

Acid-catalysed Photoisomerization of Alkylindens¹

By HARRY MORRISON* and DAVID GIACHERIO

(Department of Chemistry, Purdue University, W Lafayette, Indiana 47907)

Summary Irradiation of acidic solutions of indens containing an alkyl group at the 2-position results in the facile formation of 2-alkylideneindans

RECENT investigations of the photochemistry of substituted indens have revealed two modes of isomerization, one involving the [1,5] shift of an aromatic ring² and the other resulting in the transposition of C-1 and C-2³. We report here a third photoisomerization, the acid-catalysed transformation [$h\nu$ (254 nm), *ca* 0.01 M inden, MeCN (non-degassed), 0.01 M HCl] of 2-alkylindens into 2-alkylideneindans. The reaction is illustrated in equations (1) and (2)

(the yields shown were determined by v p c using internal standards).

A more general survey of the scope of this novel reaction is presented in the Table. Though the quantum efficiencies (ϕ) are generally *ca* 0.1, note that alkyl substitution at the 3-position markedly inhibits the reaction (as it does to the previously reported C-6-C-2 phototransposition)^{3a}. There is also a striking effect by substituents on the aromatic ring, with the 6-methoxy-group and 5-trifluoromethyl group appreciably lowering the quantum efficiency.

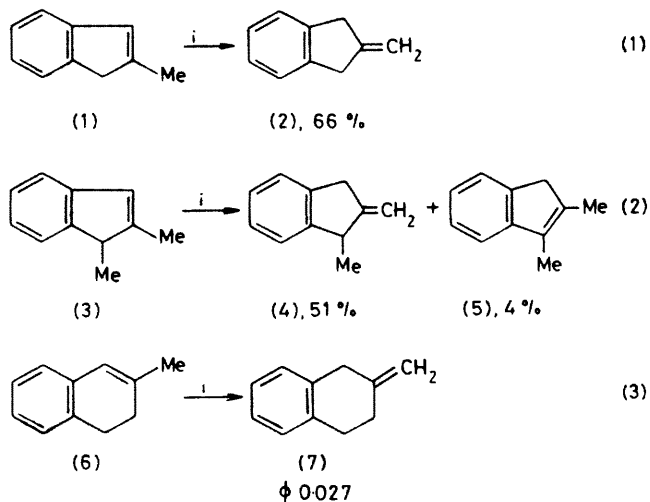
The reaction is not unique to the inden series, the dihydronaphthalene (6) also isomerizes under the same

TABLE Quantum efficiencies for the formation of *exo*-cyclic olefins.*

Reactant	Product	ϕ Product
(1)	(2)	0.13
(3)	(4)	0.10
2,3-Dimethylinden	(4)	0.019
2-Ethyl-1-methylinden	2-Ethylidene-1-methylindan (<i>E</i> + <i>Z</i>)	0.087
3-Ethyl-2-methylinden	1-Ethyl-2-methyleneindan	0.03
5-Methoxy-2-methylinden	5-Methoxy-2-methyleneindan	0.12
6-Methoxy-2-methylinden	"	0.004
5-Trifluoromethyl-2-methylinden	5-Trifluoromethyl-2-methyleneindan	< 0.01
6-Trifluoromethyl-2-methylinden	"	0.085

* Determined at 24 ± 2 °C using 254 nm light on *ca* 0.01 M solutions of the inden in non-degassed acetonitrile containing 0.01 M HCl.

photolysis conditions [equation (3)]. However, no comparable double bond migration is observed for 2-methyl-1-phenylpropene.



i, $h\nu$ (254 nm), MeCN, 0.01 M HCl.

(Received, 10th June 1980; Com. 634.)

Though the inden isomerization superficially resembles the photoinduced, acid-catalysed endo- to exo-cyclic olefin migrations of cyclohexenes,⁴ these latter reactions proceed *via* a *trans*-cycloalkene not observable in 5-membered ring systems. [Such a mechanism cannot be eliminated for reaction (3), but the direction of the requisite protonation and the lack of 1,4-dihydronaphthalene as a product argue against a *trans*-cycloalkene intermediate.] Isomerization of the indens proceeds *via* the excited singlet-state (triplet sensitization only gives dimers) and involves protonation at C-3 [photolysis of (1) with DCl results in the incorporation of one deuterium atom at the benzylic position of (2)]. Trifluoroacetic acid also catalyses the reaction, but perchloric and sulphuric acids are ineffective. The acid is interacting with an intermediate other than the excited singlet-state since fluorescence is *not* affected by HCl. Acid-catalysed olefin-migration is accompanied by a corresponding quenching of the inden phototransposition. A detailed discussion of the mechanism of this interesting double bond migration and its relationship to the transposition will be presented in a full paper.

Partial financial support by the U.S. Army Research Office and a David Ross Fellowship to D. G. by the Purdue Research Foundation are gratefully acknowledged.

¹ For the previous paper in the series 'Organic Photochemistry,' see H. Morrison, D. Avnir, C. Bernasconi, and G. Fagan, *Photochem. Photobiol.*, in the press.

² J. K. DeFonseka, C. Manning, J. J. McCullough, and A. J. Yarwood, *J. Am. Chem. Soc.*, 1977, **99**, 8257 and references therein.

³ (a) F. J. Palensky and H. Morrison, *J. Am. Chem. Soc.*, 1977, **99**, 3507; (b) D. Giacherio and H. Morrison, *ibid.*, 1978, **100**, 7109; (c) A. Padwa, R. Loza, and D. Getman, *Tetrahedron Lett.*, 1977, 2847; (d) A. Padwa, S. Goldstein, R. Loza, and M. Pullwer, *ibid.*, 1979, 4895.

⁴ W. G. Dauben, H. C. H. A. van Riel, J. D. Robbins, and G. J. Wagner, *J. Am. Chem. Soc.*, 1979, **101**, 6383.