

## An Unexpected Outcome of a General Thiazole Synthesis

G. Denis Meakins,\* Michael D. J. Padgham, (Miss) Neeta Patel, and (Mrs) Josephine M. Peach

Dyson Perrins Laboratory, Oxford University, South Parks Road, Oxford OX1 3QY, U.K.

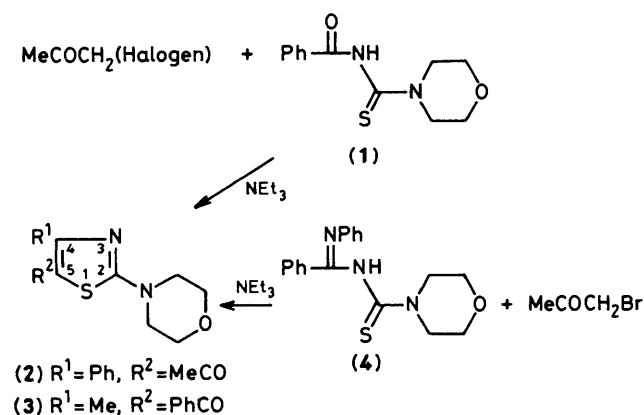
The reaction between *N*-benzoyl-*N'*-methyl-*N'*-phenylthiourea and chloroacetone gives 5-benzoyl-4-methyl-2-(*N*-methyl-*N*-phenylamino)thiazole rather than the 5-acetyl-4-phenylisomer which would be predicted from previous work on this route to thiazoles.

Alternatives to the Hantzsch synthesis of 2-aminothiazoles have been studied actively during the last two decades. One such approach involves condensation between suitable *N,N'*-substituted thioureas and compounds with reactive methylene groups: in effect the ends of a C–N–C–S chain are joined to the methylene C which becomes C-4 of the thiazole ring.<sup>1</sup> The preparation of compound (2) from the thiourea (1) shown in Scheme 1 is an example from the first general application,<sup>2</sup> and a subsequent variation<sup>3</sup> employs the related *N*-benzimidoyl derivative (4) [which may be represented equally well as the Ph–C(NHPh)=N– tautomer]. Later,<sup>4</sup> it was established that an *N*-benzoyl-*N'*-monosubstituted thiourea gives a 2-benzoylimino-3-substituted  $\Delta^4$ -thiazoline rather than the 2-*N*-monosubstituted aminothiazole, as reported,<sup>2</sup> but the latter is obtained from the corresponding *N*-benzimidoyl-*N'*-monosubstituted thiourea. Notwithstanding this complication no question has been raised hitherto about the nature of the products obtained from the *N,N'*-disubstituted thioureas [e.g., structure (2) for the product from the thioureas (1) and (4)].

In developing work on 2-aminothiazoles with electron-withdrawing groups at position 5 we thought to prepare the

5-acetyl compound (6) from *N*-benzoyl-*N'*-methyl-*N'*-phenylthiourea (7).<sup>5</sup> Doubts about the product's structure were raised by the mass spectrum which indicated the presence of a benzoyl group, as in structure (8). The form and C–H coupling constants of the <sup>13</sup>C n.m.r. signals logically ascribed to the C-4 and C(O) atoms also pointed to the isomeric structure, and the matter was put beyond doubt by a crystallographic examination.<sup>†</sup> For comparison, the compound truly represented by structure (6) was prepared unambiguously from 2-(*N*-methyl-*N*-phenylamino)-4-phenylthiazole (5) (itself readily obtained from phenacyl bromide and *N*-methyl-*N*-phenylthiourea).<sup>6</sup> The main spectrometric properties of isomers (6) and (8),<sup>‡</sup> both new compounds, confirm the structural assignments, and (*sic*) the 5-acetyl compound (6) has its C–Me <sup>1</sup>H n.m.r. signal at  $\delta$  1.98. A mechanism for the formation of compound (8) from the benzoylthiourea (7) will be proposed in the full paper; a key observation is that compound (8) is the main product of a Hantzsch condensation between PhCOCHBrCOMe and *N*-methyl-*N*-phenylthiourea.

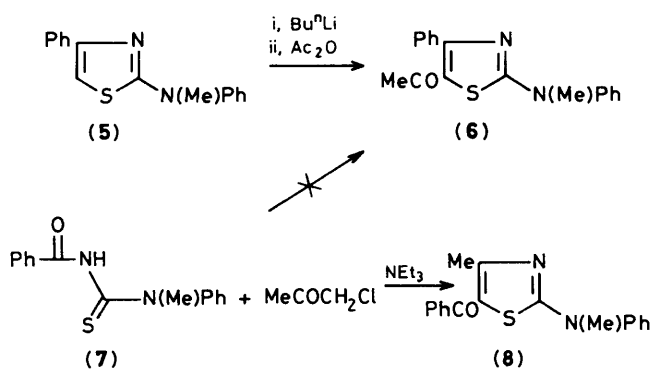
Repetition of the reaction between the benzoylthiourea (1) and chloroacetone gave the reported product.<sup>2</sup> Spectrometric examination showed that it, too, has the unexpected features of a benzoyl group *etc.*, denoting structure (3) rather than structure (2) as published.<sup>2</sup> Now although Ried and Kaiser<sup>3</sup> concluded that the product from the *N*-benzimidoylthiourea (4) was the same as that<sup>2</sup> from the *N*-benzoylthiourea, comparison of the experimental details argues strongly against this: the melting points are 150–153 °C<sup>2</sup> and 204 °C.<sup>3</sup> Signifi-



Scheme 1

<sup>†</sup> We are indebted to Dr. C. K. Prout, Chemical Crystallography Laboratory, Oxford University, for this study.

<sup>‡</sup> Spectrometric data: for (6), i.r. (CCl<sub>4</sub>, cm<sup>-1</sup>) 1635; u.v. [EtOH, nm (ε)] 252 (13 400), 344 (13 200); *m/z* (abundance) 308 (*M*<sup>+</sup>, 100), 293 (69), 43 (43); n.m.r. (CDCl<sub>3</sub>, δ), <sup>1</sup>H: 1.98 (CH<sub>3</sub>-CO), <sup>13</sup>C: 158.5 (t, *J* 3.8 Hz, C-4), 190.5 [q, *J* 6.1 Hz, C(O)]. For (8), i.r. (CCl<sub>4</sub>, cm<sup>-1</sup>) 1633; u.v. [EtOH, nm (ε)] 242 (10 700), 356 (16 700); *m/z* (abundance) 308 (*M*<sup>+</sup>, 100), 105(43), 77(60); n.m.r. (CDCl<sub>3</sub>, δ), <sup>1</sup>H: 2.41 (CH<sub>3</sub>-C-4), <sup>13</sup>C: 160.1 (q, *J* 6.8 Hz, C-4), 188.0 [t, *J* 3.6 Hz, C(O)].



cantly, Ried and Kaiser reported a C-Me  $^1\text{H}$  n.m.r. signal at  $\delta$  1.90 in their product<sup>3</sup> which, therefore, is almost certainly correctly formulated as the 5-acetyl compound (2), a conclusion in consonance with the work of Rajappo *et al.*<sup>4</sup> on the

$N'$ -monosubstituted thioureas. Other related compounds<sup>2,3</sup> will be discussed when the results obtained with a series of  $N$ -acyl- $N',N'$ -disubstituted thioureas and  $\alpha$ -halogenoketones<sup>7</sup> are presented.

Received, 5th April 1984; Com. 482

### References

- 1 S. Rajappo and B. G. Advani, *Ind. J. Chem.*, 1970, **8**, 1145; *ibid.*, *Sect. B*, 1978, **16**, 749.
- 2 J. Liebescher and H. Hartmann, *Z. Chem.*, 1974, **14**, 470.
- 3 W. Ried and L. Kaiser, *Leibigs Ann. Chem.*, 1976, 395.
- 4 S. Rajappo, M. D. Nair, B. G. Advani, R. Sreenivasan, and J. A. Desai, *J. Chem. Soc., Perkin Trans. I*, 1979, 1762.
- 5 I. B. Douglass and F. B. Dains, *J. Am. Chem. Soc.*, 1934, **56**, 719.
- 6 T. N. Birkinshaw, D. W. Gillon, S. A. Harkin, G. D. Meakins, and M. D. Tirel, *J. Chem. Soc., Perkin Trans. I*, 1984, 147.
- 7 Part II Theses of J. C. Brindley and J. M. Caldwell, Oxford, 1983.