

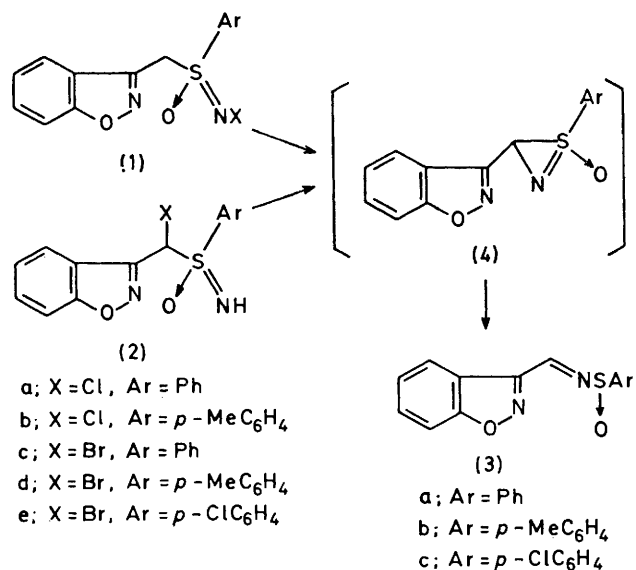
## Novel Rearrangements of *N*- and $\alpha$ -Halogeno-derivatives of *S*-Aryl-*S*-[(1,2-benzisoxazol-3-yl)methyl]sulphoximides to the Corresponding *N*-Sulphonylimines

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The title *N*- and  $\alpha$ -halogeno-sulphoximides (**1**) and (**2**) undergo base-induced rearrangement reactions to give the corresponding *N*-sulphonylimines (**3**), suggesting the intermediacy of a three-membered cyclic sulphoximide (**4**) with an endocyclic S=N moiety.

Much attention has been focused on the synthesis and properties of sulphoximide derivatives which have wide chemical and biological interest.<sup>1</sup> We have previously reported<sup>2</sup> the rearrangement of the *N*-halogeno-*S*-aryl-*S*-[(1,2-benzisoxazol-3-yl)methyl]sulphoximides (**1**) to the corresponding

$\alpha$ -halogeno-derivatives (**2**). We now report the novel rearrangement of compounds (**1**) and (**2**) with base to give the same *N*-sulphonylimines, the 3-arylsulphonyliminomethyl-1,2-benzisoxazoles (**3**), suggesting that these rearrangements involve the same intermediate, a thiazirine *S*-oxide (**4**), which



has a novel three-membered ring system with an endocyclic S=N moiety.†

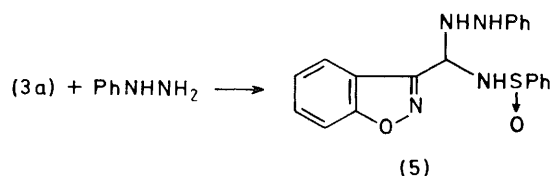
Compounds (1) and (2) were treated with 1,5-diazabicyclo-[5.4.0]-undec-5-ene (DBU) or potassium carbonate in chloroform or dichloromethane, followed by column chromatography on silica gel with chloroform as an eluant to afford compounds (3) as shown in Table 1.

The structure of compounds (3) was confirmed by elemental and spectral analyses [*e.g.*, (3a): m.p. 135–137 °C;  $\nu$  1608 and 1593 cm<sup>-1</sup> (C=N), 1106 cm<sup>-1</sup> (SO);  $\delta$  (60 MHz; CDCl<sub>3</sub>) 7.2–8.4 (m, 9H, ArH) and 9.28 (s, 1H, CH=N)], and further by the following chemical transformation of compound (3a). Thus, reaction of (3a) with phenylhydrazine in ethanol gave the adduct (5), m.p. 127–131 °C, in 60% yield. The structure

Table 1. Rearrangements of (1) and (2) with base.<sup>a</sup>

Compound	Base	Solvent	Time	Product	Yield (%)
(1a)	DBU	CH <sub>2</sub> Cl <sub>2</sub>	5 min	(3a)	83
(1a)	K <sub>2</sub> CO <sub>3</sub>	CH <sub>2</sub> Cl <sub>2</sub>	3–5 h	(3a)	85
(1b)	DBU	CH <sub>2</sub> Cl <sub>2</sub>	5 min	(3b)	85
(1c)	DBU	CH <sub>2</sub> Cl <sub>2</sub>	5 min	(3a)	56 <sup>b</sup>
(1c)	K <sub>2</sub> CO <sub>3</sub>	CH <sub>2</sub> Cl <sub>2</sub>	5 h	(3a)	85
(1d)	K <sub>2</sub> CO <sub>3</sub>	CH <sub>2</sub> Cl <sub>2</sub>	5 h	(3b)	85
(2a)	DBU	CHCl <sub>3</sub>	5 h	(3a)	15 <sup>c</sup>
(2b)	DBU	CHCl <sub>3</sub>	4–5 h	(3b)	16 <sup>c</sup>
(2c)	DBU	CHCl <sub>3</sub>	2 h	(3a)	86
(2c)	K <sub>2</sub> CO <sub>3</sub>	CHCl <sub>3</sub>	6 h	(3a)	22 <sup>c</sup>
(2d)	DBU	CHCl <sub>3</sub>	2 h	(3b)	83
(2e)	DBU	CHCl <sub>3</sub>	2 h	(3c)	80

<sup>a</sup> Carried out at room temperature using a slight excess of DBU or 2 mol. equiv. of potassium carbonate unless otherwise noted. <sup>b</sup> *S*-[1,2-Benzisoxazol-3-yl)methyl]-*S*-phenylsulphoximide was also obtained in 22% yield. <sup>c</sup> Carried out under reflux. Recoveries of (2a), (2b), and (2c) were 65, 63, and 63%, respectively.



of (5) was also confirmed by elemental and spectral analyses, especially by the <sup>1</sup>H n.m.r. spectrum which showed the presence of the partial structure PhNHNHCHNH–;  $\nu$  3050 cm<sup>-1</sup> (NH), 1025 and/or 1040 cm<sup>-1</sup> (SO);  $\delta$  (100 MHz; CDCl<sub>3</sub>) 6.7–8.1 (m, 14H, ArH), 6.24 (s, 1H, PhNH), 5.92 (dd, *J* 8.1 and 9.7 Hz, 1H, NHCHNH), 5.56 (d, *J* 9.7 Hz, 1H, CHNHS), and 4.20 (d, *J* 8.1 Hz, 1H, NHNHCH); deuteration with D<sub>2</sub>O resulted in disappearance of the peaks at  $\delta$  6.24, 5.56, and 4.20 and in collapse of the dd peak at  $\delta$  5.92 into a singlet. These data support the sulphonylimine structure of compounds (3).

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- T. Yoshida, S. Naruto, and H. Nishimura, *Tetrahedron Lett.*, 1980, **21**, 2315; T. Yoshida, S. Naruto, H. Uno, and H. Nishimura, *Chem. Pharm. Bull.*, accepted for publication.

† C. R. Johnson and H. G. Corkins, *J. Org. Chem.*, 1978, **43**, 4140, reported the intermediacy of a three-membered cyclic sulphoximide with an exocyclic S=N moiety in the base-induced Ramberg–Backlund-type reaction of  $\alpha$ -halogeno-*N*-(*p*-tolylsulphonyl)-sulphoximides to give alkenes, and also that the reaction of *S*-butyl-*S*-(1-chlorobutyl)sulphoximide with KOH–MeOH gave butane-1-sulphinamide, the production of which could be rationalized in a number of ways, including a three-membered S–N heterocyclic intermediate, but the data were not sufficient for an unambiguous determination of the exact mechanism; the present results seem to support this speculation.