Preparation, Purification, and X-Ray Crystal Structure of a Dimeric Manganese(IV) Porphyrin Complex

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Summary The highly reactive dimeric complex μ -oxobis[azidotetraphenylporphyrinatomanganese(IV)] has been prepared, purified, characterized by X-ray crystallography and other physical methods, and some of its reactions examined.

MUCH recent research activity has been directed to the investigation of manganese(IV) porphyrin complexes owing to their ability thermally¹ and photochemically² to oxidize water and to oxidize a variety of other substrates.³ Despite the importance of these complexes and the rapidly expanding literature dealing with their preparation and characterization, little or nothing is known with certainty regarding their structure.⁴ The lack of structural characterization of manganese(IV) porphyrin complexes is a consequence of the fact that all these species prepared thus far are highly reactive at 25 °C rendering their purification very difficult. We report here the first purification and X-ray crystal structure determination of a managanese(IV) porphyrin complex.

A solution of manganese(III) tetraphenylporphyrin (TPP) azide⁵ (7·1 \times 10⁻⁴ mol) in 25 ml of rigorously purified chlorobenzene† was vigorously stirred under nitrogen with iodosylbenzene (3 equiv.) for 4 min, and then the mixture was filtered into 350 ml of deolefinated hexane at -78 °C. The amorphous brown impure precipitated product was recrystallized to yield large lustrous black diffraction-quality crystals of a chlorobenzene solvate dimer [{N₃Mn^{IV} (TPP)}₂O]·3PhCl, by layering a 21 mM chlorobenzene solution of the impure dimer with heptane at 25 °C. A crystal of the dimeric complex, a non-stoicheiometric

† The optimal preparations and manipulations of the manganese(IV) porphyrins require the use of scrupulously purified solvents and anaerobic reaction conditions.

chlorobenzene solvate, was selected for structural analysis by X-ray crystallography.

Crystal data: $[\{N_{a}Mn^{IV}(TPP)\}_{o}O] \cdot nPhCl, n = ca. 1.5,$ M = 1604.2, orthorhombic, space group *Pbcn*; a = 21.208(5), b = 16.826(4), c = 22.620(3) Å, U = 8072(4) Å³, $D_c =$ 1.320, $D_{\rm m}=1.36(2)~{\rm g/cm^3}$. 3276 Unique reflections with $F_{0^{2}} > 3\sigma(F_{0^{2}})$ were measured using Mo- K_{α} radiation on an Enraf-Nonius CAD-4 automatic diffractometer. The structure was solved by the heavy-atom method and refined by least-squares to a final R factor of 0.094.

The crystal structure consists of well separated molecules of the $[\{N_3Mn^{IV}(TPP)\}_2O]$ dimer with chlorobenzene mole-



FIGURE. ORTEP⁶ plot of {N₃Mn^{IV}(TPP)}₂O] viewed from the side, perpendicular to the two-fold axis. Atoms are represented by 50% probability ellipsoids. The terminal azide ligands are two-fold disordered in the crystal.

cules partially occupying the holes resulting from the packing of the dimers. Each dimer possesses rigorous crystallographic C_2 symmetry, with both Mn atoms, the bridging oxygen atom, and the ligating nitrogen atoms of each N_a ligand lying on the two-fold axis parallel to b (see Figure). The terminal azide ligands are two-fold disordered in the crystal as is apparent from the Figure, the two Mn atoms are displaced from the mean N_4 planes of the two TPP ligands toward the bridging oxygen by 0.10 and 0.08 Å, respectively, and the N_4 plane shows some S_4 ruffling. Important distances are as follows: Mn-N(TPP) (average of the four unique distances) 2.014(19) Å; Mn(1)-N(1) (azide) 1.998(8); Mn(2)-N(2) (azide) 1.993(7); Mn(1)-O 1.743(4); Mn(2)-O 1.794(4) Å.

The pure dimeric complex gives no e.s.r. spectrum from 4 to 300 K and has a magnetic moment at 25 °C determined by the Gouy method of $2.01 \,\mu_{\rm B}$ per dimer, implying an antiferromagnetic coupling interaction between the two manganese atoms. The complex reacts with triphenylphosphine to give triphenylphosphine oxide in 99% yield and with alkanes, alkenes, or chlorocarbons to give a variety of products in low yields. In all the reactions the manganese(IV) is reduced to manganese(III) with the monomeric $[Mn^{III}(TPP)]^+$ chromophore being produced in effectively quantitative yield.

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¹G. Porter, Proc. R. Soc. London, Ser. A, 1978, 362, 281; I. Tabushi and S. Kojo, Tetrahedron Lett., 1975, 305.

 ² A. Harriman, and G. Porter, J. Chem. Soc., Faraday Trans. 2, 1979, 75, 1543.
³ I. Willner, J. W. Otvos, and M. Calvin, J. Chem. Soc., Chem. Commun., 1980, 964; I. Tabushi and N. Koga, J. Am. Chem. Soc., 1979, 101, 6456.

⁴ The most definitive structural characterization of a Mn^{IV} porphyrin is the e.s.r. study of the Mn^{IV} (TPP). O_2^{2-} complex reported by B. M. Hoffman, C. J. Weschler, and F. Basolo, J. Am. Chem. Soc., 1976, 98, 5473. ⁵ Mn^{III} (TPP)N₃ was made by ligand exchange from Mn^{III} (TPP) (OAc) and NaN₃ by the procedure used by H. Ogoshi, E. Watanabe,

Z. Yoshida, J. Kincaid, and K. Nakamoto, J. Am. Chem. Soc., 1973, 95, 2845 for the corresponding iron complexes. Mn¹¹¹ was prepared by the method of A. D. Adler, F. R. Longo, F. Kampas, and J. Kim J. Inorg. Nucl. Chem., 1970, 32, 2443. ⁶C. K. Johnson, Report ORNL-3794, Oak Ridge National Laboratory, Oak Ridge, Tennessee, 1965. Mn^{III} (TPP)(OAc)