

Stereoselective γ -*cis*-Vinyllic Metallation of Tertiary Allylic Alcohols

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Tertiary allylic alcohols are converted by Bu^nLi -tetramethylethylenediamine into the γ -*cis*-vinyl-lithium derivatives which can be alkylated in moderate yields.

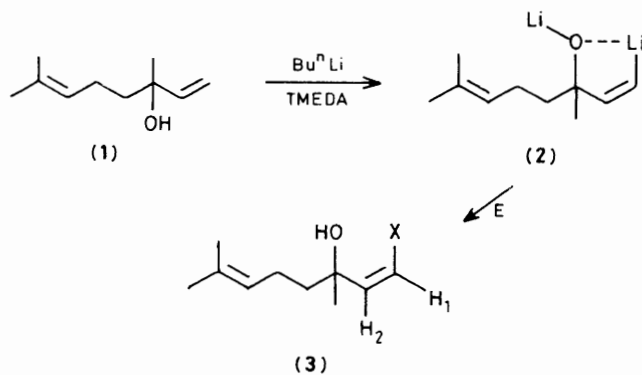
Olefins can be metallated in an allylic position and heteroatoms have been shown to have a profound influence on the ease and direction of metallation¹ mainly of aromatic substrates.²

Linalo-ol (1) was treated with Bu^nLi -tetramethylethylenediamine (TMEDA) in hexane and the reaction mixture quenched with D_2O (Scheme 1). ^1H N.m.r. spectroscopy showed that the product was quantitatively and stereoselectively ($J_{1,2} = 11$ Hz), deuteriated in the γ -*cis*-vinyllic position (3; X = D).

There are, to our knowledge, few examples of the substitution by a metal of a vinyllic hydrogen in non-activated olefins.³ Stabilisation of the metallation product by intramolecular co-ordination seems reasonable.

The reaction of the intermediate (2) with a variety of electrophiles was investigated: methyl bromide leads to 4,8-dimethylnona-2,7-dien-4-ol (3; X = Me) (55% yield; 98% Z, 2% E). The yields with other electrophiles are moderate to fair but stereochemistry is clearly Z (Table 1). The reaction with higher alkyl halides had to be catalysed with transition metal complexes.

The vinyl-lithium intermediate (2) thus appears less reactive than normal vinyl-lithium derivatives. A similar observation has been made with the Z addition product of Grignard



Scheme 1

Table 1. Metallation and alkylation of linalo-ol (1).

E	X ^j	% Yield	$J_{1,2}/\text{Hz}$
D_2O	D	72 ^{a,b}	11
MeBr	Me	55 ^{c,d}	12
$\text{Me}_2\text{CH}[\text{CH}_2]_2\text{Br}$	$\text{Me}_2\text{CH}[\text{CH}_2]_2$	5 ^e	g
		28 ^{a,e}	
		35 ^{a,f}	
Me_3SiCl	Me_3Si	52 ^a	15.5 ^h
$\text{Me}_2\text{C}=\text{O}$	Me_2COH	51 ^a	13
CO_2	$-\text{C}(\text{O})-\text{O}^-$	48 ^a	7

^a Isolated yield. ^b Deuteriation is quantitative by ^1H n.m.r. or mass spectrometry. ^c G.l.c. yield. ^d 98% Z, 2% E by capillary g.l.c. ^e Catalyst: $\text{CoCl}_2(\text{PPh}_3)_2$, 5%. ^f Catalyst: $\text{Fe}(\text{PhCOCHCOPh})_3$, 2%. ^g Could not be determined by ^1H n.m.r. at 250 MHz. ^h In agreement with literature value for silylated compounds (ref. 6). ⁱ Lactone. ^j All new compounds gave analytical results (^1H , ^{13}C n.m.r., m.s.) in agreement with their structure.

reagents to prop-2-ynylic alcohols,⁴ the E products being much more reactive.⁵

A similar stereoselective metallation could be observed with dihydrolinalo-ol and *trans*-nerolidol.

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