Heteroannulation of 4-Oxochromene-3-carbaldehyde Derivatives using Halogenoalkanols and Base

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4-Oxochromene-3-carbaldehyde derivatives react with 2-iodoethanol and potassium carbonate in acetone to produce tetrahydrofuro[2,3-b][1]benzopyran-4-ones.

4-Oxochromene-3-carbaldehydes are highly functional molecules, capable of reacting as heterodienes,¹ cuprate acceptors,² or dienophiles,³ and undergoing a variety of rearrangements.⁴ Herein we describe a new transformation of these versatile heterocycles, namely their conversion to the hitherto unknown tetrahydrofuro[2,3-*b*][1]benzopyran-4-one system *via* a one-step heteroannulation procedure.

Treatment of 6-hydroxy-4-oxochromene-3-carbaldehyde (1a)⁵ with 2-iodoethanol under standard alkylation conditions gave a major product, isolated by extraction and crystallisation, of m.p. 174—175 °C (ethyl acetate–light petroleum, 1:1). From spectral data (v_{max} 3425 and 1670 cm⁻¹; *m/z* 206) it was apparent that the expected hydroxyethylation had not occurred, but that a chromanone had been formed with concomitant deformylation. The ¹H n.m.r. spectrum of the product [δ (CDCl₃), 2.0—2.5 (m, 2H), 2.9—3.2 (m, 1H), 4.0—4.4 (m, 2H), 4.9 (s, 1H, D₂O exchangeable), 5.9 (d, 1H, *J* 4Hz), and 6.9—7.3 (m, 3H)] was consistent with the tetrahydrofuro[2,3-b][1]benzopyran-4-one structure (2a).

With other derivatives of (1)^{6.7} analogous results were obtained (Table 1). Each product displayed the spectral signals characteristic of the acetal hydrogen [δ 5.9—6.0 (d, 1H, J 4—4.5 Hz)] and chromanone carbonyl group (v_{max} . 1680—1695 cm⁻¹). The reaction of the 6-methoxy derivative (1c) was also efficient using 2-bromoethanol, but was less so using 3-iodopropanol,⁸ which furnished the tetrahydropyrano[2,3-b][1]benzopyran-5-one (3) in only 25% yield



despite more vigorous reaction conditions. The ¹H n.m.r. spectrum of the latter [δ 5.5 (d, 1H, J 3 Hz)] is consistent with the *cis* arrangement depicted.⁹

The heteroannulation is probably initiated by addition of the halogenoalkanol to the 2-position of compound (1). This

 Table 1. Reactions of 4-oxochromene-3-carbaldehyde derivatives (1)

 with halogenoalkanols and base.

Derivative				
of (1) ^a	R	Halogenoalkanol	Product	% Yield ^b
(1a)	OH	I[CH ₂] ₂ OH	(2a)	65
(1b)	Н	I[CH ₂] ₂ OH	(2b)	53°
(1c)	OMe	$I[CH_2]_2OH$	(2c)	70
(1c)	OMe	Br[CH ₂] ₂ OH	(2c)	60
(1d)	NO_2	ICH ₂ ₂ OH	(2d)	40°
(1c)	OMe	ICH ₂ ₃ OH ^d	(3)	25°

^a Reaction conditions: compound (1) (1 mmol), the halogenoalkanol (3 mmol), and anhydrous potassium carbonate (0.3-0.5 g) were heated in acetone (30 ml) under reflux for 4 h. Products were isolated by evaporation, extraction with ethyl acetate, and crystallisation. ^b Yields refer to isolated chromatographically homogeneous material. All products were characterised by elemental analysis and spectroscopy. ^cPurified by chromatography over silica gel. ^d Using halogenoalkanol (8 mmol) and anhydrous potassium carbonate (2 g) in acetone (30 ml), and heating under reflux for 72 h.

produces an intermediate (4) capable of base-induced intramolecular alkylation, which gives (5). Subsequent deformylation, also promoted by base, leads to the observed product. We thank the \tilde{S} .E.R.C. and Glaxo Group Research, Greenford, for a C.A.S.E. studentship.

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