

## Longikaurin A and B; New, Biologically Active Diterpenoids from *Rabdosia longituba*

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**Summary** Chemical investigation of the biologically active substances of *Rabdosia longituba* led to the isolation of two new diterpenoids having an *ent*-kaurene skeleton, longikaurin A (**1**) and B (**2**), and their structures were established by spectroscopic and chemical data

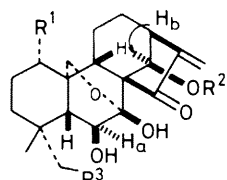
DURING studies on the biologically active diterpenoids of plants belonging to the genus *Rabdosia* [= *Isodon*]<sup>1</sup> (Labiatae), we examined the constituents of the leaves of *R. longituba* (Miquel) Hara [*I. longitubus* (Miq.) Kudo] and isolated two new diterpenoids, longikaurin A (**1**) and B (**2**), together with known kamebakaurin (**6**)<sup>2</sup> Longikaurin A (**1**) shows cytotoxicity‡ *in vitro* against cultured rat

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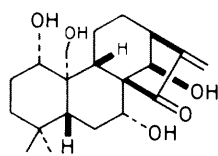
‡ 1  $\mu\text{g ml}^{-1}$  of longikaurin A (**1**) shows inhibitory activity (74%) on the growth of cultured rat mammary cancer FM 3A/B cells. Although longikaurin B (**2**) is likely to show a similar activity, poor solubility in  $\text{Me}_2\text{SO}-\text{H}_2\text{O}$  inhibited testing. The detailed study will be published elsewhere.

mammary cancer FM 3A/B cells, and both longikaurin A (1) and B (2) show antibacterial activity.†

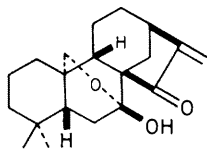
Longikaurin A (1), C<sub>20</sub>H<sub>28</sub>O<sub>5</sub>, m.p. 223–225 °C, [ $\alpha$ ]<sub>D</sub><sup>25</sup> –91.1° (*c* 0.21, C<sub>5</sub>H<sub>5</sub>N), has been shown to contain a five membered ketone conjugated with an *exo*-methylene group from the following spectral data:  $\lambda_{\max}$  (MeOH) 235 nm ( $\epsilon$  9530);  $\nu_{\max}$  (Nujol) 1705 and 1640 cm<sup>-1</sup>; <sup>1</sup>H n.m.r.  $\delta$  5.52 and 6.14 (each 1H, br s); <sup>13</sup>C n.m.r.  $\delta$  117.8 (t), 151.2 (s) [ $>C=CH_2$ ], and 207.1 p.p.m. [ketone]. Compound (1) shows a strong absorption band in the i.r. spectrum (3400–3050 cm<sup>-1</sup>) due to hydroxy groups and <sup>1</sup>H n.m.r. signals at  $\delta$  4.90 (1H, m), 6.10 (1H, m), and 6.59 (1H, d, *J* 10 Hz) which are assigned to three hydroxy groups. The <sup>1</sup>H [ $\delta$  3.77 (1H, d, *J* 7 Hz after D<sub>2</sub>O treatment, H<sub>a</sub>) and 4.76 (1H, d, *J* 2 Hz, H<sub>b</sub>)] and <sup>13</sup>C n.m.r. signals [ $\delta$  71.7 and 72.4 p.p.m. (each doublets)] suggest that two of the three hydroxy groups are secondary and the third is tertiary. The <sup>13</sup>C n.m.r. spectrum further shows signals at  $\delta$  96.4 and 64.5 p.p.m. due to the acetalic carbon and –CH<sub>2</sub>O– group, respectively. These data show that longikaurin A (1) is pentacyclic. Considering the structures of



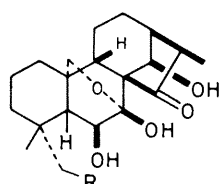
- (1) R<sup>1</sup> = R<sup>2</sup> = R<sup>3</sup> = H  
 (2) R<sup>1</sup> = R<sup>2</sup> = H, R<sup>3</sup> = OAc  
 (3) R<sup>1</sup> = R<sup>3</sup> = H, R<sup>2</sup> = Ac  
 (4) R<sup>1</sup> = OH, R<sup>2</sup> = R<sup>3</sup> = H  
 (5) R<sup>1</sup> = H, R<sup>2</sup> = Ac, R<sup>3</sup> = OAc



(6)

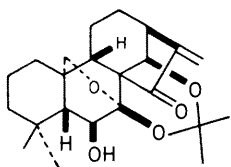


(7)



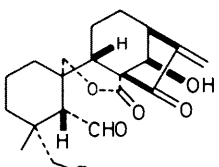
(8) R = H

(9) R = OAc



(10) R = H

(11) R = OAc



(12) R = H

(13) R = OAc

diterpenoids isolated so far from *Rabdosia* species,<sup>3</sup> we presumed that (1) had the *ent*-7 $\beta$ ,20-epoxy-kaur-16-*en*-15-*on*-7 $\alpha$ -ol (7) structure as a basic skeleton. This presumption was supported by the fact that the dihydro-compound (8) shows a negative Cotton effect [ $\lambda_{\max}$  (MeOH) nm ( $\phi$ ): 316 (–5141), 280 (+2981)].

Acetylation of (1) (acetic anhydride–pyridine) gave the monoacetate (3), m.p. 233–235 °C. The <sup>1</sup>H n.m.r. signal for H<sub>b</sub> of (1) was shifted downfield to  $\delta$  5.68 (1H, *J* 1 Hz) indicating that H<sub>b</sub> is located at C-14 $\alpha$  analogous to oridonin (4).<sup>4</sup> A secondary hydroxy group is located at the C-6 $\beta$  position. In INDOR<sup>5,6</sup> experiments, a signal due to nuclear Overhauser enhancement (n.O.e.) was observed for the methyl groups ( $\delta$  1.09, 6H) when monitored at the frequency of H<sub>a</sub>, which appears as a doublet (*J* 7 Hz) after D<sub>2</sub>O treatment. On irradiation of the methyl groups, n.O.e. (6%) was observed for H<sub>a</sub>.

Treatment of (1) with 2,2-dimethoxypropane in HCONMe<sub>2</sub> in the presence of *p*-toluenesulphonic acid gave the acetonide (10), m.p. 194–196 °C [<sup>1</sup>H n.m.r.  $\delta$  1.32 and 1.64 (each 3H, 2  $\times$  s, acetal Me)], confirming  $\beta$ -configuration of the tertiary hydroxy group at C-7. Oxidation of (1) with periodic acid gave the aldehyde-lactone (12) [<sup>1</sup>H n.m.r.  $\delta$  9.90 (1H, d, *J* 6 Hz, aldehydic proton); i.r. (CHCl<sub>3</sub>)  $\nu_{\max}$  1740, 1720, and 1710 cm<sup>-1</sup>], the formation of which established chemically the existence of a hydroxy group at the C-6 $\beta$  position. Thus it was shown that longikaurin A has the structure (1).

Longikaurin B (2), C<sub>22</sub>H<sub>30</sub>O<sub>7</sub>, m.p. 238–239.5 °C, has the following physical properties: [ $\alpha$ ]<sub>D</sub><sup>25</sup> –115.9° (*c* 0.12, C<sub>5</sub>H<sub>5</sub>N);  $\lambda_{\max}$  (MeOH) 234.5 nm ( $\epsilon$  9510);  $\nu_{\max}$  (Nujol) 3490, 3350–3100; 1745, 1710, and 1650 cm<sup>-1</sup>. The <sup>1</sup>H n.m.r. spectrum (in C<sub>5</sub>D<sub>5</sub>N) is very similar to that of longikaurin A (1) except for the signals due to a tertiary methyl group ( $\delta$  1.34, 3H, s), an acetyl group ( $\delta$  1.95, 3H, s), and an acetoxy methyl group [ $\delta$  4.40 and 4.68, each 1H, each AB doublets, *J* 11 Hz]. The <sup>13</sup>C n.m.r. spectrum when compared with that of (1) differs only in the number of methyl groups which decreases from 2 to 1 and the number of oxygenated methylene groups which increases from 1 to 2. These spectral data suggest that in the structure of longikaurin B (2), one of the two methyl groups at C-4 of longikaurin A (1) is oxidized to an acetoxy methyl group. In INDOR<sup>5,6</sup> experiments with (2), signals due to n.O.e.'s were observed for the signals at  $\delta$  3.99 and 4.11 (each 1H, AB doublets, *J* 10 Hz, 20-H<sub>2</sub>) and  $\delta$  1.37 (tertiary methyl group) when an AB doublet at  $\delta$  4.68 was monitored. N.O.e.'s were observed for an AB doublet at  $\delta$  4.68 on irradiation of the tertiary methyl group and 20-H<sub>2</sub> (8 and 11%, respectively). These facts suggest that longikaurin B (2) corresponds to 19-acetoxy-longikaurin A.

Compound (2) gave rise to the monoacetate (5), m.p. 182–183 °C [<sup>1</sup>H n.m.r.  $\delta$  2.04 and 2.06 (each 3H, 2  $\times$  s, 2  $\times$  OAc), 5.66 (1H, br s, 14 $\alpha$ -H)], the acetonide (11) [<sup>1</sup>H n.m.r.  $\delta$  1.34 and 1.64 (each 3H, 2  $\times$  s, acetal Me)], the aldehyde-lactone (13) [<sup>1</sup>H n.m.r.  $\delta$  9.97 (1H, d, *J* 6 Hz,

† The minimal inhibitory concentrations (m.i.c.) of (1) against *Staphylococcus aureus* FDA 209P and *Escherichia coli* NIHJ are 12.5 and >200  $\mu$ g ml<sup>-1</sup>, respectively. Compound (2) also shows m.i.c. against the same bacteria (25 and >200  $\mu$ g ml<sup>-1</sup>, respectively). The detailed study will be published elsewhere.

¶ Unless otherwise noted, <sup>1</sup>H n.m.r. spectra were recorded in CDCl<sub>3</sub> solution and <sup>13</sup>C n.m.r. spectra were recorded in C<sub>5</sub>D<sub>5</sub>N solution, using tetramethylsilane as internal standard.

aldehydic proton),  $\nu_{\max}$  (CHCl<sub>3</sub>) 1740 and 1705 cm<sup>-1</sup>] and the dihydro-compound (9) [ $\alpha_D^{25}$ ,  $\lambda_{\max}$  (MeOH)/nm ( $\phi$ ) 317 (-4004), 280 (+3022)] These facts established that longikaurin B has the structure (2)

The isolation of longikaurin B (2), which is oxygenated at C-19, is the first example among diterpenoids having an

*ent*-kaurene nucleus so far from *Rabdosia* species This substance may be a biosynthetic precursor of the 6,7-seco-*ent*-kaurenoid diterpene of the shikodomin<sup>7</sup> type

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