## Phosphocations with Co-ordination Number 2 and 3: Phospheniumiminophosphonium Bisphosphocations

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New phosphocations containing phosphorus with co-ordination number 2 and 3 and deriving from  $CI-\tilde{P}-NR_2 AICI_4^-$  have been obtained: chloroiminophosphonium compounds (2) and the bisphosphocations (phosphenium-iminophosphoniums) (3).

In a prior investigation<sup>1a</sup> we have shown that azides RN<sub>3</sub> react with the bis(dialkylamino)phosphenium compound  $(R_2N)_2P^+$ AlCl<sub>4</sub> to give the corresponding bis(dialkylamino) iminophosphonium  $(R_2N)_2P^+=NR$ . Five cations of this latter type were described. Several experiments show that this extension of the Staudinger reaction actually occurs by reaction of the phosphenium cation and not the initial chlorophosphine Cl-P(NR\_2)<sub>2</sub> which was used to prepare  $(R_2N)_2P^+$  by Cl<sup>-</sup> abstraction.

To determine the scope of this new reaction, we have explored the reactivity of the chlorophospheniums  $Cl_{-}P_{-}NR_{2}$  (1). Few such entities are known<sup>2</sup> and we used  $Pr_{2}N_{-}P_{-}Cl$  (1a) and  $Me_{2}N_{-}P_{-}Cl$  (1b). In the traditional Staudinger reaction, the replacement of the  $R_{2}N$  group by Cl on the tricovalent phosphorus usually strongly decreases and even

$$R_2N + NPh AICI_4^{-}$$
(2)  
a; R = Pr<sup>i</sup>  
b; R = Me

inhibits the formation of iminophosphine,<sup>3</sup> so the reaction of (1) with azides  $(RN_3)$  seems worthy of study.

The reaction of phenyl azide on chlorophospheniums (1) in stoicheiometric amounts, gives the expected evolution of  $N_2$  and leads to the chloroiminophosphoniums (2). In the absence of accurate kinetic data, we noted qualitatively that the reaction occurs as quicky as with  $(R_2N)_2P^+$  cations. Com-

pound (2a) has been isolated as a solid and the structure has been well established by n.m.r. spectroscopy: <sup>31</sup>P (36.44 MHz)  $\delta$  + 34.6 p.p.m. (t, <sup>3</sup>J<sub>P-H</sub> 25.5 Hz); <sup>1</sup>H 60 MHz)  $\delta$  1.33 (6H, d, <sup>3</sup>J<sub>H-H</sub> 7 Hz, Me<sub>2</sub>C), 4.1 [2H, hept. d, <sup>3</sup>J<sub>H-H</sub> 7, <sup>3</sup>J<sub>H-P</sub> 26 Hz, N(CHMe<sub>2</sub>)<sub>2</sub>].

The course of the reaction is the same for (1b), giving (2b) which has been identified by <sup>31</sup>P n.m.r. [ $\delta$  43.8 p.p.m. (hept., <sup>3</sup>J<sub>P-H</sub> 15.5 Hz)], but the adduct Cl<sub>2</sub>(Me<sub>2</sub>N)P=N-Ph,AlCl<sub>3</sub> is also present in the solution ( $\delta$  46 p.p.m.).<sup>1b</sup>

With trimethylsilylazide ( $Me_3SiN_3$ ) the reaction of (1) is not predictable due to the fact that the Staudinger reaction is in competition with possible elimination of  $ClSiMe_3$ .<sup>4</sup>

Several experiments have been performed; at room temperature, addition of  $Me_3SiN_3$  to (1a) in solution ( $CH_2Cl_2$ ) leads to an unexpected result: isolation of a solid and stable phosphenium-iminophosphonium cation (3a).

The structure of this first bisphosphocation is consistent with <sup>31</sup>P n.m.r. signals for the two kinds of P atoms [ $\delta$  P(1) 311.3, P(2) 27 p.p.m.,  ${}^{2}J_{P-N-P}$  111.8 Hz]. Furthermore the trico-ordinated  $P(2)^+$  nucleus is coupled with the CH isopropylic group (t,  ${}^{3}J_{P(2)-H}$  29 Hz). The other n.m.r. data confirm the unique dicationic structure:  ${}^{27}A1$  (23.47 MHz)  $\delta$ 104.9 p.p.m. (narrow singlet characteristic of AlCl<sub>4</sub><sup>-</sup> anion); <sup>1</sup>H (90.00 MHz) (Me<sub>2</sub>CH)<sub>2</sub>N–P(1)<sup>+</sup> group  $\delta$  3.48(2H, hept., <sup>3</sup>J<sub>H-H</sub> 6.9 Hz, CH) and 1.48 (6H, d, <sup>3</sup>J<sub>H-H</sub> 6.9 Hz, Me),  $(Me_2CH)_2N-P(2)^+$  group  $\delta$  3.73 and 3.89 (2H, 2 hept. d,  ${}^{3}J_{H-H}$  6.8,  ${}^{3}J_{H-P}$  29.0 Hz, CH);  ${}^{13}C$  (22.63 MHz) (Me<sub>2</sub>CH)<sub>2</sub>N-P(1)+ group & 48.74 (2C, s, CH) and 19.43 (4C, s, Me),  $(Me_2CH)_2N-P(2)^+$  group  $\delta$  50.3 (1C, d,  ${}^2J_{C-P}$  8.8 Hz, CH) 49.5 (1C, d, <sup>2</sup>J<sub>C-P</sub> 7.4 Hz, CH), and 21.6 p.p.m. (4C, s, Me). The bisphosphocation (3b) can be unambigously detected in solution [ $\delta$  P(1) 293.5, P(2) 30 p.p.m.,  ${}^{2}J_{P-N-P}$  90.8,  ${}^{3}J_{P(2)-H}$ 15.6 Hz]. However this compound is less stable than (3a) and we were unable to isolate it in pure form.

Production of biscation (3) can be explained by the route shown in Scheme 1. We have some experimental arguments to support the pathway. We have tried to obtain (4a) by a traditional method according to Scheme 2.

The phosphenium azide intermediate (4a) reacting upon itself gives the bisphosphocation (5a) [ $\delta$  P(1) 301.4, P(2) 20.5 p.p.m.,  ${}^{2}J_{P-N-P}$  93,  ${}^{3}J_{P(2)-H}$  23.8 Hz;  $\nu$ (N3) 2160 cm<sup>-1</sup>] and polymers.

Note that (5a) is different from (3a) only by the replacement



of Cl by an N<sub>3</sub> group. So Schemes 1 and 2 demonstrate the ability of the phosphenium cations  $(R_2N-P-Cl \text{ or } R_2N-P-N_3)$  to react with the phosphenium azide (4). Furthermore when the reaction in Scheme 1 is made unfavourable by changing the order of addition: *i.e.* (1a) is added to Me<sub>3</sub>SiN<sub>3</sub>, both bisphosphocations (3a) and (5a) are present in the reaction mixture.

In conclusion this communication proves that the extension of the Staudinger reaction to the chlorophosphenium cation is feasible; in the case of trimethylsilylazide, a bisphosphocation has been prepared for the first time and the particular phosphenium azides (4) appear as interesting intermediate sources of polyphosphocations.

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