Stereocontrol at 'Off-template' Sites in 1,2-O-Isopropylidene Glycofuranoses

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Hydrogenation of the double bond at C(5) of diester (12) occurs exclusively from the *si-si* face provided that a bulky substituent is present at C(3) in *cis* relationship to the olefinic side chain.

A research programme in our laboratory is concerned with the use of 'diacetone glucose' (1a) as a latent α -methylene lactone in the synthesis of sesquiterpene lactones.¹ We reported recently that the Dieckmann cyclization of diester (2) gave a 9:1 mixture of the regioisomers (3) and (4). From the sesquiterpene numbering shown in Scheme 1 it is evident that the minor product (4) would be more compatible with most sesquiterpene skeleta since (i) C(10) is activated for elaboration of the A ring and (ii) the methoxycarbonyl group would represent the ubiquitous C(10) angular Me group. The desired regioselectivity would be ensured with the lactonic ester (6) since pathway (a) is much preferred to (b) which would involve a tetrahedral intermediate containing an impossible trans-fused [3.2.1] bicyclic system. To this regiocontrolling constraint, a tandem stereocontrolling factor could be added if the C(5) stereocentre of (6) were of (S) configuration, for the cis-fused 6-5 system (7) would then be favoured overwhelmingly vis a vis its trans-counterpart. This analysis ultimately requires stereocontrol at C(5) of (5) which, being an 'off-template' site, is not readily achieved.² We report here on success in using this objective to form (7) as well as in the formation of an unrelated system reported in the following communication.3

According to the plan in Scheme 1, the C(3) and C(5) centres of 'diacetone glucose' (1a) are to be equipped with two-carbon side chains, and it would be easier to deal with C(3) first since (8) is a known compound⁴ (Scheme 2). However, hydrogenation of (9), obtained as the only geometric isomer, gave a *ca.* 3:2 mixture of (5) and its 5-epimer.

The alternative sequence was therefore examined beginning with benzoate (1b) from which ketone (10) was obtained by routine transformations. Wittig reaction in acetonitrile afforded the geometric isomers (11) and (12) in the ratios of 5:1(n.m.r. spectroscopic estimation). However, with dimethylsulphoxide (DMSO), (11) was obtained exclusively. The configurational assignments were based on the fact that upon desilylation of the mixture, the minor component [(12)] formed the butenolide (13).

Surprisingly, attempts to hydrogenate the silylated compounds (11a) and (12), using palladium and platinum at varying pressures of hydrogen, were unsuccessful. However, upon desilylation, hydrogenation of (11b) over platinum proceeded smoothly to give a mixture of (14) and (15). Treatment of this mixture with a catalytic amount of toluenep-sulphonic acid afforded pure (15). In contrast, desilylation of (12) yielded the butenolide (13) directly which, upon hydrogenation, furnished a mixture of (15) and its 5-epimer in a ratio of 5.5:1.

The high stereoselectivity in the hydrogenation of (11) [in comparison with (9) or (13)] can be rationalized on the grounds that the rotamer shown in Scheme 2 is distinctly favoured since the methoxycarbonyl group is 'away' from the furan ring. In this conformation, the bulky benzoate ester group shields the *si-si* face of (11) thereby forcing hydrogenation to yield the (S) configuration at C(5) of (14). The results suggest that this conformational bias does not exist in the butenolide (13).



Scheme 1

The furano-lactone (15) obtained as indicated in Scheme 2 was then processed to give the desired lactonic ester (6), cyclization of which afforded keto-lactone (7) in 91% yield. The structure of (7) was apparent from the ¹H n.m.r. parameters for H(7) (δ 3.47, $J_{5,7}$ 6.4 Hz) and H(4) (δ 4.08, $J_{3,4}$ 11.6, $J_{4,5}$ 6.8 Hz). Alkylation of (7) proceeded very smoothly at room temperature in 15 min to give (16) in which the H(4) ¹H n.m.r. parameters were again consistent with the *cis*-ring junction (δ 3.97, $J_{3,4}$ 11.0, $J_{4,5}$ 7.5 Hz).



Scheme 2. i, H_3O^+ ; ii, Bu^tMe_2SiCl (TBDMSCl); iii, pyridinium dichromate; iv, Ph_3PCHCO_2Et ; v, H_2 ; vi, Bu^n_4NF ; vii, Bu^tOK , PhH; viii, Bu^tOK -DMSO-MeI.

The structure of (16) was further confirmed by an X-ray analysis.

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References

- 1 K. M. Sun, B. Fraser-Reid, and T. F. Tam, J. Am. Chem. Soc., 1980, **104**, 367; T. F. Tam and B. Fraser-Reid, J. Chem. Soc., Chem. Commun., 1980, 556.
- 2 H. Redlich and H-J. Neumann, Chem. Ber., 1981, 114, 2020.
- 3 D. Liang, H. W. Pauls, and B. Fraser-Reid, following communication.
- 4 A. Rosenthal and L. Nguyen, J. Org. Chem., 1969, 34, 1029.