Penicillin Biosynthesis: Multiple Pathways from a Modified Substrate

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Preparations of the enzyme isopenicillin N synthetase from *Cephalosporium acremonium* CO 728 convert the modified tripeptide substrate δ -(L- α -aminoadipoyl)-L-cysteinyl-D-allylgycine† into penam, homoceph-3-em, hydroxy-cepham, and hydroxyhomocepham type products; these have been isolated and their structures established.

The conversion of the natural precursor δ -(L- α -aminoadipoyl)-L-cysteinyl-D-valine (**1a**) into isopenicillin N (**2a**) by a cell-free extract of isopenicillin N synthetase from *Cephalosporium acremonium* is established.¹ Recently we reported² that the

+ δ -(L- α -aminoadipoyl) = 5-(5S)-amino-5-carboxypentanoyl.

conversion of a modified substrate δ -(L- α -aminoadipoyl)-Lcysteinyl-D-(α -aminobutyrate) (1b) with a highly purified sample of the enzyme isopenicillin N synthetase occurs *via* dual pathways into both penam (2b) and cepham (3a) type products. We now report that the conversion of a similarly modified tripeptide δ -(L- α -aminoadipoyl)-L-cysteinyl-Dallylglycine (1c) with this purified enzyme at pH 7–8 leads



 $\dot{c}O_2^ O^ \dot{c}O_2^ \dot{c}O_2^-$

simultaneously to six distinct β -lactam containing products (2c), (2d), (3b), (4a), (4b), and (5) (Figure 1).

The first of these products was the vinylpenicillin (2c) which showed penicillin type antibacterial activity[‡] against Grampositive organisms (*Staphylococcus aureus* N.C.T.C.6571, *Sarcina lutea* 400E, and *Bacillus megaterium* 1571E) but had no effect upon Gram-negative species (*Alcaligenes faecalis* DS.367, *Pseudomonas aeruginosa* N.C.T.C.10701, *Escheri*- schia coli 2110E, and Salmonella typhi DS.48) at a concentration of 50 μ g ml⁻¹. The structure§ (2c) follows from its ¹H n.m.r. spectrum, $\delta_{\rm H}$ (²H₂O, 500 MHz)¶ 1.70–1.93 (4H, m, CH₂CH₂CH₂CO), 2.43–2.45 (2H, m, CH₂CO), 3.74–3.76 (1H, m, CH[CH₂]₃), 4.63 (1H, d, J 3.3 Hz, 3-H), 5.18 (1H, d, J 10 Hz, vinyl-H), 5.36 (1H, d, J 17 Hz, vinyl-H), 5.45, 5.47 (2H, AB q, J 3.8 Hz, 5,6-H), and 5.88-5.95 (1H, m, vin, 1-11), (2-H obscured by ²HOH). The coupling constant $J_{2,3}$ 3.3 Hz for 3-H was in agreement with a β -configuration for the vinyl group at C(2) and the penam (2c) also gave the expected molecular ion MH^+ m/z 358 upon positive argon fast-atom bombardment (F.A.B.) mass spectrometry. The vinylpenicillin (2d) with an α -configuration was also isolated as a minor product [ratio $(2c):(2d) = 4:1], \delta_{H} (^{2}H_{2}O, 500 \text{ MHz}) 1.71-1.93 (4H, m,$ CH₂CH₂CH₂CO), 2.42—2.44 (2H, m, CH₂CO), 3.75—3.77 (1H, m, CH[CH₂]₃), 5.20 (1H, d, J 10.5 Hz, vinyl-H), 5.34 (1H, d, J 17.5 Hz, vinyl-H), 5.47, 5.58 (2H, AB q, J 3.8 Hz, 5,6-H), and 5.75-5.81 (1H, m, vinyl-H), (2,3-H obscured by ²HOH). (2d) showed the expected penicillin type antibacterial activity against S. aureus N.C.T.C.6571.

The second class of β -lactam products isolated was the homoceph-3-ems (4a,b) [ratio (4a): (4b) = 10:1], whose structures were established by a combination of n.m.r. and mass spectrometry. For (4a), $\delta_{H}\,(^{2}H_{2}O,\,500$ MHz) 1.71–1.95 $(4H, m, CH_2CH_2CH_2CO), 2.43 - 2.47 (2H, m, CH_2CO), 3.06 (1H, dd, J 8, 16 Hz, 2-H), 3.53 (1H, dd, J 6, 16 Hz, 2-H),$ 3.73-3.76 (1H, m, CH[CH₂]₃), 5.20 (1H, m, 5-H), 5.30, 5.48 (2H, AB q, J 4 Hz, 7,8-H), 5.78 (1H, dd, J 5, 11.5 Hz, 4-H), and 6.04–6.10 (1H, m, 3-H); m/z (positive argon F.A.B.) *M*H⁺ 358; for (**4b**) as (**4a**) except $\delta_{\rm H}$ 5.24, 5.31 (2H, AB q, J 4 Hz, 7,8-H), and 5.95-6.03 (2H, m, 3,4-H); a Jeener n.m.r. spectrum** established the connectivity S-CH2-CH=CH- $CH(CO_2H)$ - for (4a) and (4a) showed no antibacterial activity against S. aureus N.C.T.C.6571 at a concentration of 400 µg ml-1. A similar lack of antibacterial activity has been reported for a homoceph-4-em.³ When a homogeneous sample of (4a) was resuspended in buffer (pH 7-8) some equilibration to (4b) occurred (by 1H n.m.r. and h.p.l.c.); likewise, equilibration of (4b) to (4a) was also observed. A similar experiment in deuteriated media at pH 8-9 gave the 5-[2H]-labelled homoceph-3-em (4a), $\delta_{\rm H}$ 5.36, 5.56 (2H, AB q, J 4 Hz, 7,8-H), 5.85 (1H, d, J 11 Hz, 4-H), and 6.12-6.16 (1H, m, 3-H), (5-H absent), and 5-[²H]-labelled homoceph-3-em (4b), $\delta_{\rm H}$ 5.30 $(1H, d, J 4 Hz, \beta$ -lactam-H) and 6.03 (1H, d, J 11 Hz, 4-H), (other resonances obscured), ratio (4a): (4b) ca. 9:1. A preference for an α -carboxylate over a β -carboxylate under equilibrating conditions has precedent.4

The structure of (**3b**), representing the third class of β -lactams was established as a hydroxymethylcepham from its n.m.r. and mass spectra, $\delta_{\rm H}$ (²H₂O, 500 MHz), 1.71–1.95 (4H, m, CH₂CH₂CH₂CO), 2.14–2.22 and 2.35–2.40 (2H, m, 3-H), 2.43–2.46 (2H, m, CH₂CO), 3.21–3.26 (1H, m, 2-H), 3.73–3.76 (3H, m, CH[CH₂]₃ and CH₂OH), 4.36–4.38 (1H, m, 4-H), and 5.24–5.34 (2H, AB q, J 4 Hz, 6,7-H); *m/z* (positive argon F.A.B.) *M*H⁺ 376. A Jeener n.m.r. spectrum established the connectivity S–CH(CH₂OH)–CH₂–CH(CO₂H); irradiation of the C(2)–CH₂OH group, $\delta_{\rm H}$ 3.74 generated a nuclear Overhauser enhancement (n.O.e.) of 6-H consistent with an α -configuration at the methylenehydroxy

[‡] The specific activity of (2c) against *Staphylococcus aureus* N.C.T.C.6571 was comparable to that of isopenicillin N, and was destroyed by β -lactamase I from *Bacillus cereus*.

[§] For (2c), (2d), (3b), and (4a), an α -configuration for the carboxyl group is assumed, not proven.

[¶] Chemical shifts are referenced to internal sodium $(2,2,3,3-^{2}H_{4})-3-$ trimethylsilylpropionate (TSP) = 0.00 p.p.m.

^{**} Further details of these experiments will be reported elsewhere.



Figure 1. ¹H N.m.r. spectrum (500 MHz, δ_H 5.1–6.1) of a crude incubation mixture [(1c) with isopenicillin N synthetase] after protein precipitation. The characteristic β -lactam resonances are labelled. The ratio (2c):(2d):(3b):(4a):(4b):(5) was approximately 4:1:2:10:1:5. These products were subsequently purified by preparative electrophoresis (at pH 3.5) and by h.p.l.c. (reverse phase O.D.S.column, pH 8 buffer).



Scheme 1

group. Conversely, irradiation of the C(2)-H group, δ_H 3.23 gave no n.O.e. to 6-H. The hydroxymethylcepham (**3b**) showed no antibacterial activity against *S. aureus* N.C.T.C.-6571 at a concentration of 100 µg ml⁻¹.

The fourth class of product was established as the hydroxyhomocepham (5) from its n.m.r. and mass spectra, δ_H (²H₂O, 500 MHz) 1.70–1.97 (4H, m, CH₂CH₂CH₂CO), 2.09–2.17 and 2.30–2.34 (2H, m, 4-H), 2.42–2.45 (2H, m, CH₂CO),

2.76-2.80 and 3.03-3.06 (2H, m, 2-H), 3.73-3.76 (1H, m, CH[CH₂]₃), 4.09-4.14 (1H, m, 3-H), 4.18 (1H, dd, J 2.5, 12.5 Hz, 5-H), and 5.26, 5.38 (2H, AB q, J 4 Hz, 7,8-H); m/z (positive argon F.A.B.) MH+ 376. A Jeener n.m.r. spectrum established the connectivity S-CH2-CH(OH)-CH2-CH- (CO_2H) – and the product (5) showed no antibacterial activity against S. aureus N.C.T.C.6571 at a concentration of 300 µg ml⁻¹. The stereochemistry of the hydroxy group as $3-\alpha$ -OH was established by n.O.e. experiments. Thus irradiation of the higher field C(4)-H ($\delta_{\rm H}$ 2.78) gave a n.O.e. (8%) to 7-H but not to 3-H, whereas irradiation of the lower field C(4)-H ($\delta_{\rm H}$ 3.05) gave no enhancement of 7-H but instead n.O.e.'s to 3-H(13%)and 5-H (14%). A similar experiment was performed on the C(2)-H's. Thus irradiation of the high field C(2)-H ($\delta_{\rm H}$ 2.13) gave n.O.e.'s to 7-H (5%) and 3-H (3%) whereas irradiation of the lower field C(2)-H ($\delta_{\rm H}$ 2.32) gave n.O.e.'s to 5-H (3%) and 3-H (8%) (in all these experiments large geminal n.O.e.'s were observed: 20–25%). These results are consistent with formulation of the structure as (5).

In summary we suggest that the diverse structural range of products formed from the allylglycine tripeptide (1c) with isopenicillin N synthetase may be explained by two discrete pathways (routes a, b) from a common monocyclic β -lactam intermediate (6)⁵ (Scheme 1). From route a, a radical (or carbon-metal) type intermediate (7) would lead, on ring closure, to the penicillin products (2c) and (2d) whilst the rearranged radical (8) would give the homoceph-3-em (4a). The preference of (2c) over (2d) has precedent² in a related system, whilst the predominance of (4a) over (2c) and (2d) probably reflects a lower energy barrier for the second ring closure. The product (4b) probably arises by epimerisation of the biosynthetic (4a). The novel hydroxylated products (3b) and (5) are thought to arise by a different mechanism, route b. Whether the source of the oxygen atom [*e.g.* at C(3) of (5)] is the co-substrate dioxygen⁶ or, alternatively, water is currently under investigation.

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