Rearrangement of Allylic Silylsulphonium Ylides. A Synthesis of βγ-Unsaturated Aldehydes

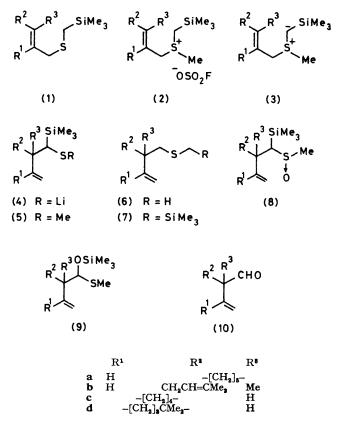
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Summary The ylides derived from the reaction of allylic silylmethylsulphonium salts with n-butyl-lithium rearrange to homoallylic α -methylthiosilanes which are precursors to $\beta\gamma$ -unsaturated aldehydes.

SILICON can have a pronounced effect on the chemistry of sulphur compounds. For example carbanions co-operatively stabilized by both elements¹ form readily at low temperature and are stable and moderately nucleophilic. Similarly, silicon-activated Pummerer rearrangement of sulphoxides² proceeds at room temperature or below to give hydrolytically labile *O*-trimethylsilylhemithioacetals. A combination of these advantageous properties was recently used in a new procedure for nucleophilic formylation.^{3,4} We now report a permutation which gives $\beta\gamma$ -unsaturated aldehydes.

When the S-allylic thiomethylsilane (1c) was metallated at 0 °C with n-butyl-lithium in tetrahydrofuran (THF), a [2,3] sigmatropic rearrangement of the resultant carbanion was observed, to give the lithium thiolate (4c) which could be alkylated with methyl iodide to give the sulphide (5c) in 55% yield accompanied by several minor by-products which were difficult to separate. By contrast, the S-allylic Strimethylsilylmethylsulphonium fluorosulphonate salts (2), prepared in high yield from the sulphides $(1)^5$ by reaction with methyl fluorosulphonate in CH₂Cl₂ at 0 °C, reacted with n-butyl-lithium in THF at -78 °C to give the ylides derived from selective deprotonation of the carbon bearing silicon and sulphur. The ylides (3) underwent a [2,3] sigmatropic rearrangement below -20 °C to give the homoallylic α methylthiosilanes (5) cleanly and in good yield (65-75%). Minor by-products (6) and (7) (ca. 10% each) presumably



arose from rearrangement of alternative ylides generated by deprotonation of the SMe group in the case of (7) and desilylation in the case of (6).

The rearranged α -methylthiosilanes (5) were converted into the aldehydes (10) in 40-60% overall yield via the sila-Pummerer reaction. Oxidation of (5) with 1 mol. equiv. of m-chloroperbenzoic acid in CH₂Cl₂ at -40 °C gave the thermally labile sulphoxides (8) which had rearranged to the O-trimethylsilylhemithioacetals (9) after the usual extractive work-up. These were hydrolysed in aqueous THF at 20 °C for 18 h in the presence of 1 mol. equiv. of oxalic acid. It is noteworthy that in the synthesis of γ -cyclocitral (10d),

the hydrolysis of (9d) proceeded without rearrangement of the unstable double bond into conjugation with the carbonyl group.

The present work expands the scope of carbon-carbon bond-forming reactions available to nucleophiles stabilized by sulphur and silicon and provides a useful application of silicon-substituted sulphonium ylides about which little is known.6

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- ¹ D. Peterson, J. Org. Chem., 1958, **33**, 780. ² A. G. Brook, Acc. Chem. Res., 1974, **7**, 77.
- ³ P. J. Kocienski, Tetrahedron Lett., 1980, 21, 1559.
- ⁴ D. J. Ager and R. C. Cookson, Tetrahedron Lett., 1980, 21, 1677.
- ⁵ Prepared by reaction of the allylic chloride with trimethylsilylmethanethiol in ethanolic sodium ethoxide. The thiol is known: D. C. Noller and H. W. Post, J. Org. Chem., 1952, 17, 1393. ⁶ F. Cooke, P. Magnus, and G. L. Bundy, J Chem. Soc., Chem. Commun., 1978, 714.