetrahedron*, 1973, 29, 773.
 ³ S. Fushiya, H. Hikino, and T. Takemoto, *Tetrahedron Letters*, 1974, 183.
 ⁴ H. Kakisawa, T. Kojima, M. Yanai, and K. Nakanishi, *Tetrahedron*, 1965, 21, 3091.