New Synthesis of Indoles from o-Acylaminobenzyltriphenylphosphonium Salts

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Summary The action of bases on *o*-acylaminobenzyltriphenylphosphonium salts gives indoles in high yields.

MADELUNG synthesis of indoles¹ by cyclization of *o*-methylanilides is initiated only by strong bases at elevated temperatures and the useful scope of this synthesis is therefore limited to molecules which can survive these very drastic conditions. We report herein a new general synthetic procedure for the synthesis of indoles which is based on the intramolecular reaction of amides with alkylidenephosphoranes.[†] The synthetic approach is outlined in reaction (1).



A typical procedure is as follows: $1\cdot 1$ equiv. of a suitable base (MeONa, Bu^tOK, or sodium t-pentoxide) was added (10-15 min) to a stirred suspension of 1 equiv. of dry phosphonium salt (1) in boiling toluene. The solution was stirred for 10 min, filtered, and evaporated to dryness. The crude material was essentially the desired indole, contaminated with triphenylphosphine oxide. The yields (Table) for reaction (1) are for isolated pure products (2)obtained by preparative layer chromatography on silica gel (1: 1 ether-benzene), or by crystallisation from ethanol.

The examples in the Table demonstrate the generality of the synthesis. Not only is this reaction compatible with a wide variety of substituents at the 2 position, but our procedure can also be used to prepare 1-substituted indoles.

TABLE.	Preparation	of	indoles	(2).
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R1	R ²	Yield/%	Mp./°C
н	Ме	96	61ª
н	CH _o =C(Me)	93	118-120
н	"Ph` ´	97	189—190 ^b
н	p-NO ₂ C ₂ H ₄	93	249251°
н	p-MeOC H	85	228-2304
Ph	1 Me	80	Oil

[•] Lit., m.p. 62 °C; L. Marion and C. W. Oldfield, *Can. J. Res.*, 1947, 25B, 1. ^b Lit., m.p. 188–189 °C; J. I. G. Cadogan and M. Cameron-Wood, *Proc. Chem. Soc.*, 1962, 361. °Lit., m.p. 251–252 °C, and ^c lit., m.p. 230–230.5 °C; C. E. Blades and A. L. Wilds, *J. Org. Chem.*, 1956, 21, 1031.

 \dagger To our knowledge, amides failed to react with phosphorus ylides; the only published exception is the reaction of Ph₃P=CHCO₂Et with a lactam (W. Flitsch and H. Peters, *Tetrahedron Lett.*, 1969, 1161).

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The requisite o-acylaminobenzylphosphonium salts (1) are available from either (i) o-nitrobenzylphosphonium salts² (3) via reduction with zinc in hydrobromic acid-ethanol $(R^1 = H; yield 95\%)$ (the reductive cleavage of the carbon-phosphorus bond³ can be avoided by using only 2 equiv. of reagent), or (ii) anthranilic alcohols (4) via condensation with triphenylphosphine hydrobromide ($R^1 =$ Ph; yield 76%). In the presence of 2 equiv. of triethylamine or pyridine, acyl chlorides (1.25 equiv.) react in toluene-dichloromethane (1:1) with the *o*-aminobenzylphosphonium salts formed by either of the above methods



to give 85-96% yields of the required salts (1) [reaction (2)].

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- ¹ W. Madelung, Chem. Ber., 1912, 25, 1128.
- A. Ardanaki, N. Maleki, and M. R. Saadein, J. Org. Chem., 1978, 43, 4128.
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