

Synthesis of 3 β -Methoxy-5 α ,14 α -card-20(22)-enolide from 3 β -Methoxy-5 α -androstan-17-one. A Method for the Construction of the Cardenolide Side-chain

By ALICJA KUREK, MARIA GUMUŁKA, and JERZY WICHA*

(Institute of Organic Chemistry of the Polish Academy of Sciences, 01-224 Warszawa, Kasprzaka 44, Poland)

Summary The cardenolide side-chain has been constructed on the androstane skeleton *via* Knoevenagel condensation of the 17-oxo-derivative (**2a**) with ethyl cyanoacetate, three-step transformation of the resultant product (**2b**) to the protected hydroxy-aldehyde (**4**), followed by the formation of the cyanohydrin (**5a**), hydrolytic butenolide ring closure, and dehydration.

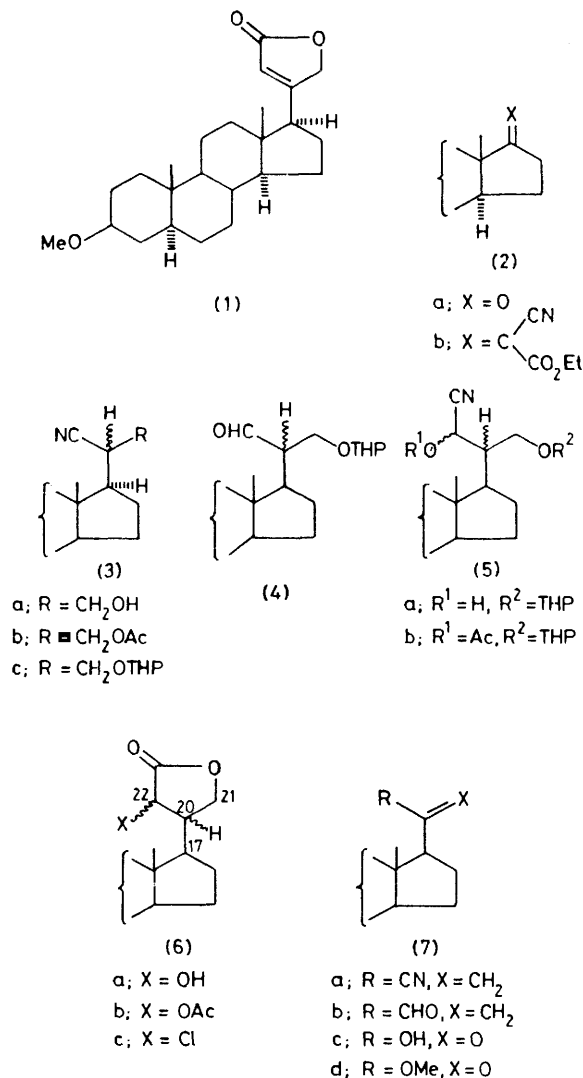
THE construction of the butenolide side-chain at the 17 β -position of the androstane skeleton is a major problem in the synthesis of cardenolides and attracts considerable attention.¹ All methods recently developed for this purpose,² with one exception,³ are based on 20-oxy-pregnane derivatives as starting materials. We now report a simple method for cardenolide synthesis starting from easily available 17-oxo-androstane derivatives, exemplified by the synthesis of compound (**1**).

The condensation⁴ of 3 β -methoxy-5 α -androstan-17-one (**2a**) with ethyl cyanoacetate gave the cyano-ester† (**2b**) as a mixture of *E*- and *Z*-isomers⁵ [96% yield; major isomer, m.p. 174–176 °C; λ_{\max} (EtOH) 238 nm, ϵ 12,300], which

was used for the next step without purification. Treatment of (**2b**) with an excess of sodium borohydride in ethanol-tetrahydrofuran resulted⁶ in saturation of the double bond, reduction of the ester group, and furnished the hydroxy-nitrile (**3a**) [97%; *m/e* 359 (M^+); δ (CDCl₃) 3.70 (2H, d, *J* 6 Hz, CH₂OH); ν_{\max} (KBr) 3500 and 2250 cm⁻¹] as a mixture of isomers.‡ The stereochemistry of this product was elucidated as follows. The mixture of alcohol isomers (**3a**) was acetylated (acetic anhydride-pyridine), and the acetates (**3b**) were heated in dimethylformamide (DMF) in the presence of lithium carbonate (150 °C; 6 h). Pyrolysis gave the homogeneous α,β -unsaturated nitrile (**7a**) [93%, m.p. 112–113 °C; *m/e* 341 (M^+); δ (CDCl₃) 6.02 and 5.78 (2H, 2s, C=CH₂); ν_{\max} (CCl₄) 2250 and 1600 cm⁻¹]. Oxidation of compound (**7a**) with potassium permanganate-tetrabutylammonium chloride-benzene⁷ afforded the known⁸ carboxylic acid (**7c**) [identified as the methyl ester (**7d**), m.p. 157–159 °C; *m/e* 348 (M^+)]. These experiments showed that (**3a**) is a mixture of compounds which are stereoisomeric only at C-20, and also that the side-chain at C-17 has the β -orientation.

† All new compounds obtained in homogenous form or as a mixture of stereoisomers had the expected spectral data and satisfactory combustion analysis or high-resolution mass spectra.

‡ No separation of isomers on t.l.c. plates was noted (hexane-ethyl acetate or benzene-acetone as eluant).



The cyano-alcohol (**3a**) was converted into the tetrahydro-pyranyl ether (**3c**), which upon treatment⁹ with 3.5 equiv. of di-isobutylaluminium hydride (hexane-toluene; -78°C ; 1 h), followed by standard work-up gave the aldehyde (**4**) [70–85% from (**3a**), m/e 446 (M^+); δ (CDCl_3) 9.80 (1H, m); ν_{max} (film) 1720 cm^{-1}]. On filtration of this product through a silica gel (MN 100–200 mesh ASTM) column a small amount (ca. 5%) of the α,β -unsaturated aldehyde (**7b**) [m.p. 123–126 $^\circ\text{C}$; m/e 344 (M^+)] was formed and separated off.

The crude aldehyde (**4**) was treated in methanol with hydrogen cyanide generated *in situ* from an excess of potassium cyanide and conc. hydrochloric acid (room temp.; 30 min). The unstable, diastereoisomeric cyanohydrins (**5a**) formed in this way could not be fully characterised; however, the corresponding acetates (**5b**) showed the expected properties [m/e 515 (M^+); δ (CDCl_3) 5.70 (1H, m, 22-H in diastereomeric compounds)].

At this stage of the synthesis the side-chain contains all the structural elements necessary for the hydrolytic ring closure to give the cardenolide ring. For this purpose the crude cyanohydrin (**5a**) was dissolved in ethanol containing hydrochloric acid, and the mixture was refluxed for 20 min. Structure (**6a**) was ascribed to the product, isolated in 76% yield, on the basis of its i.r. [ν_{max} (KBr) 3500 (OH) and 1785 (C=O) cm^{-1}] and n.m.r. spectra, and the spectral data of its acetate (**6b**) [ν_{max} (CHCl_3) 1785 and 1740 cm^{-1} ; δ (CDCl_3) 5.39 and 5.21 (1H, 2 d, J 10 Hz, 22-H)].

The hydroxy-lactone (**6a**) upon treatment with thionyl chloride in pyridine (reflux; 1 h) gave the chloride (**6c**), which was dehydrochlorinated (DMF; lithium carbonate; 150 $^\circ\text{C}$; 1.5 h) without purification. In this reaction the chiral centres at C-20 and C-22 were destroyed, and the butenolide (**1**) [m.p. 147–148 $^\circ\text{C}$; m/e 372 (M^+); δ (CDCl_3) 5.90 (1H, br. s), 4.88 (2H, s), 3.34 (3H, s), 3.10 (1H, m), 0.80 (3H, s), and 0.60 (3H, s); ν_{max} (KBr) 1790, 1750, and 1630 cm^{-1} ; λ_{max} (EtOH) 217 nm] was formed (90% from (**6a**)).

This work was supported by a grant from the Polish Academy of Sciences.

(Received, 1st August 1980; Com. 847.)

¹ For recent reviews see: M. B. Gorovitz and N. K. Abubakirov, *Khim. Prirod. Soedinienii*, 1978, 283; T. Thomas, J. Boutagy, and A. Gelbert, *J. Pharm. Sci.*, 1974, **63**, 1648.

² S. F. Donovan, M. A. Avery, and J. E. McMurry, *Tetrahedron Lett.*, 1979, 3287; G. R. Lenz and J. A. Schulz, *J. Org. Chem.*, 1978, **43**, 2334; E. Yoshii, T. Koizumi, H. Ikeshima, K. Ozaki, and I. Hayashi, *Chem. Pharm. Bull. Jpn.*, 1975, **23**, 2496.

³ T. Y. R. Tsai, A. Minta, and K. Wiesner, *Heterocycles*, 1979, **12**, 1397.

⁴ D. K. Patel, V. Petrow, R. Royer, and J. A. Stuart-Webb, *J. Chem. Soc.*, 1952, 161.

⁵ Cf. K. Annen, H. Hofmister, H. Laurent, A. Seeger, and R. Wicbert, *Chem. Ber.*, 1978, **111**, 3094.

⁶ J. A. Marshall and R. D. Carroll, *J. Org. Chem.*, 1965, **30**, 2748.

⁷ T. Ogino and K. Michizuki, *Chem. Lett.*, 1979, 443.

⁸ J. Hora, *Collect. Czech. Chem. Commun.*, 1966, **31**, 2737.

⁹ Y. Letoureneux, G. Bujuktur, M. T. Ryzlak, A. J. Banarjee, and M. Gut, *J. Org. Chem.*, 1976, **41**, 2288.