Deprotonation of a Trimethylsilylmethaneboronic Ester

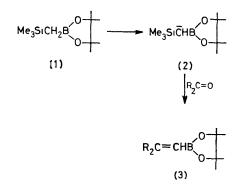
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Summary Deprotonation of pinacol trimethylsilylmethaneboronate yields an anion which reacts with carbonyl compounds to form alkeneboronic esters, or with alkyl halides to form α -trimethylsilylalkaneboronic esters

DEPROTONATION of gem-diboronic esters¹ or α -phenylthiomethaneboronic esters² yields carbanions useful in synthesis We have now extended this chemistry to the readily accessible pinacol trimethylsilylmethaneboronate (1)

Trimethylsilylmethylmagnesium chloride,³ added concurrently with trimethyl borate to diethyl ether stirred at -78 °C followed by aqueous work-up led to trimethylsilylmethaneboronic acid (65–75% yield), which was esterified with pinacol² to yield (1), b p 70–72 °C (12 Torr) Treatment of (1) with lithium 2,2 6,6-tetramethylpiperidide and tetramethylethylenediamine in tetrahydrofuran at 0 °C² yielded a solution of the anion (2), which was alkylated



efficiently by alkyl halides (see Γ able) Carbonyl compounds condensed with (2) with exclusive elimination of the silicon component to form alkene-boronic esters (3) (see Table) Products were isolated by distillation and characterised by ¹H n m r spectra and elemental analyses

TABLE

Reactions of pinacol trimethylsilylmethaneboronate anion (2)

Reactant	Product	% Yield
n-C ₅ H ₁₁ Br	n-C ₅ H ₁₁ -CH(SiMe ₃)BO ₂ C ₂ Me ₄	85
PhČH,Br	PhCH ₂ -CH(SiMe ₃)BO ₂ C ₂ Me ₄	83
$PhCH_{2}CH_{2}I$	PhCH ₂ CH ₂ -CH(SiMe ₃)BO ₂ C ₂ Me ₄	79
n-C ₆ H ₁₃ CHO	$n-C_6H_{13}CH=CHBO_2C_2Me_4$ (mostly <i>cis</i>)	73
PhCHO	$PhCH=CHBO_2C_2Me_4$ (mostly <i>cis</i>)	84
CH ₂ [CH ₂] ₄ - C=O	$CH_2[CH_2]_4C=CHBO_2C_2Me_4$	87
Bu ⁿ ₂ C=O	$\mathrm{Bu}^{n}_{2}\mathrm{C}=\mathrm{CHBO}_{2}\mathrm{C}_{2}\mathrm{Me}_{4}$	74

The exclusive elimination of silicon was unexpected in view of the ease of elimination of boron in other systems.⁴ However, ¹H n.m.r. spectra of the distilled alkeneboronic esters revealed no methylsilyl peak at δ 0, implying complete loss as hexamethyldisiloxane.

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 ⁵ D. S. Matteson and P. K. Jesthi, J. Organometallic Chem., 1976, 110, 25.

A further unexpected result was the predominance of the cis isomer in the alkeneboronic esters derived from aldehydes. The α -proton of cis pinacol 2-phenyletheneboronate, PhCH=CHBO₂C₂Me₄, appeared in the n.m.r. spectrum at δ 5.53, J 16 Hz and of the *trans* isomer at δ 6.12, J 18 Hz,⁵ in a ca. 2:1 ratio. Similarly, for pinacol oct-1-eneboronate, the cis α -proton is at δ 5.34, J 14 Hz and the trans α -proton at δ 5.44, ⁵ J 18 Hz, ratio 2:1. This stereochemistry contrasts with that of the condensation of diborylmethide ions with aldehyde, which yields >90%trans alkeneboronic esters.5

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