

## Nitroamine Radicals as Intermediates in the Functionalization of Non-activated Carbon Atoms

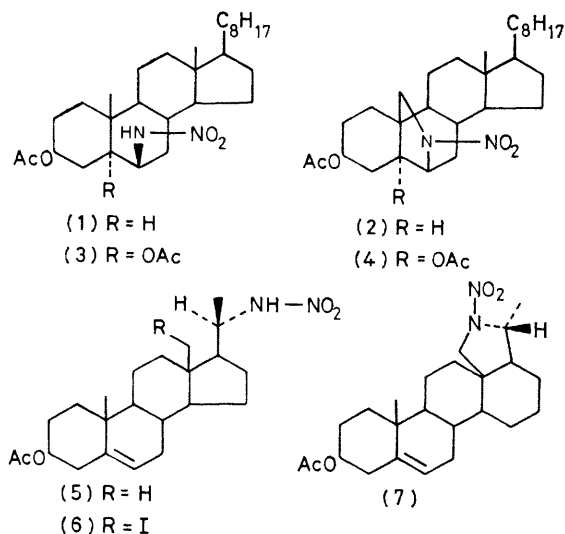
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**Summary** Photolysis of *N*-iodonitroamines generated *in situ* from the steroidal nitroamines 6 $\beta$ -nitroamino-5 $\alpha$ -cholestan-3 $\beta$ -ol acetate, 6 $\beta$ -nitroamino-5 $\alpha$ -cholestane-3 $\beta$ ,5 $\alpha$ -diol diacetate, and 20*R*-nitroaminopregn-5-en-3-ol acetate removes hydrogen atoms from the Me-18 and Me-19 groups to give 6 $\beta$ ,19-*N*-nitroepimino-5 $\alpha$ -cholestan-3 $\beta$ -ol acetate, 6 $\beta$ ,19-*N*-nitroepimino-5 $\alpha$ -cholestane-3 $\beta$ ,5 $\alpha$ -diol diacetate, and 18,20*R*-*N*-nitroepimino-5-en-3 $\beta$ -ol acetate

THE free radicals needed to introduce substituents into the C-18 and C-19 steroidal methyl-groups and also into other

skeletal positions have been generated by the fragmentation of *N*-halogenoamines (Hofmann-Löffler reaction), *N*-halogenoamides, azides, nitrites (Barton reaction), alcohols, and hypohalites, and also by ketone irradiation<sup>1</sup> Of these, those that give the epimino-compounds are generally the most difficult to apply Hence we are interested in developing a method which enables remote functionalization by the amine radicals R-N $\cdot$ -X We report here results obtained by the generation of these radicals (X = NO<sub>2</sub>) in the steroidal substrates (1) [m p 194—196 °C, [ $\alpha$ ]<sub>D</sub> - 42° (CHCl<sub>3</sub>)], (3) [m p 246—247 °C, [ $\alpha$ ]<sub>D</sub> - 84° (CHCl<sub>3</sub>)], and (5) [m p 189—190 °C, [ $\alpha$ ]<sub>D</sub> - 56° (CHCl<sub>3</sub>)] †

† The nitroamines were prepared by NaBH<sub>4</sub> reduction (M J Haire and G A Boswell, Jr, *J Org Chem*, 1977, **42**, 4251) of the corresponding nitroimines The 6-nitroimino-5 $\alpha$ -cholestan-3 $\beta$ ,5 $\alpha$ -diol diacetate has been described previously (A G González, R Freire, M G García-Estrada, J A Salazar, and E Suárez, *An Quim*, 1972, **68**, 1145) The 6-nitroimino-5 $\alpha$ -cholestan-3 $\beta$ -ol acetate [amorphous,  $\nu_{\max}$  (CHCl<sub>3</sub>) 1625, 1560, and 1320 cm<sup>-1</sup>] and the 20-nitroiminopregn-5-en-3 $\beta$ -ol acetate [m p 175—176 °C, [ $\alpha$ ]<sub>D</sub> - 21°,  $\nu_{\max}$  (KBr) 1620, 1580, and 1315 cm<sup>-1</sup>] have been prepared from the corresponding ketoximes and nitrous acid (G Buchi and H Wuest, *J Org Chem*, 1979, **44**, 4116) Full details will be reported elsewhere



In a typical procedure a solution of the nitroamine in cyclohexane or dichloromethane, treated with  $I_2$  (1 mol. equiv.) and  $Pb(OAc)_4$  or  $HgO$  (4 mol. equiv.) at  $45^\circ C$ , was irradiated with two 150 W tungsten filament lamps for 1 h.

The acetate of 6 $\beta$ ,19-*N*-nitroepimino-5 $\alpha$ -cholestan-3 $\beta$ -ol (2), amorphous, was obtained from (1)† in a yield of 63%§ [chemical ionisation mass spectrum  $M^+ + NH_3$ ,  $m/e$  505; mass spectrum  $M^+ - NO_2$ ,  $C_{29}H_{46}NO_2$ ,  $m/e$  442·3767;  $\nu_{max}$  ( $CHCl_3$ ) 1730, 1490, and 1315  $cm^{-1}$ ;  $\lambda_{max}$  (EtOH)

241 nm ( $\epsilon$  7400);  $^1H$  n.m.r. *inter alia*  $\delta(CDCl_3)$  4·39 (d,  $J$  4 Hz, H-6 $\alpha$ ) and 3·73br (s,  $w_{\frac{1}{2}}$  4 Hz,  $2 \times$  H-19);  $^{13}C$  n.m.r. *inter alia*  $\delta(CDCl_3)$  65·1 (C-6) and 51·2 p.p.m. (C-19)].

Similarly, irradiation of 6 $\beta$ -nitroamino-5 $\alpha$ -cholestan-3 $\beta$ ,5 $\alpha$ -diol diacetate (3) gave (4) (60%)§, amorphous, [ $M^+ - NO_2$ ,  $m/e$  500;  $\nu_{max}$  ( $CHCl_3$ ) 1730, 1490, and 1315  $cm^{-1}$ ;  $^1H$  n.m.r.  $\delta(CDCl_3)$  5·17 (d,  $J$  4 Hz, H-6 $\alpha$ ) and 3·73br (s,  $w_{\frac{1}{2}}$  4 Hz,  $2 \times$  H-19);  $^{13}C$  n.m.r.  $\delta(CDCl_3)$  63·8 (C-6) and 49·8 p.p.m. (C-19)].

The same general process was also applied to the functionalization of the 18-methyl-group in (5). In this case the 18-iodo-derivative (6) was obtained (*ca.* 25%)§ [m.p. 192—194  $^\circ C$  (decomp.),  $M^+ - NO_2$ ,  $m/e$  470;  $\nu_{max}$  3390, 3250, 1730, 1575, and 1315  $cm^{-1}$ ,  $\lambda_{max}$  (EtOH) 224 nm ( $\epsilon$  10,300);  $^1H$  n.m.r.  $\delta(CDCl_3)$  3·05, 3·26 (ABq,  $J_{AB}$  11 Hz,  $2 \times$  H-18), and 4·25 (m, H-20);  $^{13}C$  n.m.r.  $\delta(CDCl_3)$  55·75 (C-20) and 7·1 p.p.m. (C-18)]. Treatment of compound (6) with silver acetate (4 mol. equiv.) in acetone (18 h at room temperature) gave 18,20*R*-*N*-nitroepiminopregn-5-en-3 $\beta$ -ol acetate (7) (100%) [m.p. 190—192  $^\circ C$ ,  $M^+ - NO_2$ ,  $m/e$  356;  $\nu_{max}$  1725, 1490, and 1305  $cm^{-1}$ ;  $\lambda_{max}$  (EtOH) 243 nm ( $\epsilon$  9100);  $^1H$  n.m.r.  $\delta(CDCl_3)$  4·38 (q,  $J$  6 Hz, H-20) 3·7br (s,  $w_{\frac{1}{2}}$  4 Hz,  $2 \times$  H-18);  $^{13}C$  n.m.r.  $\delta(CDCl_3)$  53·7 (C-18) and 65·5 p.p.m. (C-20)].

Since the reduction of nitroamines to amines is a known reaction,<sup>2</sup> this cyclization constitutes a formal synthesis of 1,4-epimino-compounds.

A.R. thanks the Centro Iberoamericano de Cooperación of Spain for a fellowship.

(Received, 2nd July 1980; Com. 719.)

† All new crystalline compounds gave satisfactory analytical data.

§ The yields, which have not been optimized, are of the same order as those described for the photolysis of the corresponding hypoiodites.

<sup>1</sup> K. Heusler and J. Kalvoda in 'Organic Reactions in Steroids Chemistry,' Vol. 2, eds. J. Fried and J. A. Edwards, Van Nostrand Reinhold, New York, 1971, p. 237; D. N. Kirk and M. P. Hartshorn, 'Steroid Reaction Mechanisms,' Elsevier Publishing, Amsterdam, 1968, p. 394; J. Kalvoda and K. Heusler, *Synthesis*, 1971, 501; K. Heusler and J. Kalvoda, *Angew. Chem. Int. Ed. Engl.*, 1964, 3, 525.

<sup>2</sup> P. Bruck and A. H. Lamberton, *J. Chem. Soc.*, 1955, 3997.