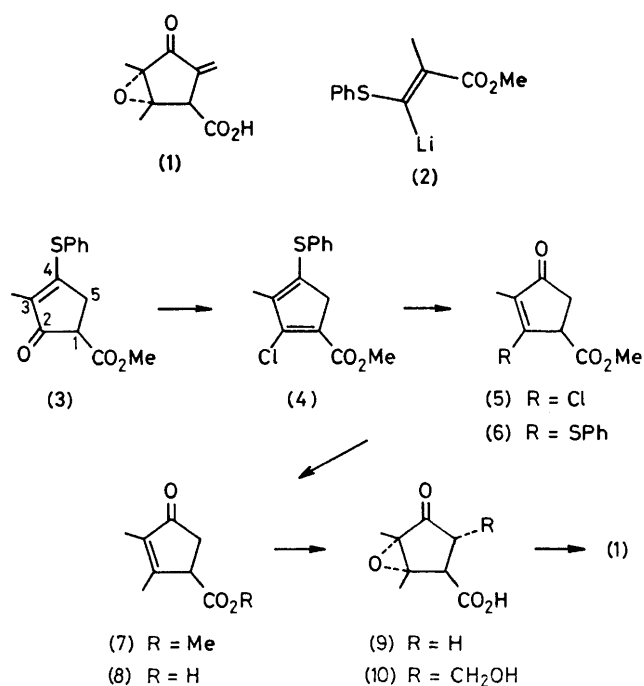


## A Total Synthesis of ( $\pm$ )-Methylenomycin A

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*Summary* A new synthesis of ( $\pm$ )-methylenomycin A (**1**) has been accomplished by applying the cyclisation reaction between the vinyl-lithium reagent (**2**) and methyl acrylate as the key step.

In connection with our synthetic studies using the vinyl-lithium reagents generated from 3-heteroatom-functionalised propenoic acid derivatives,<sup>1</sup> we found that in the reaction with  $\alpha\beta$ -enoates the vinyl-lithium reagent (2) underwent Michael addition and subsequent Dieckmann cyclisation to afford functionalised cyclopentenone derivatives in one step.<sup>1a</sup> We report here a total synthesis of ( $\pm$ )-methyl-enomycin A,<sup>2,3</sup> one of the cyclopentanoid antibiotics produced by *Streptomyces violaceoluber*, by a route which involves the new cyclisation reaction as the key skeleton-forming step (Scheme).



SCHEME

The key intermediate (3), m.p. 69—71.5 °C, was prepared in 83% yield (based on methyl acrylate) or 42% from (2) from the reaction between the vinyl-lithium reagent (2), generated from methyl (*E*)-2-methyl-3-phenylthioprop-2-enoate by the action of lithium di-isopropylamide (LDA), and methyl acrylate according to the procedure previously

reported<sup>1a</sup> except that we used a different ratio of (2) to methyl acrylate.†

The ketone transposition from C-2 in (3) to C-4 and the introduction of a methyl group at C-2 were achieved as follows.‡ Treatment of (3) with 3 mol. equiv. of PCl<sub>3</sub> in refluxing chloroform (48 h) yielded the diene chloride (4),§ m.p. 62—63.5 °C, in 84% yield. Compound (4), (8 mmol) was hydrolysed using TiCl<sub>4</sub> (2 mol. equiv., 0 °C for 0.5 h) in CH<sub>2</sub>Cl<sub>2</sub>-acetic acid (1:1; 150 ml) and then by addition of H<sub>2</sub>O (4 mol. equiv., 16 h at room temperature)<sup>4</sup> to give the chloride (5) (76% yield) together with a small amount of the rearranged sulphide (6);¶ for (5): <sup>1</sup>H n.m.r. δ (CCl<sub>4</sub>) 1.80 (3 H, d, *J* 2.0 Hz), 2.67 (2 H, d, *J* 6.0 Hz), and 3.77 (OMe overlapped with 1 H multiplet); for (6): δ (CCl<sub>4</sub>) 1.80 (3 H, d, *J* 2.0 Hz), 2.18 (1 H, dd, *J* 18.0 and 1.5 Hz), 2.68 (1 H, dd, *J* 18.0 and 3.5 Hz), 3.40 (OMe), 3.67 (1 H, m), and 7.39 (5 H). The formation of the sulphide (6) was confirmed by the fact that (5), when treated with PhSH in a mixture of CH<sub>2</sub>Cl<sub>2</sub> and acetic acid containing TiCl<sub>4</sub>, afforded (6) in high yield.

Both (5) and (6) were then treated with Me<sub>2</sub>CuLi (−65 °C, 2 h) to give the same methylated enone (7) in 60% and 70% yield, respectively; <sup>1</sup>H n.m.r. δ (CCl<sub>4</sub>) 1.70 (3 H, br. s), 2.03 (3 H, br. s), 2.50 (2 H, d, *J* 6.0 Hz), 3.57 (1 H, m), and 3.70 (OMe); i.r. ν<sub>max</sub> 1740, 1710, and 1650 cm<sup>−1</sup>. Saponification of (7) under alkaline conditions (2 mol. equiv. of NaOH, aq. MeOH, 1 h at 0 °C) gave the free acid (8) [81% yield, m.p. 82.5—84.5 °C (decomp.) (lit.,<sup>3b</sup> 81—83 °C)]. The acid (8) was converted upon treatment with 30% H<sub>2</sub>O<sub>2</sub> under alkaline conditions into the *trans* epoxy-acid (9) [m.p. 125.5—126.5 °C (lit.,<sup>3b</sup> 125.5—127 °C)] in 81% yield. The final reactions for the introduction of an  $\alpha$ -methylene unit were carried out in a similar fashion to that described by Jernow *et al.*<sup>3b</sup> Treatment of (9) with LDA in tetrahydrofuran followed by addition of ethereal formaldehyde gave (10), dehydration of which gave ( $\pm$ )-methyl-enomycin A which was purified by column chromatography [40% from (8)] and was identical spectroscopically with the natural compound.

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† The use of an excess of (2) with respect to methyl acrylate (2:1 mol. ratio) is essential for optimising the yield because the resulting  $\beta$ -keto-ester in the reaction consumes the starting vinyl-lithium (2) to form the corresponding lithium enolate.

‡ Direct introduction of a methyl group on C-2 in (3) using methylmagnesium bromide or methyl-lithium was unsuccessful, resulting only in the formation of the enolate.

§ Satisfactory analytical and spectroscopic data have been obtained for all new compounds.

¶ The yield of (6) depends on the reaction conditions; higher concentrations of (4) (40 mmol/350 ml) increased the yield of (6) to 27%, with a yield of (5) of 50%.

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<sup>2</sup> For the structural elucidation of (1) see T. Haneishi, N. Kitahara, T. Takiguchi, M. Arai, and S. Sugawara, *J. Antibiot.*, 1974, **27**, 386; T. Haneishi, A. Terahara, T. Hata, and T. Tamura, *ibid.*, p. 393.

<sup>3</sup> For syntheses of (1) see (a) R. M. Scarborough and A. B. Smith, III, *J. Am. Chem. Soc.*, 1977, **99**, 7085; (b) J. Jernow, W. Tautz, P. Rosen, and J. F. Blout, *J. Org. Chem.*, 1979, **44**, 4210; (c) M. Koreeda, Y. P. Liang Chen, and H. Akagi, 178th Natl. Am. Chem. Soc. Meeting, Washington, D.C., U.S.A., 1979 and personal communication from Prof. M. Koreeda; (d) R. M. Scarborough, B. H. Toder, and A. B. Smith, III, *J. Am. Chem. Soc.*, 1980, **102**, 3904.

<sup>4</sup> T. Mukaiyama, S. Kamio, S. Kobayashi, and H. Takei, *Bull. Chem. Soc. Jpn.*, 1972, **45**, 3723.