9,21-Didehydroryanodine: a New Principal Toxic Constituent of the Botanical Insecticide Ryania

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A major insecticidal constituent of *Ryania speciosa* is isolated and characterized by n.m.r. and mass spectroscopy as the new alkaloid 9,21-didehydroryanodine (2); reduction of (2) with hydrogen (or tritium) gas yields both ryanodine (1) and 9-epiryanodine (3) (or their 9,21-³H analogues) in a 1:9 ratio.

Ryanodine (1), an insecticidal and toxic alkaloid from *Ryania* speciosa, was structurally defined about 15 years ago by chemical degradation¹ and X-ray crystallography.² Compound (1) has been considered to be the principal active constituent of the ground stemwood of this plant, which is used as ryania powder to control agricultural and garden pests.³ Alkaloid (1) causes muscle contraction at very low concentration, possibly by dissociating the transverse tubules from the terminal cisternae of the sarcoplasmic reticulum.⁴ Further investigation of ryanodine mode of action is presently hampered by (i) the presence of an unknown major contaminant in commercial ryanodine⁵ and (ii) the lack of ryanodine radiolabelled to *high* specific activity. The present study identifies the contaminant (2) and uses this new natural product to prepare [³H] (1).

Preparative h.p.l.c.† of commercial ryanodine‡ yielded two major components, (1) m.p. 235–237 °C (lit.⁶ 229 °C) and (2) m.p. 175–179 °C. A fresh sample of powdered *R. speciosa* wood‡ was extracted with wet chloroform by partitioning with water⁷ and the aqueous extract was directly purified by h.p.l.c.† This established the presence of (2) and (1) in a 3:1 ratio totalling about 0.2% (w/w) in content. Bioassays



revealed that (1) and (2) are almost equitoxic on injection into houseflies and mice. On this basis (2) rather than (1) appears to be the principal toxic constituent in the ryania sample examined.

Analysis of the ¹H (Table 1) and ¹³C n.m.r. spectra of (1) and (2) using one- and two-dimensional methods (details to be given elsewhere) established the identity of the new compound as 9,21-didehydroryanodine. In the ¹H n.m.r. spectrum of (1), there are five methyl resonances (17-H to

[†] Altex Ultrasphere-ODS 4.6×250 mm; H₂O:MeOH, 1:1, 1 ml/ min; 268 nm detection. Retention time (min) (2) 7.6, (1) 9.6, and (3) 11.4.

[‡] Ryanodine was obtained from Penick Corp., Lyndhurst, New Jersey, and a fresh sample of powdered *R. speciosa* wood (ryania insecticide) was a generous gift from Progressive Agri-Systems, Inc., Stockertown, Pennsylvania 18083. In both cases the *R. speciosa* source was Trinidad.

	Table 1.	Selected ¹ I	H n.m.r. s	pectroscop	y δ values	(300 MH	z, CD_3OD).
	3-H	10-H	17-H	18/19-H		20-H	21-H	21-H ₂
(1)	5.64s	3.80d	1.40s 1.40s	0.76d 0.75d	1.12d 1.10d	0.90s 0.90s	1.01d	4.74d, 5.04d
(2) (3)	5.63s	4.32d	1.41s	0.76d	1.13d	0.91s	1.12d	

21-H) and two distinctive methine signals: a doublet for 10-H and a singlet for 3-H. In the case of (2), one of the methyl doublets and the 10-H methine resonance near δ 3.8 are missing. These resonances are replaced by three new peaks in the δ 4.7—5.5 region, two of which exhibit a 2 Hz coupling. The rest of the spectrum is largely unchanged. The coupled peaks were assigned as the terminal methylene protons, and the new downfield singlet as the now allylic 10-H. The ¹³C n.m.r. spectra support this assignment; two new low field resonances appear at δ 149.5 and 107.5, and in a DEPT⁸ experiment ($\theta = 3\pi/4$) the δ 149.5 resonance is nulled and the δ 107.5 resonance becomes negative. Additionally fast atom bombardment mass spectroscopy shows a parent ion two mass units less than for (1). The i.r. spectrum for (2) is similar to that of (1), but is characterized by an additional C=C absorbance at 1545 cm⁻¹ (KBr).

Compound (2) was reduced (H₂/Pd on carbon, tetrahydrofuran) to yield the expected (1) and 9-epiryanodine (3) m.p. 232—234 °C in a 1:9 ratio. Similar reduction with tritium gas enabled us to prepare [9,21-³H] (1) at high specific activity (60 Ci mm⁻¹) complementing an earlier synthesis⁹ of [pyrrole-³H] (1) (3 Ci mm⁻¹).

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