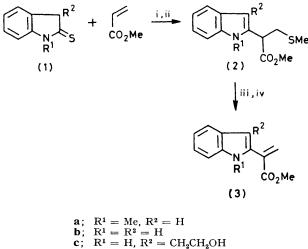
Cycloaddition Reactions of 2-(Indol-2-yl)acrylates and N-Alkoxycarbonyl-1,2-dihydropyridine; New Synthesis of Desethylcatharanthine

By CHRISTIAN MARAZANO, JEAN-LOUIS FOURREY, and BHUPESH C. DAS* (Institut de Chimie des Substances Naturelles, C.N.R.S., 91190 Gif-sur-Yvette, France)

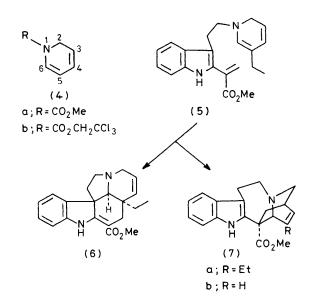
Summary A new synthesis of desethylcatharanthine (7b) by utilising a cycloaddition reaction between the 2-(indol-2-yl)acrylate (10) and the N-alkoxycarbonyl-1,2-di-hydropyridine (4b) is described.

THE photoaddition reactions between the indoline-2thiones (1) and methyl acrylate provide¹ a facile access to the methyl 2-(indol-2-yl)acrylate derivatives (3). Unlike other methods,^{2,3} this procedure¹ offers the possibility of using an indoline-2-thione (1) with or without a substituent on the nitrogen or the 3-position of the indole, thereby enabling us to obtain indoleacrylates, such as (3a) and (3b), by way of their stable precursors (2a) and (2b). (Scheme 1).

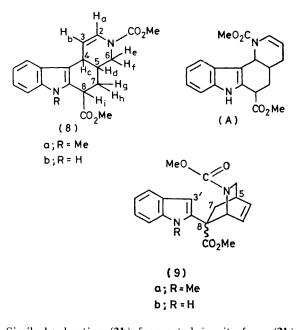


SCHEME 1. Reagents: i, h_V ; ii, CH_2N_2 ; iii, m-ClC₆H₄CO₃H; iv, toluene, heat for 5 min at 110 °C.

We have investigated the reactivity of the acrylates (3a) and (3b) toward the dihydropyridine (4a) to determine the cycloaddition reaction conditions that could, ultimately, lead to Aspidosperma- and/or Iboga-type alkaloids via a biomimetic pathway resembling the proposed⁵ intramolecular Diels-Alder type cyclisation of the hypothetical intermediate dehydrosecodine (5) into tabersonine (6) and catharanthine (7a). The scope of the method has been tested by effecting a short synthesis of desethylcatharanthine (7b).6



In refluxing toluene, the reaction of (3a) with (4a) gave, in addition to the dimers (M 430, yield 25%) of (3a), a mixture of two types of adduct (8a) and (9a) in 6 and 30%isolated yields, respectively. The isomeric cycloaddition products could be differentiated by their mass spectra. The major products (9a) $(M^+ 354)$ gave the base peak at m/z 215 [corresponding to (3a)], resulting from retro-Diels-Alder fragmentation. This peak at m/z 215 is of low intensity in the mass spectrum of (8a), which showed the most abundant fragment ions at m/z 339 (M - 15). The ¹H n.m.r. spectrum of (8a) revealed only two olefinic protons which were mutually coupled. Their chemical shift values (δ 6.70 and 5.02) were different from those observed for the olefinic protons7 of the Aspidosperma alkaloid tabersonine (6). The proposed structure of (8a) was reached principally on the basis of double-resonance ¹H n.m.r. experiments with the related isomeric compound (8b) (see below). The two isomers of (9a), with α - and β -CO₂Me, were isolated in the ratio α/β 3/2. Unlike compound (8a) they could be characterised by the presence of the indole 3'-H proton signals at δ 6.30 and 6.89. The major isomer of (9a) was assigned the α -configuration by analogy with the results previously reported by Sundberg and Bloom.6



Similarly, heating (3b) [generated in situ from (2b) in refluxing toluene] with (4a) gave (8b) (two isomers which could be separated) and (9b) in 45 and 7% yields, respectively, in addition to the dimers (M 402, yield 25%) of (3b). The spectral data were consistent with the proposed structures (8b) and (9b). Thus, the ¹H n.m.r. spectrum (250 MHz, CDCl₃, recorded at 60 °C because of the slow rotation of the carbamate group at room temperature⁶) of the major isomer of (8b) showed δ 8.12 (s, >NH), 7.6-7.1 (m, 4 H, aromatic), 6.75 (broad signal at 60 °C, Ha; at 20 °C appeared as a pair of doublets at 6.83 and 6.68, J_{ab} 8 Hz each), 5·1 (dd, J_{ab} 8 Hz, J_{bc} ca. 1·5 Hz, † H_b),

[†] The signal due to 4-H_c could not be detected as it remained obscured by the O-methyl signals.

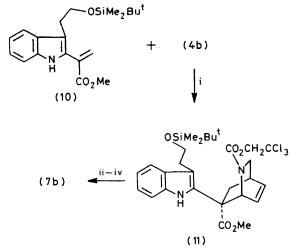
4.02 (dd, J_{ef} 13 Hz, J_{de} 5 Hz, H_e), 3.84 (dd, J_{hi} 6 Hz, Jgi 4 Hz, Hi), 3.72 (s, 3 H, CO₂Me), 3.70 (s, 3 H, CO₂Me), 3.56 (dd, Jef 13 Hz, Jdf 4 Hz; Hf), 2.56 (broad m, Hd), 2.23 (broad d, appeared as dd on irradiating H_d , J_{gh} 13 Hz, J_{gi} 4 Hz, Hg), and 2.10 (ddd, J_{gh} 13 Hz, J_{hi} 6 Hz, J_{dh} 11 Hz, H_h).

The interpretation of the ¹H n.m.r. data followed from appropriate decoupling experiments. In particular, the irradiation of the 6-C methylene protons $(H_e \text{ and } H_f)$ did not modify the splitting pattern of 3-H_b thus ruling out the alternative structure (A). Similar results were also observed with the minor isomer of (8b).

Finally, we have examined the behaviour of the 3-substituted indoleacrylate (10)[±] toward the dihydropyridine (4b).⁸ The only adduct which could be isolated (30%)yield) besides the dimers of (10) was assigned the structure (11) on the basis of its facile conversion into desethylcatharanthine (7b) (Scheme 2), an independent synthesis of which has been recently reported.6

Surprisingly, we could not detect any Aspidosperma-type cycloaddition product in the reaction mixture. It is very likely that an initial bond formation between C-5 and C-7 occurs giving, ultimately, compounds of type (8) or (9). Also, it is noteworthy that a substituent on the nitrogen or at the 3-position of indole favours the formation of the Iboga-type addition products.

The synthesis of desethylcatharanthine (7b) described here is simple and straightforward. We are currently investigating the application of a similar approach to the synthesis of catharanthine (7a).



SCHEME 2. Reagents and conditions: i, toluene, 110 °C for 18 h, 30% yield: ii, AcOH, H₂O, tetrahydrofuran, overnight, 65%; iii, MeSO₂Cl, NEt₃, CH₂Cl₂, 98%; iv, Zn-MeCN, 60 °C for 1 h, 35%.

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 \pm This compound was obtained from (2c) in 76% yield by successive treatment with chlorodimethyl-t-butylsilane-CH₂Cl₂-NEt₃-4-dimethylaminopyridine (E. J. Corey and A. Wenkateswarlu, J. Am. Chem. Soc., 1972, 94, 6190; S. K. Chaudhary and O. Hernandez, *Tetrahedron Lett.*, 1979, 99) and m-chloroperbenzoic acid followed by heating in toluene at 110 °C.

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