

General Synthesis of Imidazole C-Nucleosides From Carbohydrate Adducts of Diaminomaleonitrile

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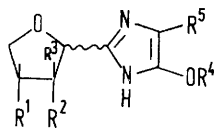
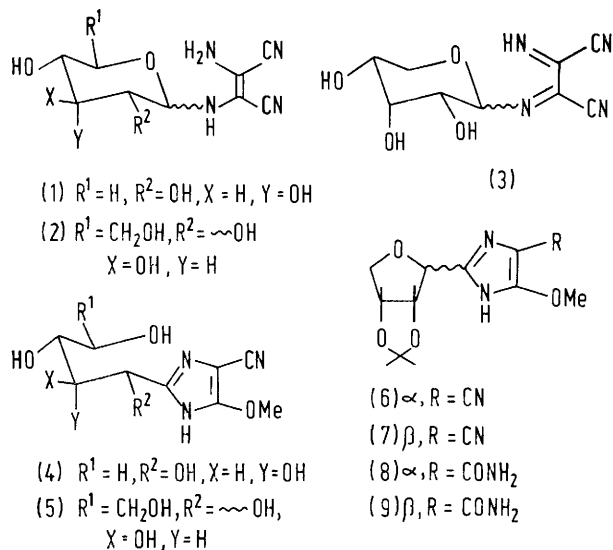
Summary The preparation of the imidazole C-nucleosides (**14**), (**15**), (**20**), and (**21**) from the adducts of ribose, arabinose, glucose, and mannose with diaminomaleonitrile is described.

RIBOSYL derivatives of diaminomaleonitrile are useful intermediates for the synthesis of imidazole and triazole ribopyranosides.¹ We describe herein an efficient and simple synthesis of C-nucleoside analogues of the antitumour and antiviral agents pyrazomycin (pyrazofurin) (**23**)² and bredinin (**22**)³ using readily available carbohydrates and diaminomaleonitrile as the starting materials. The novel

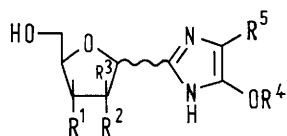
oxidative cyclization of a carbohydrate-diaminomaleonitrile adduct to an imidazole is a key step in the synthesis. The configuration at each asymmetric centre and the number of carbon atoms in the C-nucleoside can easily be varied using this synthetic approach.

Reaction of amino(ribosylamino)maleonitrile (**1**)¹ (52 mmol) with 2,3-dichloro-5,6-dicyanobenzoquinone (DDQ) (54.5 mmol) in 325 ml of MeOH for 1.5 h at room temperature resulted in the formation of 4-cyano-5-methoxy-2-(*D*-ribo-tetrahydroxybutyl)imidazole (**4**) (64%) [m.p. 179–180 °C (decomp.); u.v. λ_{max} (MeOH) 246 nm (ϵ 12,900); i.r. (Nujol) 2212 (CN) and 1567 (C=C, C=N) cm^{-1} ;

^1H n.m.r. [$(\text{CD}_3)_2\text{SO}$] δ 5.68 (d, 1H, J 4 Hz, H-1') and 3.90 (s, 3H, OMe); mass spectrum m/e 243 (M^+).† The imino-(ribosylimino)succinonitrile adduct (3) is probably the initial oxidation product⁴ which then adds methanol⁵ to give an imidate ester that cyclizes to the imidazole (4). The oxidative cyclizations of the *o*-phenylenediamine adducts of sugars to benzimidazoles have been reported.⁶



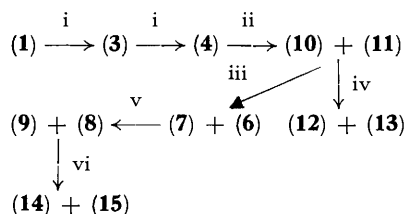
- (10) $\alpha, R^1 = R^2 = \text{OH}, R^3 = \text{H}, R^4 = \text{Me}, R^5 = \text{CN}$
 (11) $\beta, R^1 = R^2 = \text{OH}, R^3 = \text{H}, R^4 = \text{Me}, R^5 = \text{CN}$
 (12) $\alpha, R^1 = R^2 = \text{O}_2\text{CMe}, R^3 = \text{H}, R^4 = \text{Me}, R^5 = \text{CN}$
 (13) $\beta, R^1 = R^2 = \text{O}_2\text{CMe}, R^3 = \text{H}, R^4 = \text{Me}, R^5 = \text{CN}$
 (14) $\alpha, R^1 = R^2 = \text{OH}, R^3 = \text{R}^4 = \text{H}, R^5 = \text{CONH}_2$
 (15) $\beta, R^1 = R^2 = \text{OH}, R^3 = \text{R}^4 = \text{H}, R^5 = \text{CONH}_2$



- (16) $\alpha, R^1 = R^3 = \text{OH}, R^2 = \text{H}, R^4 = \text{Me}, R^5 = \text{CN}$
 (17) $\beta, R^1 = R^3 = \text{OH}, R^2 = \text{H}, R^4 = \text{Me}, R^5 = \text{CN}$
 (18) $\alpha, R^1 = R^3 = \text{OH}, R^2 = \text{H}, R^4 = \text{Me}, R^5 = \text{CONH}_2$
 (19) $\beta, R^1 = R^3 = \text{OH}, R^2 = \text{H}, R^4 = \text{Me}, R^5 = \text{CONH}_2$
 (20) $\alpha, R^1 = R^3 = \text{OH}, R^2 = \text{R}^4 = \text{H}, R^5 = \text{CONH}_2$
 (21) $\beta, R^1 = R^3 = \text{OH}, R^2 = \text{R}^4 = \text{H}, R^5 = \text{CONH}_2$

An anomeric mixture of *D*-erythrofuransylimidazoles (10) and (11) was obtained in 77% yield by heating 7.5 mmol of (4) for 3 h at 150 °C in a mixture of 15 ml of 2-ethoxyethanol and 0.1 ml of acetic acid.⁷ This mixture was converted into the corresponding isopropylidene acetals which were separated by chromatography on silica gel to give the β -anomer (7) in 62% yield [m.p. 110–111 °C; u.v. λ_{max} (MeOH) 245 nm (ϵ 14,070); mass spectrum m/e 265 (M^+)]† and the α -anomer (6) in 21% yield [m.p.

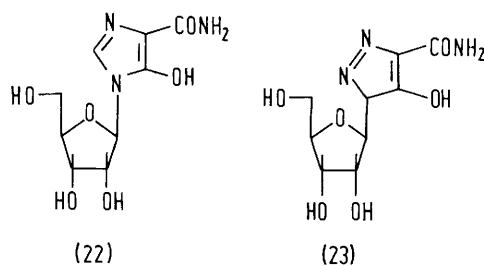
75–83 °C; u.v. λ_{max} 244 nm (ϵ 16,200); mass spectrum m/e 265 (M^+)]†. Treatment of the mixture of (10) and (11) with acetic anhydride–pyridine yielded the diacetates (12) (13%) and (13) (46%) which were separated by h.p.l.c. on silica gel. The diacetates (12) and (13) were also obtained when amino(arabinosylamino)maleonitrile was the starting point in the synthesis, a result consistent with the loss of stereochemistry at C-1' during the acid-catalysed cyclization of (4) and the corresponding *arabino*-tetrahydroxybutyl derivative. Hydrolysis of (6) and (7) with $\text{KOH}\text{-Bu}^t\text{OH}$ ¹⁴ gave the amides (8) [35%; m.p. 104–106 °C (amorphous); i.r. 1640 cm^{-1} (CONH_2); mass spectrum m/e 283 (M^+)]† and (9) [52%; m.p. 230–232 °C; i.r. 1640 cm^{-1} (CONH_2); mass spectrum m/e 283 (M^+)] (Scheme 1).



SCHEME 1. Reagents: i, (1) (52 mmol), DDQ (54.5 mmol), MeOH (325 ml), 1.5 h, r.t.; ii, (3) (7.5 mmol), $\text{EtO}[\text{CH}_2]_2\text{OH}$ (15 ml)–AcOH (0.1 ml), 150 °C, 3 h; iii, $\text{Me}_3\text{C}(\text{OMe})_2$; iv, acetic anhydride–pyridine; v, $\text{KOH}\text{-Bu}^t\text{OH}$; vi, Me_3SiI .

The anomeric configurations of compounds (6)–(9) were assigned from their ^{13}C n.m.r. spectra using the following criteria. (i) The isomer with the *trans*-relationship between the aglycone and the C-2' hydroxyl group (β -anomer in the erythrofuransose series) has C-1' at lower field than the corresponding *cis*-isomer.⁸ (ii) The difference in chemical shifts between the gem-dimethyl groups of the acetal is greater in the β -anomers than in the α -anomers.⁹ The chemical shift of C-1' in (7) is δ 83.91 and the $\Delta\delta$ (Me) is 1.63 p.p.m. while the corresponding values for (6) are 81.24 and $\Delta\delta$ (Me) 1.33 p.p.m., consistent with the assignment of (7) as the β -anomer and (6) as the α -anomer. Similar ^{13}C n.m.r. spectral data lead to the assignment of the β -configuration to (9) [C-1' δ 83.57, $\Delta\delta$ (Me) 1.68 p.p.m.] and the α -configuration to (8) [C-1' δ 81.28, $\Delta\delta$ (Me) 1.13 p.p.m.].

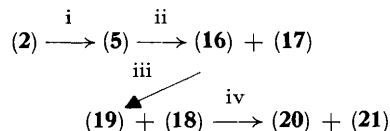
Both the isopropylidene and methyl ether groups of (8) and (9) were cleaved with trimethylsilyl iodide¹⁰ to give (14) (47%) and (15) (46%), respectively; these are erythrofuransyl C-nucleoside analogues of bredinin (22) and pyrazomycin (23).^{2,3} The structures of (14) and (15) were established from their characteristic u.v. [λ_{max} 240 nm



† The i.r., u.v., and n.m.r. spectra and combustion data were consistent with the assigned structures.

(ϵ 5500) and 280 nm (ϵ 12,000)] and mass spectra [m/e 229, (M^+)]. The ^1H n.m.r. spectrum of (15) is very similar to that of 4-carboxamido-5-hydroxy-2-(β -D-ribofuranosyl)-imidazole and the two compounds exhibit almost identical chemical shifts and $J_{1,2}$ values for H-1'.¹¹

The synthesis of the corresponding arabinofuranosyl-imidazole was initiated by the reaction of diaminomaleonitrile (0.05 mol) and either D-glucose¹² or D-mannose in a mixture of methanol and acetic acid (75:10) to give a 75% yield of (2).[†] The DDQ (0.01 mol) oxidation of 0.01 mol of (2) in 75 ml MeOH gave a 52% yield of (5) [u.v. λ_{max} (MeOH) 246 nm (ϵ ca. 15,000)].[†] The structure of (5), the oxidation product of glucosyldiaminomaleonitrile, was supported by its conversion into the penta-O-acetyl derivative by acetic anhydride and pyridine [m.p. 159–160 °C; mass spectrum m/e 483 (M^+)].[†] Heating (5) in a mixture of 30 ml of 2-ethoxyethanol and 10 ml of acetic acid for 12 h at 150 °C resulted in the formation of (16) and (17). The β -anomer, (17), was obtained in 38% yield by a combination of fractional crystallization followed by chromatography [m.p. 215–216 °C; u.v. λ_{max} (MeOH) 243 nm (ϵ 14,920); mass spectrum m/e 225 (M^+)].[†] The α -anomer (16) (13%) was purified by column chromatography [m.p. 94–97 °C (amorphous); u.v. λ_{max} (MeOH) 245 nm (ϵ 13,280)]. The β -configuration was assigned to (17) [^1H n.m.r. δ 4.66 (s, H-1); ^{13}C n.m.r. δ 71.72 p.p.m. (C-1')] and the α -configuration to (16) [^1H n.m.r. δ 4.2–3.8 (H-1' overlaps with the signals from the other protons); ^{13}C n.m.r. δ 76.48 p.p.m. (C-1')].^{8,13} The nitriles (16) and (17) were hydrolysed¹⁴ to the corresponding amides (18) (80%) [m.p. 172–174 °C (decomp.); u.v. λ_{max} (MeOH) 259.5 nm (ϵ 14,140); ^1H n.m.r. δ 4.2–3.5 (H-1' overlaps with signals from the other protons); ^{13}C



SCHEME 2. Reagents: i, (2) (0.01 mol), DDQ (0.01 mol), MeOH (75 ml); ii, EtO[CH₂]₂OH (30 ml)–AcOH (10 ml), 150 °C, 12 h; iii, KOH; iv, Me₃SiI.

n.m.r. δ 71.13 p.p.m. (C-1'); mass spectrum m/e 273 (M^+); hygroscopic, no combustion analyses][†] and (19) (45%) [m.p. 243–244 °C (decomp.); u.v. λ_{max} (MeOH) 259 nm (ϵ 16,645); ^1H n.m.r. δ 4.62 (s, H-1'); ^{13}C n.m.r. δ 76.48 p.p.m. (C-1'); mass spectrum m/e 273 (M^+)].[†] Reaction with trimethylsilyl iodide gave a 52% yield of (20) [m.p. 235 °C (decomp.); u.v. λ_{max} (H₂O) 280.5 nm (ϵ 14,910) and 240 nm (ϵ 6507); ^1H n.m.r. δ 4.1–3.5 (H-1' overlaps with the signals from the other protons); mass spectrum m/e 259 (M^+)][†] and a 42% yield of (21) [m.p. 225 °C (decomp.); u.v. λ_{max} (H₂O) 281 nm (ϵ 16,000) and 240 nm (ϵ 6865); ^1H n.m.r. δ 4.67 (s, H-1'); mass spectrum m/e 259 (M^+)].[†] It was surprising and gratifying that the primary hydroxy-groups in (18) and (19) were not converted into the corresponding iodides with trimethylsilyl iodide.¹⁵

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