A Base-labile Amine Component in Four-component Condensation (4CC) Synthesis

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Summary The use of 9-aminomethylfluorene as the amine component in four-component condensation (4CC) synthesis is described for a model peptide synthesis.

THE advantage in utilizing 9-formylfluorene as the aldehyde component in four-component condensation fragmentstrategy peptide synthesis (4CC fragment condensation) has been demonstrated.¹ The base lability of 9-(heteroatom methyl)-substituted fluorenes² offers the use of 9-aminomethylfluorene as the amine component in the one-step construction of tripeptide derivatives (4CC synthesis).^{3,4} We report herein our results in the application of this amine to 4CC synthesis of a model dipeptide.

9-Aminomethylfluorene (2) was prepared in several steps from fluorene-9-carboxylic acid. The commercially available acid by reaction in neat oxalyl chloride provided fluorene-9-carbonyl chloride.⁵ The acid chloride was dissolved in dioxan and treated with anhydrous ammonia to yield 9-fluorenecarboxamide.⁵ Reduction of the amide was accomplished with 10 equiv. of sodium trifluoroacetoxyborohydride⁶ in dioxan at 60 °C for 5 h. The amine product⁷ was isolated and spectroscopically characterized as the hydrochloride salt. 9-Aminomethylfluorene hydrochloride (2), m.p. 261–265 °C (decomp.); ¹H n.m.r. (CD₃SOCD₃-D₂O) δ 3·30 (2 H, d, J 6 Hz, CH₂), 4·27 (1 H, t, J 6 Hz, CH), and 7·17–7·86 (8 H, m, aryl), was obtained from fluorene-9-carboxylic acid in 33% overall yield.

Reaction of 0.40 mmol of (2) with equimolar quantities of N-benzyloxycarbonylglycine (1), benzaldehyde (3), and cyclohexyl isocyanide (4) was performed in 3.0 ml of methanol containing 1 equiv. of triethylamine at room temperature for a period of 16 h. Work-up as previously described¹ yielded compound (5),† recrystallized from diethyl ether (76%), m.p. 154—155 °C; δ (CDCl₃) 0.91— 2.10 (11 H, m alkyl CH), 3.14—3.81 (5 H, m, NCH₂CO and NCH₂CH), 4.93 (2 H, s, CH₂OCO), 5.78 (1 H, br.s, CH), and 6.98—7.67 (18 H, m, aryl CH). The condensation product was homogeneous as indicated by t.l.c. on Merck $60F_{254}$ silica gel with 1% methanol in chloroform ($R_{\rm F}$ 0.38) or ethyl acetate ($R_{\rm F}$ 0.75) as eluant.

Removal of the N-fluoren-9-ylmethyl substituent was accomplished in excellent yield by reaction with 1,8diazabicyclo[5.4.0]undec-7-ene (DBU). Treatment of 0·125 mmol of (5) with 1·1 equiv. of DBU in 0·35 ml of pyridine at room temperature effected removal of the substituent group in 6 h. Reaction progress was followed by t.l.c. The reaction mixture was diluted with 25 ml of ethyl acetate and then washed with equal volumes of 0·5 N aqueous hydrochloric acid and distilled water to yield the dipeptide (6),† recrystallized from chloroform (87%), m.p. 209-211 °C; δ (CDCl₃-CD₃SOCD₃) 1·07-2·23 (11 H, m, alkyl CH), 3·83 (2 H, d, J 6 Hz, CH₂), 5·06 (2 H, s, CH₂OCO), 5·51 (1 H, d, J 8 Hz, CH), and 7·27 (10 H, s, aryl)]. The peptide amide was homogeneous by





t.l.c. on Merck $60F_{254}$ silica gel with 1% methanol in chloroform ($R_F 0.17$) or ethyl acetate ($R_F 0.62$) as eluant. Dibenzofulvene (7), the cleavage reaction by-product, was isolated as a white powder (79%), δ (CDCl₃) 6.10 (2 H, s, =CH₂) and 7.23-7.87 (8 H, m, aryl)].

Other methods for removal of the N-fluoren-9-ylmethyl substituent proved less satisfactory. Treatment of (5) ammonia-saturated methanol,¹ triethylaminewith pyridine, or morpholine at room temperature for periods of 16 h did not result in substituent removal. Removal was possible by reaction of (5) in morpholine at 107 °C for 3.5 h to yield the dipeptide (6) (83%) and the dibenzofulvene-morpholine adduct (8) (81%), δ (CDCl₃) 2.60 (4 H, m, $[CH_2]_2N$), 2.63 (2 H, d, J 8 Hz, $CHCH_2N$), 3.80 (4 H, m, $[CH_2]_2O$), 4.03 (1 H, t, J 8 Hz, CH), and 7.17-7.80 (8H, m, aryl). Treatment of (5) with a catalytic amount of DBU (0.2 equiv.) in pyridine at room temperature for 72 h effected incomplete cleavage (ca. 90%).

The difficulties in cleaving the amide N-substituent can be overcome by utilizing 9-aminomethylfluorene (2) as the amine component in 4CC synthesis. The auxiliary group obtained in this approach can be efficiently removed under mild basic conditions. This procedure offers a means of cleaving the N-auxiliary group in 4CC synthesis which is complementary to methods required for the removal of common peptide blocking groups, *i.e.* benzyloxycarbonyl-, t-butyloxycarbonyl-, t-butyl ester, and 2-(biphenyl-4-yl)isopropyloxycarbonyl-groups.

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