Asymmetric Synthesis of a New Proline Analogue

By ANDREA VASELLA* and ROBERT VOEFFRAY

(Institut de Chimie Organique, Université de Fribourg, Pérolles, CH-1705 Fribourg, Switzerland)

Summary The 1,3-dipolar cycloaddition to ethylene of N-glycosylnitrones, formed in situ from the partially protected D-mannose- or D-ribose-oximes and various glyoxalates, gave compounds which could be transformed into both enantiomers of 3-t-butoxycarbonyl-isoxazolidine and derivatives thereof.

ANALOGUES of proline possessing a heterocyclic ring other than pyrrolidine¹⁻³ are useful biochemical probes.⁴ We report the asymmetric synthesis of the two enantiomers of isoxazolidine-3-carboxylic acid ('5-oxaproline') and of some of their derivatives. The synthesis is based on the 1,3dipolar cycloaddition of N-glycosylnitrones^{5,6} to ethylene. When the partially protected D-mannose-, or D-riboseoximes (1)⁵ and (2)⁶ were allowed to react with 1.3 to 3 mol. equiv. of the glyoxylic esters[†] (3), (4), or (5), in the presence of ethylene (CHCl₃, 75 °C, 65 bar, 17 h), mixtures of the diastereomeric N-glycosyl-3(R,S)-alkoxycarbonylisoxazolidines (6)—(9) were formed in high chemical yields and with useful diastereoselectivities (Scheme 1 and Table).





TABLE. Yields and diastereoselectivities of the cycloadditions.

Glvoxvlic				Diastereo-	Major isomer
Oxime	ester	Product	Yield/%	selectivitya/%	formed
(1)	(3)	(6)	86	36	(3S)
(1)	(4)	(7)	92	54	(35)
(1)	(5)	(8)	93	54	(3S)
(2)	(5)	(9)	78	72	(3R)

^a The diastereoselectivities were determined by g.l.c. (capillary column Si 60) for compounds (6)—(8), by h.p.l.c. (Lichrosorb 60) for (8), and by weighing after chromatographic separation for (9).

The major diastereomer (8), shown below to possess the (3S) configuration, crystallized spontaneously from the reaction mixture and was obtained pure after 3—4 recrystallizations from hexane (43%, m.p. 96—97 °C, $[\alpha]_{D}^{25} = -38.8^{\circ}$ (c 0.98, CHCl₃).‡ Acid hydrolysis of (3S)-(8) (1 M HCl, 6% aq. MeOH, 40 °C, 4 h, then Na₂CO₃) gave the ester (10) (90%) m.p. 52—53 °C, $[\alpha]_{D}^{25} = -25.7^{\circ}$ (c 0.98, CHCl₃). δ 27.92(q), 34.87(t), 61.11(d), 69.36(t), 82.09(s), and 171.07(s) p.p.m., further characterized as its hydrochloride, m.p. 96—97 °C, $[\alpha]_{D}^{25} = -35.2^{\circ}$ (c 0.94, EtOH).



SCHEME 2. Reagents and conditions, i, H_2 , 5% Rh-C, EtOH; ii, Z-Cl, NaHCO₃, aq. EtOH; iii, KMnO₄, acetone, AcOH; iv, isobutoxycarbonyl chloride, N-methylmorpholine, tetrahydrofuran, -20 °C; v, NH₃, H₂O, -15 to 10 °C.

Compound (10) was transformed by standard methodology into the derivatives (11)—(15) (Scheme 2). The *N*-benzyloxycarbonyl derivative (12), $[\alpha]_D^{25} = -9\cdot3^\circ$ (c 1, 95% aq. AcOH) was obtained from the ester (11), m.p. 48.5—49 °C, $[\alpha]_D^{25} = -86\cdot3^\circ$ (c 1.1, CHCl₃) and also by immediate benzyloxycarbonylation of the acid (13), formed by treating the ester (10) with CF₃CO₂H. The ¹H n.m.r. spectrum (CDCl₃) of the urethane (11) is well resolved already at 90 MHz: $\delta 2\cdot59$ (2H, ddd, *J* 8, 7\cdot8, and 6 Hz), $3\cdot78(1H, q, J 8 Hz), 4\cdot14(1H, m), 4\cdot74(t, J 7\cdot8 Hz), 5\cdot25(2H,$ s), $5\cdot82(1H, br), 5\cdot55(1H, br),$ and $7\cdot38(5H, s)$. This should facilitate the conformational analysis of 5-oxaproline derivatives and hence of proline-containing peptides.

The absolute configuration of (10) was shown to be (S) by correlation with the known⁸ L-asparagine derivative (19) (Scheme 2). This involved a novel synthesis of the L-homoserine derivatives (16), $[\alpha]_{25}^{25} = +4\cdot9^{\circ}$ ($c \cdot 2\cdot04$, EtOH), ¹³C n.m.r. (CDCl₃): $\delta \cdot 27\cdot98(q)$, $35\cdot56(t)$, $54\cdot73(d)$, $61\cdot41(t)$, $81\cdot25(s)$, and $174\cdot41(s)$ p.p.m., and (17), m.p. 78—79 °C, $[\alpha]_{25}^{25} = -12\cdot70$ ($c \cdot 2$, EtOH), ¹³C n.m.r. (CDCl₃): $\delta \cdot 27\cdot92(q)$, $37\cdot98(t)$, $51\cdot6(d)$, $58\cdot32(t)$, $67\cdot11(t)$, $82\cdot33(s)$, $127\cdot97(d)$,

† Ester (3) was prepared by periodate cleavage of tartaric acid dibenzyl ester (T. S. Patterson, J. Chem. Soc., 1913, 103, 176) (80% yield); (4) and (5) were prepared by ozonolysis (J. J. Pappas, W. P. Keaveney, E. Gancher, and M. Berger, *Tetrahedron Lett.*, 1966, 4273) of the corresponding maleic acid esters (83-85% yield).

 \ddagger All new compounds were characterized either by elemental analysis and spectroscopic (i.r. and 90 MHz ¹H and 25 MHz ¹³C n.m.r.) data or by chemical correlation with thus-characterized compounds.

128.06(d), 128.36(d), 135.95(s), 156.69(s), and 171.38(s) p.p.m.

Access to the compounds with (R)-configuration was gained via the addition products (9). The major isomer, (3R)-(9), was isolated by silica gel chromatography. Detritylation⁵ (FeCl₃ in 1% MeOH-CH₂Cl₂) followed by hydrolysis (HCl, aq. MeOH) gave the enantiomer of (10) in 43% yield from (2).

5-Oxaproline (Opro) is easily incorporated in synthetic peptides. Using O- or N-protected derivatives (10), (12), and (14), a few protected model dipeptides containing either C-terminal or N-terminal Opro have been synthesized. In all Z-protected 5-oxaproline derivatives, the hydrogenolytic cleavage of the N-benzyloxycarbonyl group was accompanied by cleavage of the N-O bond. However, in 5-oxaproline derivatives possessing an N-acyl group which is stable to hydrogenolytic conditions, such as N-acetyl or N-peptidyl 5-oxaprolines, the N-O bond is, as expected,⁹ not affected by the conditions of Pd-catalysed hydrogenolysis. Thus, preliminary experiments have shown that under hydrogenolytic conditions (H2, 10% Pd-C, EtOAc) Ac-Opro-OBzl cleanly leads to Ac-Opro-OH.

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