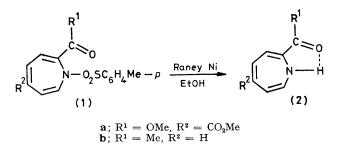
Stable 1H-Azepines

By NAGARAJ R. AYYANGAR, ARUN K. PUROHIT, and BAL D. TILAK (National Chemical Laboratory, Poona 411008, India)

Summary The hitherto unknown 1H-azepines (2), stabilized by hydrogen bonding between the NH-hydrogen and the adjacent CO-oxygen, have been prepared by detosylation of their N-tosyl derivatives (1). SEVERAL 2*H*-, 3*H*-, and 4*H*-azepine derivatives with substituents on one or more of the carbon atoms are known,¹ and a number of *N*-substituted 1*H*-azepines have also been reported.² We have reported the synthesis of *N*-tosyl derivatives of 1H-azepines with electron-withdrawing substituents such as CO2Me, Ac, and NO2 α to the ring nitrogen, $^{3-5}$ but stable 1*H*-azepines have not been reported so far.⁶ During the photolysis of phenyl azide in diethylamine, 2-diethylamino-1H-azepine was detected as an unstable intermediate on the basis of n.m.r. evidence,7 but work-up gave only the 3H-azepine derivative.

We now report the synthesis of stable 1H-azepines, with $CO_{\circ}Me$ or Ac subsituents α to the ring nitrogen atom. The N-tosylazepines (1a) and $(1b)^{3,5}$ were detosylated with Raney nickel to yield 2,5-bismethoxycarbonyl- (2a) and 2-acetyl-1*H*-azepine (2b).† Thus, a stirred mixture of (1a) and Raney nickel (1 ml) in ethanol (10 ml) was refluxed for 15 h. After removal of nickel and evaporation, column chromatography of a solution of the product on silica gel with benzene-ethyl acetate (4:1) as eluant gave (2a) as yellow needles, m.p. 112 °C (benzene) (70%) and unchanged (1a) (0.06 g). The 1H-azepine (2a) showed u.v.-visible absorptions at 234 and 396 nm, whereas (1a) showed absorption at 355 nm only; ν_{max} for (2a) 3340 (NH), and 1740 and 1700 cm⁻¹ (free and hydrogen-bonded ester CO groups); ¹H n.m.r. δ 6.8–8 (m, NH and 4 azepine ring H), and 3.73 and 3.9 (CO₂Me); m/e 209 (M⁺).

Similarly, (2b) was obtained from (1b) in 75% yield, platelets, m.p. 144 °C (from benzene); $\nu_{max} \; 3300$ (NH) and



1640 cm⁻¹ (hydrogen-bonded CO); m/e 135 (M⁺); δ 6.7–7.9 (m, NH and 4 azepine ring H) and 2.7 (COMe). There was no indication of the presence of methylene or methine protons in the n.m.r. spectra of (2a) and (2b), thus ruling out the possibility of CH-azepine structures. In both compounds, the NH proton was strongly hydrogen-bonded to the adjacent CO group and was not exchangeable with D₂O; this hydrogen-bonding clearly stabilizes these compounds.

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† Satisfactory elemental analyses were obtained.

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