## Stereospecific Intramolecular Addition of α-Lithio Sulphoxides or Sulphones to Inactivated Double Bonds

By VANDA CERÉ, CLAUDIO PAOLUCCI, SALVATORE POLLICINO, EDDA SANDRI,\* and ANTONINO FAVA\* (Istituto di Chimica Organica, Università di Bologna, Viale Risorgimento 4, 40136 Bologna, Italy)

Summary Treatment of the 1-oxides or 1,1-dioxides of trans homoallylic 8—10 membered ring cyclic sulphides with BuLi in tetrahydrofuran results in transannular addition of the  $\alpha$ -thio carbanion to the *E* double bond with formation of bicyclic products.

NUCLEOPHILIC additions to inactivated double bonds are rare.<sup>1</sup> We report the facile intramolecular addition of  $\alpha$ -sulphinyl or -sulphonyl carbanions (lithio derivatives) to inactivated double bonds. The substrates were l-oxides or 1, l-dioxides of cyclic *n*-membered (n = 8—10) homoallylic sulphides of the *E*-configuration.<sup>2</sup> Their structure is such that carbon atom *n*,  $\alpha$  to sulphur, faces the transannular double bond.<sup>3</sup> When treated with BuLi in tetrahydrofuran (THF),  $\alpha$ -lithio derivatives are formed which rapidly and quantitatively add to the double bond on the opposite side of the ring resulting in bicyclic products.<sup>†</sup> Under the same conditions the corresponding *Z*-isomers were inert. Thus,



† Satisfactory analytical and spectral data (<sup>1</sup>H and <sup>13</sup>C n.m.r.; m.s.) were obtained for all separated products.

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from the l-oxide<sup> $\ddagger$ </sup> of (E)-3,3-dimethylthiacyclo-oct-4-ene<sup>5</sup> (1), a single product was formed, exo-4,4-dimethyl-2thiabicyclo[3.3.0]octane 2-oxide (2), which implies attack of C-8 on the C-4 end of the double bond. The exo structure was established by comparison of the <sup>13</sup>C n.m.r. spectrum of (2) with that of (3), the isomeric sulphoxide obtained from (2) by sulphur inversion<sup>6</sup> via alkaline hydrolysis of the ethoxysulphonium derivative.§



In an analogous reaction E-thiacyclodec-4-ene 1,1-dioxide gave a bicyclic product, identified as trans-l-thiadecalin 1,1dioxide (m.p. 115-116 °C) by comparison with an authentic sample prepared by oxidation of trans-1-thiadecalin.<sup>7</sup> In

this case attack of C-10 occurs on the C-5 end of the double bond.



The reaction of the analogous 9-membered sulphone (4) also yields products via intramolecular attack at C-5; three major products (all m/e 174) were formed, two of which are isomeric cis- and trans-2-thiabicyclo[4.3.0]nonane 2,2-dioxides (5) and (6). The third appears to be a ring-contracted methyl sulphone (7) which probably arises by E1cb elimination from the carbanion initially formed by intramolecular addition at C-5, as depicted in the Scheme.

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 $\ddagger$  Oxidation (NaIO<sub>4</sub>) of racemic *E*-thiacyclo-oct-4-enes yields a single sulphoxide.<sup>4</sup> Models show that for steric reasons only the sulphoxide may be formed where the oxygen atom projects outside. This is the (1*R*, 4*R*; 1*S*, 4*S*)-diastereoisomeric mixture.

 $\frac{13}{12}$  N.m.r.:  $\delta$  (p.p.m. from Me<sub>4</sub>Si), (2) 75·2 (d), 62·2 (t), 59·2 (d), 46·3 (s), 30·5 (q), 29·2 (t), 28·6 (t), 26·7 (t), and 25·4 (q); (3) 64·4 (t), 61·2 (d), 59·6 (d), 40·3 (s), 30·6 (q), 29·8 (t), 27·3 (t), 24·9 (q), and 21·8 (t). Particularly revealing are the upfield shifts (exo  $\rightarrow$  endo) of C-4 (46·3  $\rightarrow$  40·3) and C-8 (minimum 26·7 to 21·7).

Assignment of structure (7) is based on n.m.r. evidence which indicates the presence of a monosubstituted ethylene unit [ $^{13}C$ ,  $\delta$ 139.8 (d) and 115.9 (t) p.p.m.; <sup>1</sup>H, 5 6.07–4.87 (3H, m)] and a methyl sulphonyl group [<sup>13</sup>C, 5 40.1 (q) p.p.m.; <sup>1</sup>H, 5 2.71 (3H, s)].

<sup>1</sup> W. Ammann and C. Ganter, *Helv. Chim. Acta*, 1977, **60**, 1924; R. A. Pfund and C. Ganter, *ibid.*, 1979, **62**, 228; R. A. Pfund, W. B. Schweizer, and C. Ganter, *ibid.*, 1980, **63**, 674; C. A. Grob and H. Katayama, *ibid.*, 1977, **60**, 1890. <sup>2</sup> E. Vedejs and J. P. Hagen, *J. Am. Chem. Soc.*, 1975, **97**, 6878; V. Ceré, C. Paolucci, S. Pollicino, E. Sandri, and A. Fava, *J. Org.*