

## New Approach to Aminoglycoside Antibiotics

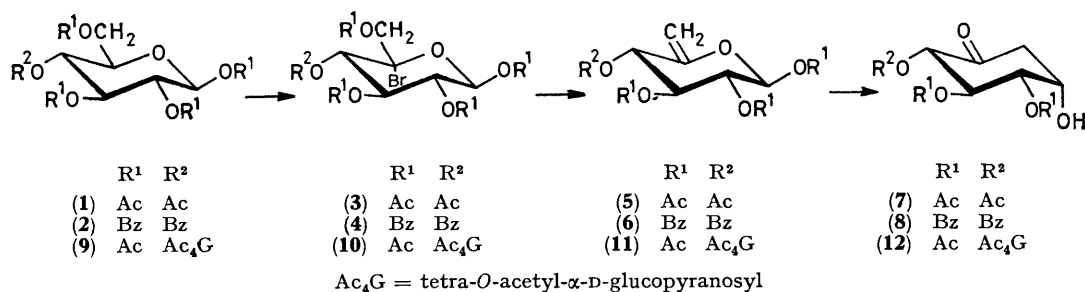
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**Summary** Photobromination of peracylated hexopyranoses and treatment of the resultant 5-bromo-derivatives with zinc and acetic acid afforded 6-deoxy-5-enose esters which, with mercury(II) salts in aqueous media, gave 2-deoxyinosose triesters, applied to  $\beta$ -D-maltose octa-acetate, these procedures afforded an  $\alpha$ -glucosylated inosose derivative which on further  $\alpha$ -glycosylation gave a 'pseudotrissaccharide' related to those upon which aminoglycoside antibiotics are based

(5) and (6), respectively<sup>3</sup> In addition, we have shown that such alkenes are converted into 2-deoxyinosose triesters when heated in aqueous media in the presence of mercury(II) salts<sup>4</sup> By a combination of these processes the inososes (7) and (8) have now been produced as shown, and in 45 and 50% yield, from penta-*O*-acetyl- and -*O*-benzoyl- $\beta$ -D-glucopyranose, respectively The acetate (7) has m p 142—144 °C,  $[\alpha]_D -5.5^\circ$  (CHCl<sub>3</sub>), and the benzoate (8) m p 196—197 °C,  $[\alpha]_D -4^\circ$  (CHCl<sub>3</sub>), and both give <sup>1</sup>H n m r spectra consistent with the assigned structures In particular, the methylene group resonances are observable near  $\delta$  2.7 and 2.9, H-1 signals are at  $\delta$  4.35 and 4.6, respectively, and the configurations at all the chiral centres are readily determined from the ring proton splitting patterns The stereochemistry at the new chiral centres (C-1, carbohydrate numbering) is the same as was found for the original rearrangement product<sup>4</sup>

We have recently reported that penta-*O*-acyl- $\beta$ -D-glucopyranoses (1) and (2), on treatment with bromine radicals generated photochemically from bromine or *N*-bromosuccinimide, undergo substitution at C-5,<sup>1,2</sup> and that the products (3) and (4), with zinc and acetic acid, afford mainly the corresponding 6-deoxyhex-5-enopyranose esters

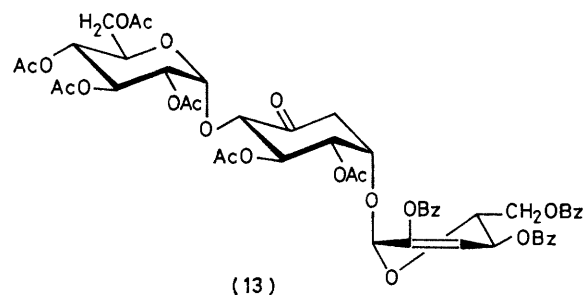


Whereas glycopyranosyl esters react with bromine radicals at C-5, methyl tetra-*O*-benzoyl- $\beta$ -D-glucopyranoside afforded 48% of a 2-bromoaldono-1,5-lactone ester,<sup>5</sup> suggesting that initial radical abstraction occurred in this case at the anomeric centre. It was also observed that  $\alpha$ -anomers of both glycopyranosides<sup>6</sup> and glycopyranosyl esters<sup>2</sup> react less readily than do the corresponding  $\beta$ -compounds, conceivably because, with the  $\alpha$ -glycosides, the equatorial anomeric hydrogen atoms are less readily abstracted<sup>7</sup> and, with the glycosyl esters, the axial acyloxy groups at C-1 inhibit approach by bromine radicals to the *syn*-axial hydrogen atoms at C-5. It was therefore rationalised that octa-*O*-acetyl- $\beta$ -D-maltose (9), on photobromination, could give the 5-bromide (10), and when it was treated in boiling carbon tetrachloride in the presence of *N*-bromosuccinimide and under a 275 W tungsten filament, reflector heat lamp for 2 h, a chromatographically more mobile product was obtained, as anticipated, together with several less mobile compounds. Treatment of the unfractionated mixture with zinc and acetic acid, followed by chromatography on a column of silica gel, gave the known alkene (11) in 12% yield after recrystallisation from methanol. Recrystallised further from ether—light petroleum, it had m.p. 132—134 °C [ $\alpha$ ] + 60° (CHCl<sub>3</sub>),  $\nu_{\max}$  1665 cm<sup>-1</sup> (C=C str). It has previously been prepared<sup>8</sup> by an 8-step procedure, and in about 6% yield, from compound (9) [m.p. 128—131.5 °C, [ $\alpha$ ] + 64° (CHCl<sub>3</sub>)].

When the alkene (11) was treated in refluxing aqueous acetone with a trace of added acetic acid, in the presence of mercury(II) acetate, the expected rearrangement<sup>4</sup> occurred to give the glycosylinosose (12) as a syrup, [ $\alpha$ ]<sub>D</sub> + 77° (CHCl<sub>3</sub>). The <sup>1</sup>H n.m.r. spectrum of this product showed the 2-proton resonance at  $\delta$  2.7 expected for the carbocyclic product as well as other consistent spectral features.

This product possesses basic structural features of many 'pseudodisaccharides' of the aminoglycoside antibiotic group;<sup>9</sup> in particular, the interunit linkage has the required  $\alpha$ -configuration. Furthermore, it has one free hydroxy-group rendering it subject to specific further glycosylation; and when treated with 1,5-anhydro-2,3,4,6-tetra-*O*-benzoyl-

*D*-arabino-hex-1-enitol in cooled dichloroethane in the presence of boron trifluoride-ether,<sup>10</sup> it gave the crystalline  $\alpha,\alpha$ -linked 'pseudotrisaccharide' (13) [m.p. 94—97 °C, [ $\alpha$ ] + 110° (CHCl<sub>3</sub>)] in 68% yield after chromatographic purification. This product had a <sup>1</sup>H n.m.r. spectrum with the appropriate acetyl and benzoyl resonances, the signal for the methylene group of the inosose moiety ( $\delta$  2.6), and a broadened singlet at  $\delta$  6.00 which is characteristic of that expected for H-3 of the unsaturated unit.<sup>10</sup>



Compound (13) contains the di-*O*- $\alpha$ -D-hexopyranosylcyclohexane structure common to many antibiotic substances, although the relationship of the glycosyl units is 1,4 rather than the normal 1,3.<sup>9</sup> The glycosylation procedure involving the unsaturated sugar derivative has been used before to prepare  $\alpha$ -linked unsaturated 'pseudodisaccharides' from which biologically active unprotected compounds containing deoxy- and aminodeoxy-groups have been prepared,<sup>11</sup> and, in parallel work, diglycosylation of a 2,5-dideoxystreptomine derivative with tri-*O*-acetyl-1,5-anhydro-2-deoxy-*D*-arabino-hex-1-enitol gave a 'pseudotrisaccharide' derivative containing two double bonds from which unprotected antibiotics were obtained.<sup>12</sup>

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<sup>1</sup> R. Blattner and R. J. Ferrier, *J. Chem. Soc., Perkin Trans. 1*, 1980, 1523.

<sup>2</sup> R. J. Ferrier and P. C. Tyler, *J. Chem. Soc., Perkin Trans. 1*, 1980, 1528.

<sup>3</sup> R. Blattner, R. J. Ferrier, and P. C. Tyler, *J. Chem. Soc., Perkin Trans. 1*, 1980, 1535.

<sup>4</sup> R. J. Ferrier, *J. Chem. Soc., Perkin Trans. 1*, 1979, 1455.

<sup>5</sup> R. J. Ferrier and P. C. Tyler, *J. Chem. Soc., Perkin Trans. 1*, 1980, accepted for publication.

<sup>6</sup> P. C. Tyler, unpublished observations.

<sup>7</sup> K. Hayday and R. D. McKelvey, *J. Org. Chem.*, 1976, **41**, 2222.

<sup>8</sup> M. Mori, M. Hoga, and S. Tejima, *Chem. Pharm. Bull.*, 1974, **22**, 1331; H. Goto, M. Mori, and S. Tejima, *ibid.*, 1978, **26**, 1926.

<sup>9</sup> S. Umezawa, *Adv. Carbohydr. Chem. Biochem.*, 1974, **30**, 111; *Pure Appl. Chem.*, 1978, **50**, 1453.

<sup>10</sup> R. J. Ferrier, N. Prasad, and G. H. Sankey, *J. Chem. Soc. (C)*, 1979, 587.

<sup>11</sup> C. Colas, B. Quiclet-Sire, J. Cléophax, J.-M. Delaunéy, A.-M. Sepulchre, and S. D. Géro, *J. Am. Chem. Soc.*, 1980, **102**, 857.

<sup>12</sup> A. Canas-Rodríguez and A. Martínez-Tobed, *Carbohydr. Res.*, 1979, **68**, 43.