## Synthesis of the First Examples of 1*H*- and 4*H*-1,2,4,3 $\lambda$ <sup>3</sup>-Triazaphospholes *via* Unexpected Ring-contraction Reactions

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The reaction of the Schiff bases of benzamidrazone or diaminomaleonitrile with bis(dimethylamino)chlorophosphine or tris(dimethylamino)phosphine affords new 1H-, 2H-, or 4H-1,2,4,3 $\lambda^3$ -triazaphospholes or 1,3,2 $\lambda^3$ -diazaphospholes with the formation of triaza- or diaza-phosphorines as intermediates.

There has been increasing interest recently in the synthesis and reactivity of unusually hybridized phosphorus compounds.¹ Until now only a few examples of five-membered ring compounds containing two-co-ordinated phosphorus with an >N-P=N- skeleton have been described.².³ To the best of our knowledge analogous six-membered-ring compounds have not yet been reported. The reaction of a Schiff base (1) or (11) with a phosphorus derivative such as bis(dimethylamino)chlorophosphine (2) or tris(dimethylamino)phosphine (6) would be expected to give this type of compound.

However, unexpected ring-contraction reactions, which occurred surprisingly with phosphorus—carbon bond cleavage, were observed instead in these cases, leading to the first examples of 1H- and 4H- $1,2,4,3\lambda^3$ -triazaphospholes or to  $1,3,2\lambda^3$ -diazaphospholes. We also report here: (i) the synthesis of a new 2H- $1,2,4,3\lambda^3$ -triazaphosphole,† and (ii) the full characterisation of a derivative of the previously unknown 3,4-dihydro- $1,4,2\lambda^5$ -diazaphosphorine (12) as a stable intermediate.

Treatment of the chlorophosphine (2) with the Schiff base (1) in the presence of triethylamine in benzene at 0 °C gave (4)

<sup>†</sup> The first example of 2H-1,2,4,3 $\lambda$ 3-triazaphosphole was described by Schmidpeter *et. al.*3b

(b.p.  $30\,^{\circ}\text{C}$  at  $0.1\,\text{mmHg})^4$  and the new two-co-ordinate phosphorus compound (3) [ $^{31}\text{P}$  n.m.r.  $8\,254\,\text{p.p.m.}$ ;  $^{1}\text{H}$  n.m.r.  $8\,7.15\,$  (5H, one broad signal) and 2.2 (1H, no detectable  $J_{HNP}$ ); major ions in mass spectrum (70 eV) at  $m/z\,$  163, 146, 134, 119 and 104]. The  $^{1}\text{H}$  n.m.r. data strongly support the structure (3a), the first example of a 1H-1,2,4,3 $\lambda^3$ -triazaphosphole, for (3).  $^5$  In benzene solution (3a) was slowly converted into its isomer (3b).

Scheme 3

The reaction of (3) with trimethylchlorosilane resulted in the exclusive formation of (5), probably because of steric hindrance [(5),‡ b.p. 89 °C at 0.1 mmHg; <sup>31</sup>P n.m.r. δ 280 p.p.m.; <sup>1</sup>H n.m.r. δ 0.5 (9H, s). 7.2 (3H, m), and 8.3 (2H)].

The isomers (**3b**) and (**3c**), the latter being the first example of a 4H-1,2,4,3 $\lambda^3$ -triazaphosphole, can be prepared by addition of tris(dimethylamino)phosphine (**6**) to an excess of (**1**) in boiling toluene [(**3b**):  ${}^{31}P$  n.m.r.  $\delta$  251.5 p.p.m.,  $J_{PNH}$  27.6 Hz); (**3c**)  ${}^{31}P$  n.m.r. 249 p.p.m.,  $J_{PNH}$  24.9 Hz]. This reaction also provides a route to 1-benzyl-3,5-diphenyl-1,2,4-triazole (**7**)‡ [m.p. 104 °C;  ${}^{1}H$  n.m.r.  $\delta$  5.34 (2H) and 7—8.2 (5H);  ${}^{13}C$  n.m.r.  $\delta$  52.6 (CH<sub>2</sub>), 126—136.5 (C<sub>6</sub>H<sub>5</sub>), and 155.9 and 161.7 p.p.m. (C-3 and C-5)]. The identity of the novel triazole (**7**) was confirmed by its independent synthesis by heating the Schiff base (**1**) with (**4**) in refluxing toluene (yield 90%). A similar trapping reaction of (**4**) must also occur in the reaction of (**1**) with (**6**) (Scheme 2).

Additional proof of the structure of compounds ( $3\mathbf{a}-\mathbf{c}$ ) is provided by the following reactions. (i) Addition of pinacol to ( $3\mathbf{b}$ ) or ( $3\mathbf{c}$ ), a reaction which is characteristic of cyclic two-co-ordinated phosphorus compounds,<sup>2.6</sup> gave rise, as expected, to the formation of the spirophosphorane ( $\mathbf{8}$ )<sup>7</sup> (m.p.  $198\,^{\circ}$ C;  $^{31}$ P n.m.r.  $\delta$  -62 p.p.m.  $J_{PH}$  766 Hz); the same reaction with the isomer ( $3\mathbf{a}$ ) afforded first another spirophosphorane ( $\mathbf{9}$ ) ( $^{31}$ P n.m.r.  $\delta$  -55 p.p.m.,  $J_{PH}$  752 Hz) which was slowly converted into ( $\mathbf{8}$ ). (ii) Diethylamine reacts at the phosphorus–nitrogen double bond of ( $3\mathbf{a}-\mathbf{c}$ ) giving unstable PIII derivatives which are transformed into the corresponding

P<sup>IV</sup> product (10) on addition of sulphur in refluxing benzene [(10), m.p. 135 °C;  ${}^{31}$ P n.m.r.  $\delta$  68 p.p.m.,  ${}^{1}$ H n.m.r.  $\delta$  9.45 (d,  ${}^{2}J_{NHP}$  33 Hz, N<sup>2</sup>-H) and 8.8 (d,  ${}^{2}J_{NHP}$  24 Hz, N<sup>4</sup>-H)].

Scheme 5

 $^{2}J_{NHP}$  33 Hz, N<sup>2</sup>-H) and 8.8 (d,  $^{2}J_{NHP}$  24 Hz, N<sup>4</sup>-H)].  $^{13}$ C N.m.r. data for the species (3) are consistent with structures (3a—c) (a weak doublet at  $\delta$  166 p.p.m.,  $^{2}J_{CP}$  14 Hz for the carbon atom of the triazaphosphole ring, analogous to the case of *N*-substituted triazaphospholes<sup>8</sup>).

It is of interest that a transient intermediate [ $\delta$  ( $^{31}P$ ) 27 p.p.m.] is formed during the reactions in Schemes 1 and 2. This species can be isolated when ( $\mathbf{6}$ ) is treated with the Schiff base ( $\mathbf{11}$ ) (Scheme 5). In a typical experiment the phosphine ( $\mathbf{6}$ ) (0.05 mol) in acetonitrile (20 ml) was slowly added, at room temperature, to ( $\mathbf{11}$ ) (0.05 mol) also in acetonitrile (20 ml). Removal of acetonitrile followed by extraction with chloroform and recrystallisation from hexane gave compound ( $\mathbf{12}$ )‡ (50% yield), which is stable in the solid state [m.p. 136 °C (decomp.);  $^{31}P$  n.m.r.  $\delta$  27 p.p.m.;  $^{1}H$  n.m.r.  $\delta$  2.16 (3H, d,  $J_{PH}$  10 Hz), 2.51 (3H, d,  $J_{PH}$  10 Hz), 4.0 (1H), 4.4 (1H, d,  $J_{PH}$  6 Hz), and 7.5 (5H, m);  $^{13}C$  n.m.r.  $\delta$  122.6, 119.6 (sp²);  $^{11}6$ , 106 (CN); 50 (P-C, d,  $J_{PC}$  90 Hz): 126—136 p.p.m. ( $C_6H_5$ );  $v_{NH}$  3200,  $v_{P=N}$  1275 cm<sup>-1</sup>; major ions in mass spectrum (70 eV) at m/z 314, 312, and 196].

In acetonitrile solution (12) is slowly converted into (13) at room temperature (3 days) by an unexpected ring contraction reaction involving an unusual cyclic phosphorus—carbon bond cleavage under very mild conditions. In this reaction (4) is trapped in its protonated form, the counter-anion being the

<sup>‡</sup> Satisfactory elemental analyses were obtained.

cyclic phosphorus derivative. Addition of methyl iodide to (13) gives (14). 3c,6

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