

Asymmetric Induction in the Rearrangement of Monocyclic Endoperoxides into γ -Hydroxy- $\alpha\beta$ -unsaturated Aldehydes

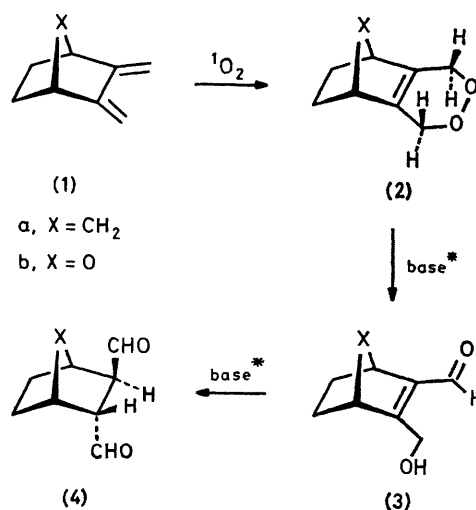
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Summary The endoperoxide obtained by addition of singlet oxygen to 2,3-bis(methylene)-7-oxanorbornane is isomerised to the chiral γ -hydroxy- $\alpha\beta$ -unsaturated aldehyde by catalytic amounts of various natural bases with an enantiomeric excess ranging up to 46%

EXAMPLES of chiral bases showing selectivity between enantiotopic protons in prochiral molecules are rare¹ Recently, Whitesell and Felman² reported a 31% optical yield for the isomerization of cyclohexene oxide into cyclohex-2-enol promoted by a chiral lithium amide We report on the asymmetric induction in the Kornblum-DeLaMare^{3,4} rearrangement of monocyclic endoperoxides into γ -hydroxy- $\alpha\beta$ -unsaturated aldehydes by optically active natural bases

The endoperoxides (**2a**)⁵ and (**2b**) (m p 57–59 °C) were obtained by singlet-oxygen addition to 2,3-bis(methylene)norbornane (**1a**) and 2,3-bis(methylene)-7-oxanorbornane (**1b**),⁶ respectively In the presence of a catalytic amount of $\text{Rh}_2(\text{CO})_4\text{Cl}_2$, (**2a**) and (**2b**) were isomerized into (**3a**) and (**3b**) [¹H n m r δ (CDCl₃) 9.98 (s, 1 H), 5.25 and 5.01 (2d, *J* 3.5 Hz, 2 H), 4.61 (s, 2 H), 2.0–1.7 (m, 2 H), and 1.25 (m, 2 H)], respectively The *s-trans*-configuration of (**3a**) and (**3b**) was indicated by the i r spectra { ν_{OH} 3630 (narrow) and 3400–3500 cm⁻¹ (broad), decreases upon dilution (more important in CCl₄ than in CH₂Cl₂), $\nu_{\text{HC=O}}$ 1665 cm⁻¹ [1660–1665 cm⁻¹ in the corresponding α -methoxy- α -trifluoromethyl- α -phenylacetate (MTPA)⁷ derivatives]} In the presence of 0.0045 M (+)-quinidine, a 10% solution of



(**2a**) in MeCN was rearranged slowly into (–)-(3a) at 20 °C ($\tau_{\frac{1}{2}}$ ca 60 h) with an enantiomeric excess (e e) of 9%, as determined by the ¹H and ¹⁹F n m r spectra of the MTPA derivative⁷ Other natural bases such as (–)-cinchonidine and (–)-ephedrine catalysed the conversion (**2a**) → (+)-(3a) with a lower optical yield While (–)-noradrenaline and (–)-proline were poor catalysts, brucine induced a relatively fast isomerization (**2a**) → (+)-(3a) ($\tau_{\frac{1}{2}}$ ca. 4 h,

20 °C, 0.0045M in MeCN) with relatively good optical yields (12–29%), but poor chemical yields. This was because brucine catalysed the isomerization (**3a**) → (**4a**)⁸ competitively and enantioselectively. Using (±)-(**3a**), we found, after 50% transformation, an e.e. of 16% in the remaining (+)-(**3a**) (for 100% e.e., an $[\alpha]_D^{25}$ ca. 135° was calculated).

In the rearrangement of (**2b**), it was envisaged that the OH group in a base such as cinchonine, cinchonidine, quinidine, or quinine could lead to better asymmetric induction because of the possibility of hydrogen bonding between the oxygen bridge of the 7-oxanorbornane and the catalyst. Indeed, while brucine (no OH group) catalysed (**2b**) → (**3b**) and (**3b**) → (**4b**) with low optical yields (< 5%), (+)-quinidine [0.0045 M, 10% (**2b**) in MeCN, 20 °C, $\tau_{\frac{1}{2}}$ ca. 17 h] induced the rearrangement (**2b**) → (**3b**) with an e.e. of 46–47% (by ¹H-n.m.r. of the MTPA derivative, see the Figure) and an n.m.r. chemical yield > 95% [(**3b**) was stable in solution only; attempts to isolate it led to decomposition]. (+)-Cinchonine (0.017 M, MeCN, 20 °C, $\tau_{\frac{1}{2}}$ ca. 5 h) and (–)-cinchonidine (0.017 M, MeCN, 20 °C, $\tau_{\frac{1}{2}}$ ca. 1.5 h) gave (**3b**) with opposite optical rotations and an e.e. of 19 and 23%, respectively.

Although the absolute configurations of (+)-(**3a**) and (+)-(**3b**) are not established yet, these optically active compounds are potential synthetic intermediates. A different combination of base and solvent might lead to even

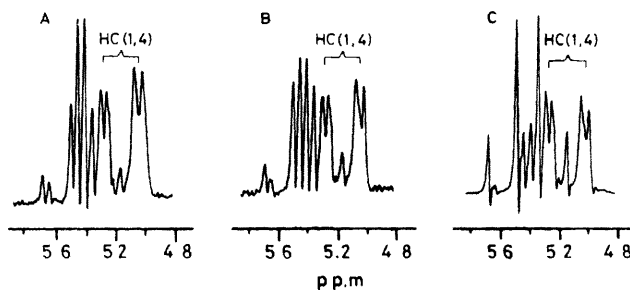


FIGURE. ¹H N.m.r. spectra (80 MHz, CD₃CN) of the MTPA of (**3b**) obtained by the catalysed isomerization of (**2b**) by: (A) (–)-cinchonidine; (B) Rh₂(CO)₄Cl₂; and (C) (+)-quinidine [–CH₂-O-MTPA and H-C(1,4) are shown].

higher optical yields. It is striking that a relatively weak base such as quinidine can distinguish between enantiotopic protons² that are three carbons away from the chiral centres C(1,4) of the norbornane derivatives (**2a**) and (**2b**).

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