Cycloaddition Reaction of Silver Ion Stabilized 6-Methoxycyclohepta-1-*cis*,-4-*trans*-diene

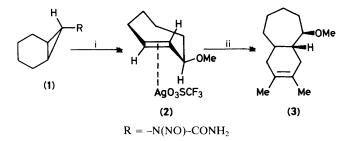
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The title olefin, stabilized by complexation to silver trifluoromethanesulphonate, was used in a stereospecific [4 + 2]-cycloaddition.

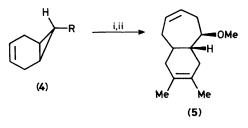
trans-Cycloheptenes are short-lived intermediates in photochemical isomerizations of the corresponding *cis*-olefins.¹ In methanolic solution at room temperature, *trans*-cycloheptene has a lifetime of 45 s^2 and *trans*-cyclohept-1-en-3-one a lifetime of $0.03 \text{ s}.^3$ Their double bond deformations⁴ induce a synthetically as well as mechanistically interesting, unique cycloaddition reactivity that is even further increased by additional ring contracting factors like endocyclic *cis*-double bonds or oxygen atoms.^{5,6} Even *trans*-cyclo-octene, that is orders of magnitude less reactive⁷ than *trans*-cycloheptene, has been used to achieve cycloaddition reactions with 1,3dipolar compounds that would not react with other olefins.^{8,9}



Scheme 1. i, AgO₃SCF₃, MeOH, 0 °C; ii, CH₂=C(Me)-C(Me)=CH₂, MeOH, 25 °C.

Unfortunately, the photochemical generation of *trans*cycloheptenes is not suitable for conducting cycloaddition reactions because of interference from numerous side reactions (*i.e.* dimerizations, solvent additions, fast *trans* \rightarrow *cis* isomerization, excitation of the cycloaddition partner, or cycloaddition of photoexcited *cis*-olefin).¹

Recently we prepared and utilized the silver trifluoromethanesulphonate (triflate) complex of stereospecifically substituted *trans*-cycloheptenes for some cycloaddition reactions (Scheme 1).¹⁰ In the sequence (1)—(3), the silver ion performs several functions: (a) it catalyses the deamination of norcarane nitrosourea (1) and renders a base unnecessary, (b)



 $R = -N(NO)-CONH_2$

Scheme 2. i, AgO_3SCF_3 , MeOH; ii, $CH_2=C(Me)-C(Me)=CH_2$, MeOH.

it stabilizes the *trans*-cycloheptene by complexation and allows for the isolation of (2),¹⁰ (c) it probably catalyses the [4 + 2]-cycloaddition of the *trans*-cycloheptene by pre-coordination of the diene.^{10,11} Conveniently, the more strained (5RS, 6RS)-6-methoxycyclohepta-1(Z),4(E)-diene can be used, avoiding the losses associated with the isolation of the silver complex.

Using a similar procedure to that in Scheme 1, compound $(4)^{12}$ was stirred with 2 equiv. of silver triflate (MeOH, 0 °C, dark, 24 h). The reaction mixture was filtered and the filtrate concentrated *in vacuo*. 1.2 Equiv. of 2,3-dimethylbuta-1,3-diene was added and the saturated methanolic solution stirred at 25 °C (dark, 24 h). The solvent and excess of diene were then removed *in vacuo*; product (5) was obtained in 65% yield (vacuum transferred at 25 °C) (Scheme 2).†

We are investigating the use of this method for the synthesis of natural products containing *trans*-annellated seven membered rings.

This work was supported by the Deutsche Forschungsgemeinschaft (DFG).

Received, 27th March 1984; Com. 424

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[†] Spectroscopic data for (5): 13 C n.m.r. (CDCl₃): δ 132.91, 126.34, 124.78, 124.74, 83.51, 56.20, 48.75, 41.46, 38.21, 37.52, 34.46, 29.80, 18.53, 18.44; 14 n.m.r. (CCl₄, 90 MHz): δ 5.94—5.40 (m, 2H), 3.20 (s, 3H, OCH₃), 2.83—2.60 (m, 1H, 2-H), 2.40—1.30 (m, 10H), 1.54 (s br., 6H, 2 × CH₃); i.r. (CCl₄): 3030, 1658(w), 1097(s), 717 cm⁻¹; *m/z* 206.16715 (*M*⁺, calc. 206.16707). The stereochemistry of the methoxy group in (5) corresponds to a preservation of the *trans, trans* relationship of the cyclopropyl protons in (4). See also: G. H. Whitham and M. Wright, *J. Chem. Soc. C*, 1971, 883; C. B. Reese and A. Shaw, *J. Chem. Soc., Perkin Trans.* 1, 1975, 2422; W. Kirmse and H. Jendralla, *Chem. Ber.*, 1978, 111, 1857.