

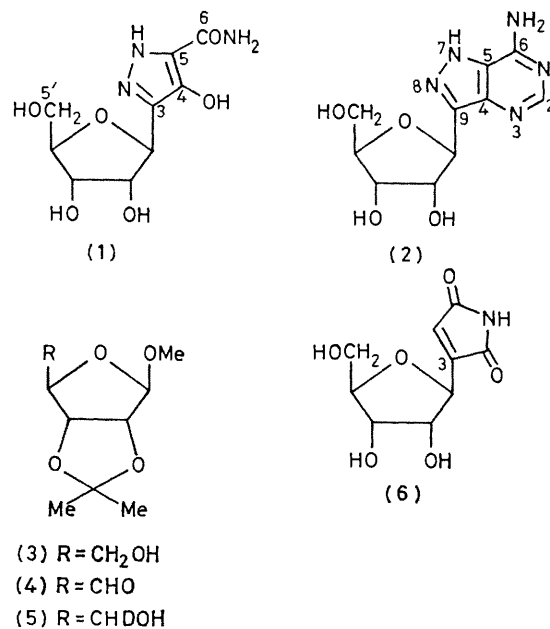
The Biosynthesis of Pyrazofurin and Formycin

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Summary Evidence is presented that D-ribose and L-glutamate are the principal biosynthetic precursors of pyrazofurin (1) and formycin (2).

THE C-nucleoside antibiotic pyrazofurin (pyrazomycin)¹ (1), a metabolite of *Streptomyces candidus* possessing both anti-viral and anti-tumour activity,¹ contains a pyrazole ring, a very rare feature amongst natural products. We now report experiments casting light on the biosynthesis of pyrazofurin (1), and of the structurally related antibiotic formycin¹ (2).

The results of experiments with various potential precursors of pyrazofurin in labelled form are summarised in the Table. The high incorporation of [1-¹⁴C]ribose, considerably greater than that of [1-¹⁴C]glucose, strongly implies that ribose is a direct precursor of the ribofuranosyl residue of pyrazofurin. This was confirmed by means of an experiment using deuterium labelling. Oxidation of methyl 2,3-O-isopropylidene-β-D-ribofuranoside² (3) with chromium trioxide-pyridine, and reduction of the resultant aldehyde³ (4) with sodium borodeuteride gave the labelled ribofuranoside (5); the introduction of one deuterium atom at C-5 was confirmed by ¹H n.m.r. spectroscopy. The deuteriated ribose (0.64 g) produced by acid hydrolysis of (5) was administered, mixed with [1-¹⁴C]ribose, to *Streptomyces candidus*. The pyrazofurin subsequently isolated and recrystallised (0.15 g) had specific activity 7.9% that of the ribose fed. The proton-decoupled ²H n.m.r. spectrum (55.3 MHz, H₂O) of this pyrazofurin showed one signal at δ 3.82.† The ¹H n.m.r. spectrum of pyrazofurin under similar conditions (360 MHz, D₂O) is completely resolved,



and the protons of the 5'-position appear, as the AB part of an ABX system, centred at δ 3.73.† The intact and specific incorporation of ribose is thus demonstrated.

With regard to the origins of the pyrazole ring and amide carbon, the negligible incorporation of label from hydrogen carbonate (see Table) indicates that the amide carbon is

† Chemical shifts were measured relative to HOD (taken as δ 4.70) as internal standard; thus some deviations between spectra are to be expected.

