Novel Syntheses of 1,6-Methano[10]annulene and 4-Methylazulene

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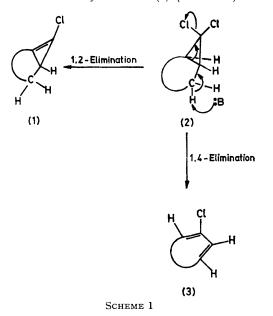
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Summary Reaction of 11,11-dichlorotricyclo[4.4.1.0^{1,6}]-undeca-3,8-diene (4) with potassium t-butoxide in dimethyl sulphoxide affords 1,6-methano[10]annulene (9) and 4-methylazulene (10); similar treatment of the dibromopropellane (5) yields the azulene (10), the mono-

bromopropellane (11), but diminished yields of the annulene (9).

DIHALOGENOCARBENE adducts (2) of medium- and large-ring cycloalkenes are believed to undergo initial 1,2-elimination

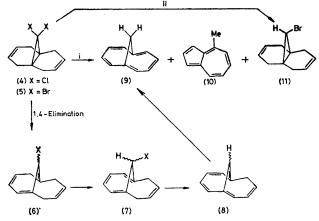
to give ω -chlorobicyclo[n.1.0]alk-1(ω)-enes (1) on treatment with strong base.¹ We have recently reported² the basepromoted reactions of tri- and tetra-halogeno-cis-transoidcis-tricyclo[5.1.0.0^{3,5}]octanes. In spite of the possibilities for 1,2-elimination in these systems, evidence was presented for 1,4-elimination with cleavage of the cyclopropyl moiety and formation of a cyclic diene (3) (Scheme 1). An in-



vestigation of the reactions of gem-dihalogeno[n.m.1]propellanes in strongly basic media, where possible competition from 1,2-elimination reactions is avoided, would be of interest.[†] Herein we report the preliminary results of such a study which demonstrate novel and simple syntheses of 1,6-methano[10]annulene (9) and 4-methylazulene (10).

Reaction of readily available³ 11,11-dichlorotricyclo- $[4.4.1.0^{1,6}]$ undeca-3,8-diene (4) with 5 mol. equiv. of potassium t-butoxide (in the form of its t-butyl alcohol monosolvate) in oxygen-free dimethyl sulphoxide at room temperature for 16 h afforded, after extractive work-up and t.l.c., 1,6-methano[10]annulene (9) (13%), and the deep blue 4-methylazulene (10) (10%). Similar treatment of 11,11-dibromotricyclo[4.4.1.0^{1,6}]undeca-3,8-diene (5) afforded 4-methylazulene (10) (10%), diminished quantities (1%) of 1,6-methano[10]annulene (9), and a small amount (4%)of 11-bromotricyclo[4.4.1.0^{1,6}]undeca-3,8-diene (11). The annulene (9) obtained in these reactions was identical in all respects with an authentic sample prepared by the method of Vogel and his co-workers.³ Comparison of the ¹H n.m.r. spectrum of the azulene (10) with the published⁴ spectra of

the five possible methylazulenes allowed its unequivocal assignment as the 4-methyl isomer. All additional spectroscopic data are consistent with this regioisomer. The monobromopropellane (11) was identical in all respects with an authentic sample prepared by half-reduction of the dibromocompound (5) with tri-n-butyltin hydride.⁵



SCHEME 2. Reagents: i, Bu^tOK, Me₂SO; ii, Buⁿ₃SnH, benzene.

A plausible mechanism for the conversion of the propellanes (4) and (5) into compound (9) is illustrated in Scheme 2. Initial 1,4-elimination would afford the bridgehead dienes (6). [Species closely related to compound (6) have been proposed to explain apparent retention of configuration in bimolecular nucleophilic substitution at a cyclopropyl carbon atom.⁶] Base-promoted prototropic rearrangement of the tetraenes (6), to give the tetraenes (7), followed by dehydrohalogenation would give rise to the pentaene (8). Further prototropic rearrangement would yield the aromatic annulene (9). The mechanism for azulene formation is less obvious. The annulene (9) is inert to treatment with potassium t-butoxide in dimethyl sulphoxide and is thus precluded as a precursor to 4methylazulene (10).

The supposition that the tetraenes (6) are intermediates en route to 1,6-methano[10]annulene suggests that basepromoted 1,4-eliminations in halogeno[n.m.1] propellanes may provide a useful route to bridgehead dienes, which are species of considerable current interest.7 Also, in view of the recent efforts⁸ devoted to azulene syntheses, the application of strategies based on the above observations may be of some synthetic utility for the preparation of such compounds.

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† To the best of our knowledge only one study of base-promoted reactions of halogeno[n.m.1]propellanes has been reported; L. A. M. Turkenburg, J. W. Van Straten, W. H. De Wolf, and F. Bickelhaupt, J. Am. Chem. Soc., 1980, 102, 3256 and references cited therein. ¹ W. E. Billups, B. A. Baker, W. Y. Chow, K. H. Leavell, and E. S. Lewis, J. Org. Chem., 1975, 40, 1702; C. J. Ransom and C. B. Reese, J. Chem. Soc., Chem. Commun., 1975, 970; J. Arct and B. Migaj, Tetrahedron, 1981, 37, 953. ² M. G. Banwell and B. Halton, Aust. J. Chem., 1979, 32, 849; Tetrahedron Lett., 1979, 3191; Aust. J. Chem., 1979, 32, 2689; ibid.,

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