Preparation of Useful Chiral Lactone Synthons via Stereospecific Enzyme-catalysed Oxidations of meso-Diols

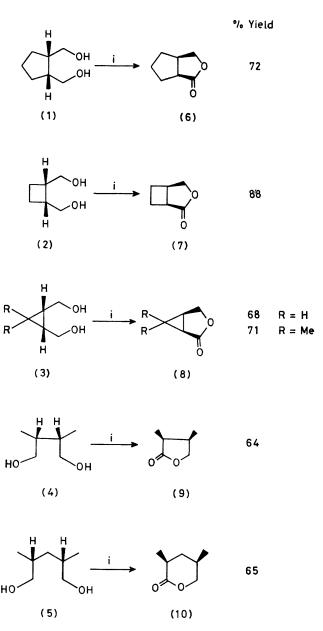
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Summary Horse liver alcohol dehydrogenase-catalysed oxidations of symmetrical acyclic and cyclic *meso*-diol substrates give chiral γ - and δ -lactones in high yields and of 100% enantiomeric excess and provide access to a broad range of useful synthes of value in asymmetric syntheses of natural products.

The unique opportunities provided by enzymes in asymmetric synthesis are becoming increasingly recognised and exploited; of particular value is the ability of many enzymes to catalyse stereospecific transformations of symmetrical substrates into chiral products.¹ A powerful illustration was provided recently by the stereospecific, high-yield, horse liver alcohol dehydrogenase (E.C.1.1.1.1) (HLADH)-catalysed oxidation of the *meso*-substrate *cis*-1,2-bis-(hydroxymethyl)cyclohexane to (1S,2R)-*cis*-2-hydroxymethylcyclohexane carboxylic acid lactone.²

Preparative-scale (up to 2 g of substrate) HLADHcatalysed oxidations of the *cis*-diols (1)— $(5)^{3}$ [†] were effected by the standard procedure⁴ using flavin mononucleotide (riboflavin phosphate)-recycling⁵ of catalytic amounts of the nicotinamide adenine dinucleotide (NAD+) coenzyme. For each of the cyclic or acyclic diols (1)--(5), the oxidations were stereospecific for the hydroxymethylgroups attached to chiral centres of the same absolute configuration types, t with the corresponding hydroxyaldehydes formed initially undergoing further HLADHcatalysed oxidation^{2,4} to give the lactones (6)--(10) of 100% enantiomeric excess (e.e.) in excellent yields. The yields recorded in the Scheme are of the isolated, purified products.§ The optical purities of the lactones (6)-(10) were established by n.m.r. spectroscopy,⁶ and their absolute configurations by conversion into, or comparison with, known reference compounds, as follows: (1S,2R)-(6), $[\alpha]_{D}^{20}$ + 96.9° (c 1, CHCl₃), into (-)-(1R,2R)-methyl 2-methylcyclopentanoate;⁷ (1S,2R)-(7), $[\alpha]_D^{20}$ + 118.7° (c 10, $CHCl_3$, into (-)-(1R,2R)-trans-bis(hydroxymethyl)cyclobutane;⁸ (1S,2R)-(8; R=H), $[\alpha]_{\rm D}^{20}$ -61.8° (c 6.6, CHCl₃), by c.d. comparison with (-)-(1R,2S)-(8; R=Me);(1R,2S)-(8; R=Me), $[\alpha]_{\rm D}^{20}$ -36.6° (c 13.8, CHCl₃), into (+)-(1R,2S)-cis-methylchrysanthemate;⁹ (2S, 3R) - (9), $[\alpha]_{D}^{20}$ + 39.9° (c 11.3, CHCl₃), into (+)-(2S)-ethyl 2,3dimethylbutanoate;¹⁰ and (2S, 4R)-(10), $[\alpha]_{D}^{20}$ + 39.1° (c 10, CHCl₃), into (-)-(2S)-2,4-dimethylpentanol.¹⁰

The above results demonstrate the HLADH-catalysed oxidation of *meso*-diols to be an efficient and versatile route to a broad structural range of enantiomerically pure chiral lactones. Such compounds, which are not readily available



SCHEME. i, HLADH, 20 °C, pH 9, Nad+-recycling, 3 days. 100% E.e. was achieved in each case.

† All new compounds reported have been fully characterised.

 \ddagger *I.e.* S, except for (3; R=Me) for which, as a consequence of the R/S nomenclature rules, the hydroxymethyl group oxidised is the one attached to the *R*-centre.

The reaction mixtures obtained from the oxidations of (4) and (5) contained some of the initially formed hemiacetal precursors of (9) and (10), respectively. These were readily converted into the corresponding lactones by silver carbonate oxidation (ref. 4).

via traditional chemical approaches, are valuable synthons for a number of important natural products For example, (10) could be a key intermediate for the macrolide antibiotic methynolide¹¹ Lactone (10) can also be readily elaborated into the Dutch elm bark beetle pheromone multistriatin,12 as may (8, R=Me) into pyrethroids13 and (7) into the boll weevil pheromone grandisol 14

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- ¹ J B Jones and J F Beck, Tech Chem (N Y), 1976, 10, 107
 ² H B Goodbrand and J B Jones, J Chem Soc, Chem Commun, 1977, 469
 ³ L N Owen and A G Peto, J Chem Soc, 1955, 2383, N L Allinger, M Nakazaki, and V Zalkow, J Am Chem Soc, 1959, 81, 4074, E Voget, K Heinzott, and K Gajik, Justus Liebigs Ann Chem, 1961, 644, 172, L P Kuhn, P R von R Schleyer, and W F Baitinger, J Am Chem Soc, 1964, 86, 650 J Cason and F J Schmitz, J Org Chem, 1963, 28, 555
 ⁴ A J Irwin and J B Jones, J Am Chem Soc, 1976, 98, 8476, 1977, 99, 556, 1625
 ⁵ J B Jones and K E Taylor, Can J Chem, 1976, 55, 2969
 ⁶ I J Jakovac and J B Jones, J Org Chem, 1979, 44, 2165
 ⁷ R K Hill, P J Foleg, Jr, and L A Gardella, J Org Chem, 1967, 32, 2330
 ⁸ J C A Windhorst, PhD thesis, Rijksuniversiteit Te Leiden, Holland, 1975, p 28
 ⁹ W Cocker and H St J Lauder, J Chem Soc, Perkin Trans 1, 1975, 332
 ¹⁰ P A Levine and R E Marken, J Biol Chem, 1935, 111, 299
 ¹¹ A Nakano, S Takimoto, J Inanaga, T Katsuki, S Ouchida, K Inoue, M Aiga, N Okukado, and M Yamaguchi, Chem Lett.
- ¹¹ A Nakano, S Takimoto, J Inanaga, T Katsuki, S Ouchida, K Inoue, M Aiga, N Okukado, and M Yamaguchi, Chem Lett, 1979, 1019
- ¹² P A Bartlett and J Myerson, J Org Chem, 1979, 44, 1625
 ¹³ M Elhott and N F Janos, Chem Soc Rev, 1978, 7, 473
 ¹⁴ cf R D Clark, Synth Commun, 1979, 325