

Corrigendum**Stabilization by a Strongly Acidic Medium of Trivalent Copper Tetra-aza Macrocyclic Complexes**

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The calculated value of $E_{\frac{1}{2}}(\text{Fe}^{3+}/\text{Fe}^{2+})$ in 10 M HClO_4 is 0.804 V vs. normal hydrogen electrode (N.H.E.). Accordingly, the $E_{\frac{1}{2}}$ values for the $\text{M}^{\text{III}}/\text{M}^{\text{II}}$ redox couple in tetra-aza macrocyclic complexes, wrongly reported in the diagram in Figure 2, are: $[\text{Cu}\cdot(1)]$ 1.14 V vs. N.H.E., $[\text{Cu}\cdot(2)]$ 1.18, $[\text{Cu}\cdot(3)]$ 1.01, $[\text{Ni}\cdot(1)]$ 1.20, $[\text{Ni}\cdot(2)]$ 1.10, $[\text{Ni}\cdot(3)]$ 0.89.