

Cyclopentadienyl-ruthenium and -osmium chemistry. Cleavage of Tetracyanoethylene under Mild Conditions: X-Ray Crystal Structures of $[\text{Ru}\{\eta^3\text{-C}(\text{CN})_2\text{CPh}=\text{C}(\text{CN})_2\}(\text{PPh}_3)(\eta\text{-C}_5\text{H}_5)]$ and $[\text{Ru}\{\text{C}[\text{C}(\text{CN})_2]\text{CPh}=\text{C}(\text{CN})_2\}(\text{CNBu}^t)(\text{PPh}_3)(\eta\text{-C}_5\text{H}_5)]$

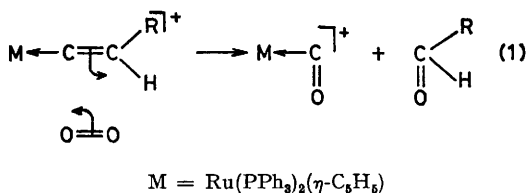
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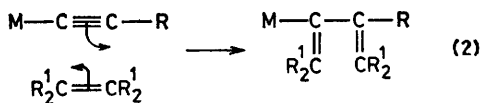
Summary The reaction between $[\text{Ru}(\text{C}_2\text{Ph})(\text{PPh}_3)_2(\eta\text{-C}_5\text{H}_5)]$ and $\text{C}_2(\text{CN})_4$ affords $[\text{Ru}\{\eta^3\text{-C}(\text{CN})_2\text{CPh}=\text{C}(\text{CN})_2\}(\text{PPh}_3)(\eta\text{-C}_5\text{H}_5)]$ (**2**), which forms $[\text{Ru}\{\text{C}[\text{C}(\text{CN})_2]\text{CPh}=\text{C}(\text{CN})_2\}(\text{L})(\text{PPh}_3)(\eta\text{-C}_5\text{H}_5)]$ (**5**) with $\text{L} = \text{CO}$ or CNBu^t ; the structures of (**2**) and (**5**; $\text{L} = \text{CNBu}^t$) have been determined by X-ray crystallography.

95.2 (s, C-19 and C-21), **87.0** (s, C_5H_5), **113.4**, **114.1**, **114.5** and **118.4** (all s, CN), **128.4**—**136.6** (m, PPh_3), and **178.1** (d, $J_{\text{CP}} = 2 \text{ Hz}$, RuCN). The formulation of (**2**), and the presence of three ^{13}C resonances coupled to the ^{31}P nucleus of the remaining PPh_3 ligand, suggested that the adduct contained

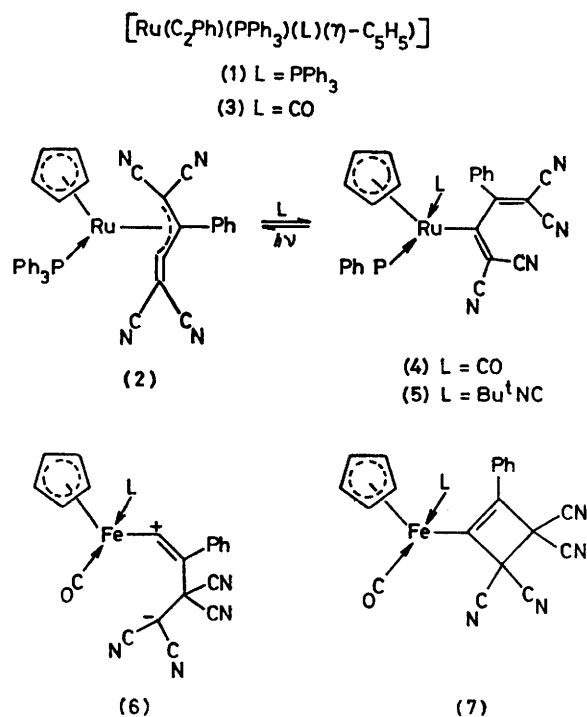
We have previously reported¹ the oxidation of vinylidene complexes to the corresponding aldehyde and metal carbonyl derivative [reaction (1)]. We now report a novel



example of this type of reaction, which is formally related to olefin metathesis, that of a metal-substituted acetylide with an electron-deficient olefin [reaction (2)].



An intense dark green colour develops on mixing solutions of tetracyanoethylene (tcne) and $[\text{Ru}(\text{C}_2\text{Ph})(\text{PPh}_3)_2(\eta\text{-C}_5\text{H}_5)]$ (**1**) in benzene; the intermediate, as yet uncharacterised, gives a strong e.s.r. signal ($g \text{ ca. } 2.01$). After some hours, orange crystals of a 1:1 adduct, $[\text{Ru}(\text{C}_2\text{Ph})(\text{tcne})(\text{PPh}_3)(\eta\text{-C}_5\text{H}_5)]$ (**2**) separate in 82% yield [M^+ at m/e 658 (for ^{102}Ru); $\nu(\text{CN})$ 2215 m, $\nu(\text{C}=\text{C})$ 1595 m cm^{-1} ; ^1H n.m.r. $\delta(\text{CDCl}_3)$ 4.76 (s, C_5H_5) and 7.4—7.5 (m, PPh_3); ^{13}C n.m.r. $\delta(\text{CDCl}_3)$ 7.3 (d, $J_{\text{CP}} = 5 \text{ Hz}$), 84.9 (d, $J_{\text{CP}} = 7 \text{ Hz}$), (two of three allylic C); 66.6 (s, C-4), 92.2 (s, C_5H_5), 111.0, 115.9, 118.7 and 119.0 (all s, CN), and 128.4—135.9 (m, Ph)]. A similar reaction between tcne and $[\text{Ru}(\text{C}_2\text{Ph})(\text{CO})(\text{PPh}_3)(\eta\text{-C}_5\text{H}_5)]$ (**3**) afforded $[\text{Ru}(\text{C}_2\text{Ph})(\text{tcne})(\text{CO})(\text{PPh}_3)(\eta\text{-C}_5\text{H}_5)]$ (**4**) (82%), also formed by carbonylation of (**2**) (100 °C; 53 bar; 17 h) [M^+ at m/e 686; $\nu(\text{CN})$ 2200 m, $\nu(\text{CO})$ 1941 vs, $\nu(\text{C}=\text{C})$ 1505 m cm^{-1}]. Short u.v. irradiation of (**4**) results in loss of CO and conversion into (**2**). The related complex $[\text{Ru}(\text{C}_2\text{Ph})(\text{tcne})(\text{CNBu}^t)(\text{PPh}_3)(\eta\text{-C}_5\text{H}_5)]$ (**5**) was obtained (91%) from (**2**) and Bu^tNC [M^+ at m/e 741; $\nu(\text{CN})$ 2207w, 2200w, 2193w, and 2167s; $\nu(\text{C}=\text{C})$ 1506w cm^{-1} ; ^1H n.m.r. $\delta(\text{CDCl}_3)$ 1.33 (s, Me), 4.62 (s, C_5H_5), and 7.0—7.4 (m, PPh_3); ^{13}C n.m.r. $\delta(\text{CDCl}_3)$ 30.5 (s, CMe_3), 58.1, (s, CMe_3), 73.6 and



an allylic ligand. The ready addition of 2e donor ligands to give (**4**) or (**5**) could be explained by conversion of the η^3 -ligand in (**2**) into a related η^1 -ligand in these complexes. These ideas have been confirmed by single-crystal X-ray diffraction studies of complexes (**2**) and (**5**).

Crystal data: (**2**), $\text{C}_{32}\text{H}_{20}\text{N}_4\text{PRu}$, M_r 592.6, monoclinic, space group $P2_1/c$, $a = 8.870(3)$, $b = 17.762(5)$, $c = 19.966(2)$ Å, $\beta = 99.22(2)^\circ$, $U = 3104.96$ Å³, $Z = 4$, $D_c = 1.267$ g cm^{-3} ; $F(000) = 1888$; $\mu(\text{Mo-K}\alpha) = 6.08$ cm^{-1} ; (**5**) $\text{C}_{37}\text{H}_{20}\text{N}_5\text{PRu}$, M_r 675.7, monoclinic, space group $P2_1/c$, $a = 17.204(2)$, $b = 11.349(2)$, $c = 18.679(3)$ Å, $\beta = 90.29(3)^\circ$, $U = 3646.9$ Å³; $Z = 4$, $D_c = 1.23$ g cm^{-3} ; $F(000) = 1972$; $\mu(\text{Mo-K}\alpha) = 4.60$ cm^{-1} .

Both structures were solved by Patterson techniques and refined by block-matrix least squares methods using the SHELX² system of programmes. The hydrogen atom positions were calculated geometrically with all C-H

distances fixed at 0.965 Å. The final R -factors are: (2), $R = 4.64\%$, $R_w = 4.18\%$, $w = 1.8106/(\sigma^2 F_0 + 0.0001 F_0^2)$ for 4064 independent reflexions [$I > 2.5\sigma(I)$; $2.5 \leq 2\theta \leq 50^\circ$, Stoe 2-circle diffractometer]; (5), $R = 3.71\%$, $R_w = 3.69\%$, $w = 2.5186/(\sigma^2 F_0 + 0.0001 F_0^2)$ for 4727 independent reflexions [$I > 2.5\sigma(I)$; $2.5 \leq 2\theta \leq 50^\circ$, Stoe 2-circle diffractometer].†

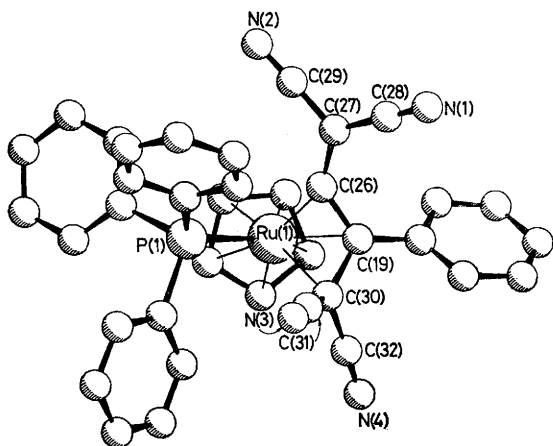


FIGURE 1. PLUTO plot (ref. 5) of $[\text{Ru}\{\eta^3\text{-C}(\text{CN})_2\text{CPh}=\text{C}(\text{CN})_2\}(\text{PPh}_3)(\eta\text{-C}_6\text{H}_5)]$. Selected bond lengths: Ru(1)–C(19), 2.135(4); Ru(1)–C(26), 1.919(5); Ru(1)–C(30), 2.231(4); C(26)–C(27), 1.383(6); C(19)–C(26), 1.432(7); C(19)–C(30), 1.476(6) Å.

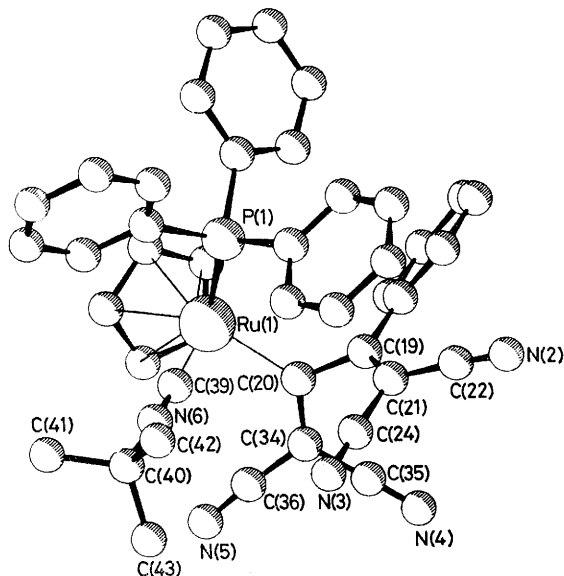


FIGURE 2. PLUTO plot (ref. 5) of $[\text{Ru}\{\text{C}=\text{C}(\text{CN})_2\}\text{CPh}=\text{C}(\text{CN})_2\}(\text{CNBu}^t)(\text{PPh}_3)(\eta\text{-C}_6\text{H}_5)]$. Selected bond lengths: Ru(1)–C(20), 2.074(3); Ru(1)–C(39), 2.003(3); C(19)–C(21), 1.362(4); C(19)–C(20), 1.478(4); C(20)–C(34), 1.382(5) Å.

† Atomic co-ordinates are available on request from the Director of the Cambridge Crystallographic Data Centre, University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW. Any request should be accompanied by the full literature citation for this communication.

¹ M. I. Bruce, A. G. Swincer, and R. C. Wallis, *J. Organomet. Chem.*, 1979, **171**, C5.

² G. M. Sheldrick, 'SHELX—Programme for Crystal Structure Determination,' University of Cambridge, 1976.

³ A. Davison and J. P. Solar, *J. Organomet. Chem.*, 1979, **166**, C13.

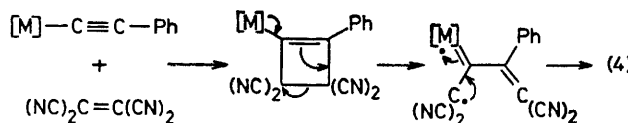
⁴ F. Kataoka, N. Shimizu, and S. Nishida, *J. Am. Chem. Soc.*, 1980, **102**, 711; J. K. Williams, D. W. Wiley, and B. C. McKusick, *ibid.*, 1982, **84**, 2210; A. Felzenstein, S. Sarel, and J. Yovell, *J. Chem. Soc., Chem. Commun.*, 1975, 918.

⁵ W. D. S. Motherwell, 'PLUTO—Programme for Crystal and Molecular Plotting,' University of Cambridge, 1977.

Figure 1 shows the molecular structure of (2), with significant structural parameters listed in the caption. The main feature of interest is the η^3 -allylic ligand formed by addition of two dicyanomethylene fragments, one to each acetylenic carbon of the phenylacetylide ligand of (1). Figure 2 similarly shows the molecular structure of (5), from which it can be seen that the allylic ligand in (2) has been converted into the η^1 -1,1,4,4-tetracyano-3-phenylbuta-1,3-dien-2-yl group by addition of the isocyanide ligand. Of some interest is the angle between the planes containing the two dicyanomethylene groups [torsion angle about C(21)–C(19)C(20)C(34), 105°], indicating that the diene fragment is considerably twisted, with consequent localisation of the two double bonds.

We note that similar reactions between tcne and $[\text{Fe}(\text{C}_2\text{Ph})(\text{CO})(\text{L})(\eta\text{-C}_6\text{H}_5)]$ ($\text{L} = \text{CO}$ or PPh_3) proceed to give complexes formulated as the dipolar intermediate (6), and the (2 + 2) cycloadduct (7); for $\text{L} = \text{PPh}_3$ these two derivatives form an equilibrium mixture.⁹

To our knowledge no previous example of cleavage of the olefinic bond of tcne on reaction with an alkyne is known; the product expected from these reactions is the (2 + 2) cycloaddition product. Several reports describe the ring-opening of cyclobutyl derivatives [(2 + 2) cycloadducts of tcne and olefins] on heating (100–180 °C), probably *via* radical intermediates.⁴ The present reaction appears to be related to these; back-bonding into the acetylide group from the metal atom, perhaps to the extent of formation of a formally ruthenium(III) complex, may serve to facilitate the formation of the intermediate radical species, and hence cleavage of tcne under mild conditions (Scheme). The



SCHEME.

formation of (4), rather than (2), from the reaction between (3) and tcne suggests that the latter does not first coordinate to the metal atom, but that attack occurs initially at the acetylide carbon; the formation of the η^3 -allylic system in (2) is encouraged by the ready loss of one of the PPh_3 ligands, as is commonly found for $\text{Ru}(\text{PPh}_3)_2(\eta\text{-C}_6\text{H}_5)$ complexes.

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